

Lecture Notes

Introduction to Strongly Correlated Electron Systems

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Introduction to strongly correlated electron systems

I. Introduction

Brief summary of electrons in solids, origin of strong electron correlations

II. Classes of strongly correlated electron systems

(a) Transition metal compounds: 3d-electrons

- Hubbard model, Mott insulator, metal-insulator transition
- Spin, charge, and orbital degrees of freedom and ordering phenomena, selected materials

(b) Heavy fermion systems: 4f (5f) – electrons

- The Kondo effect, heavy fermion systems, non-Fermi liquid behavior,
- Quantum phase transitions, unconventional superconductivity, selected materials

(c) Nanoscale structures:

- Quantum confinement, unusual properties for potential applications

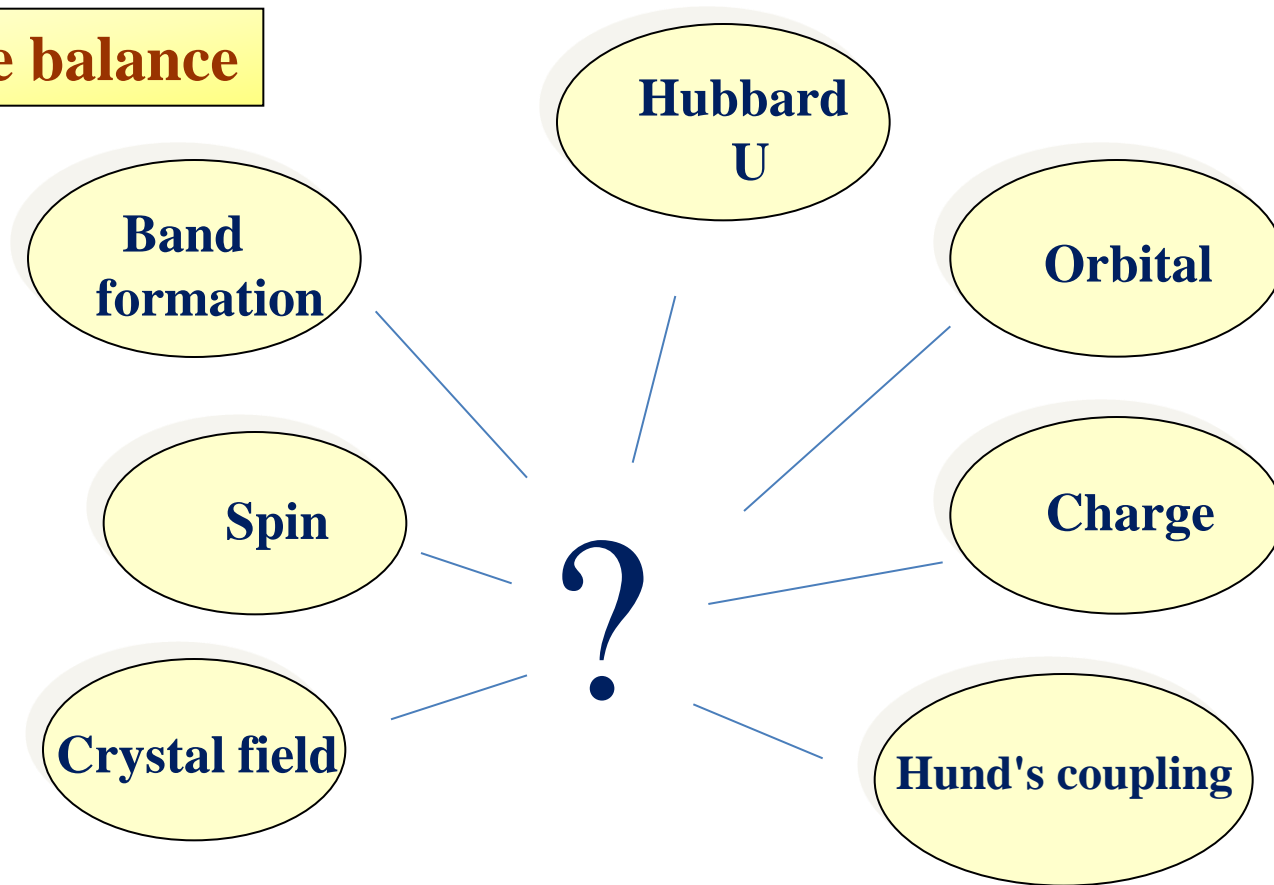
III. Pressure effect on the ground state properties:

- Recent experimental results on heavy fermions and transition metal compounds

IV. Summary and open discussion

Transition metal systems and electron correlations

Delicate balance



systems can be quite close to the borderline $U \sim W(t)$

Thus many interesting transitions to unusual ground states can occur by changing T, P, filling, structure, etc.

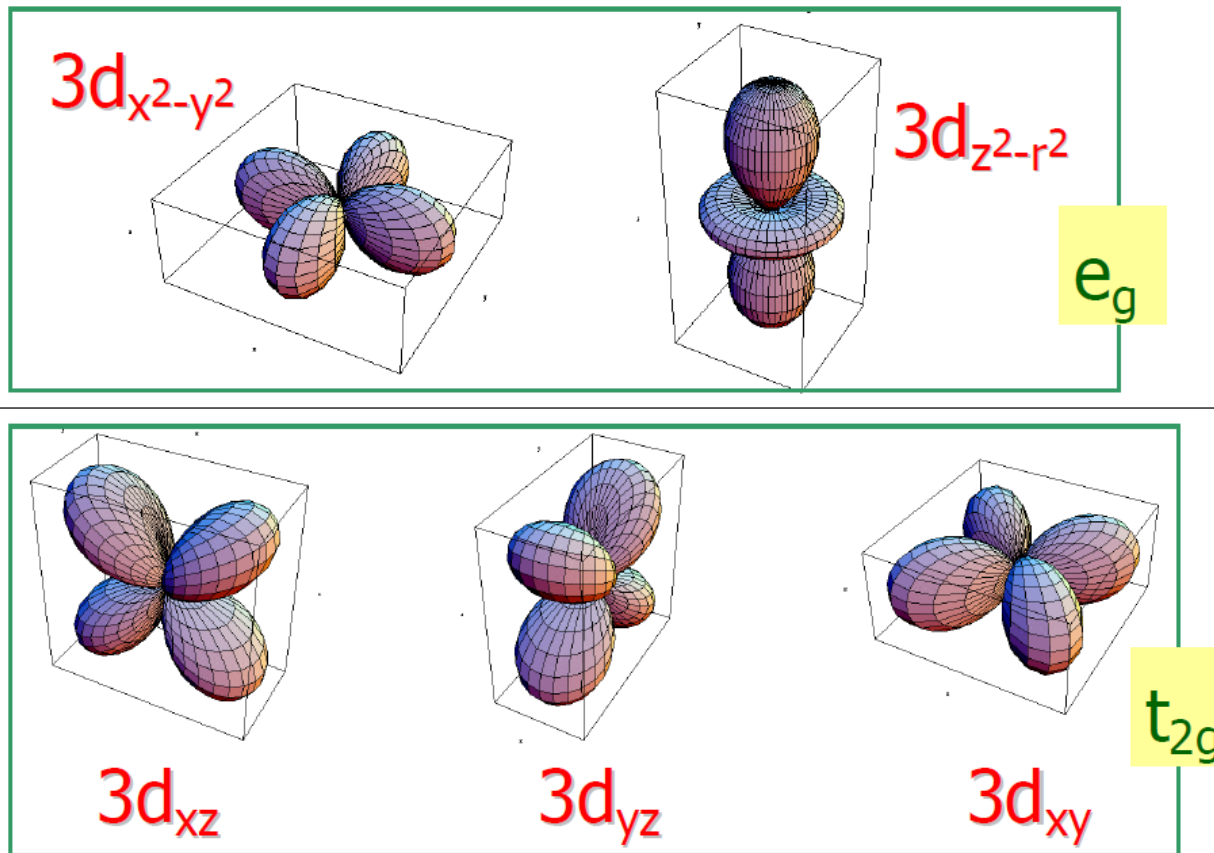
more discussion with examples later!

Orbital degree of freedom

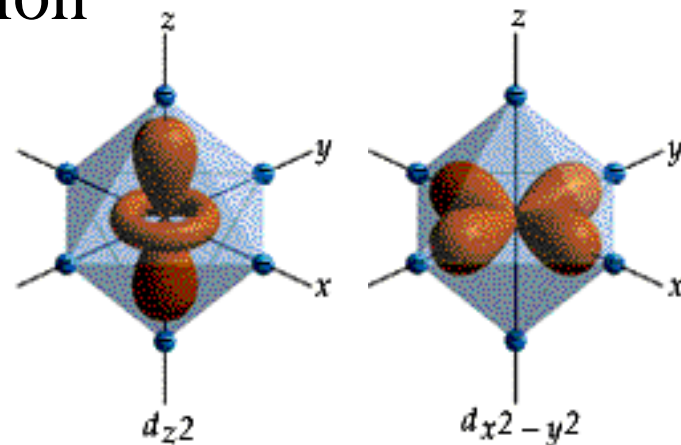
3d orbitals in central potential of a free atom/ ion

5-fold degenerate 3d orbitals

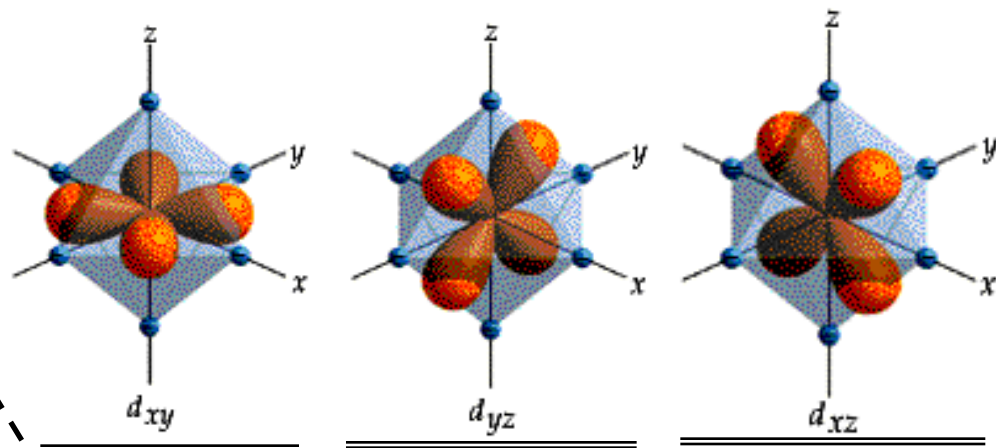
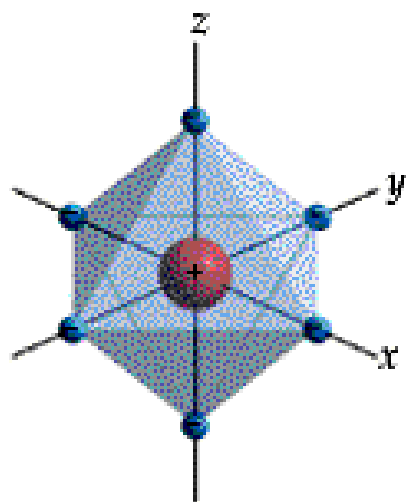
3d: $n=3$, $\ell=2$, $m_\ell = -2, -1, 0, 1, 2$ \longrightarrow 5 orbitals



d-Orbitals and Ligand Interaction (Octahedral Field)



**d-orbitals pointing directly at axis
are affected most by electrostatic
interaction**

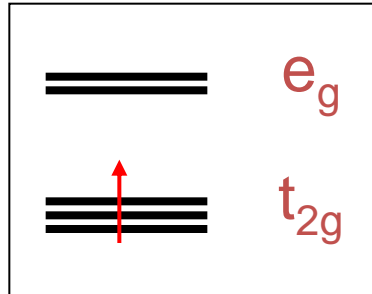


**d-orbitals not pointing directly at axis
(stabilized) by electrostatic interaction**

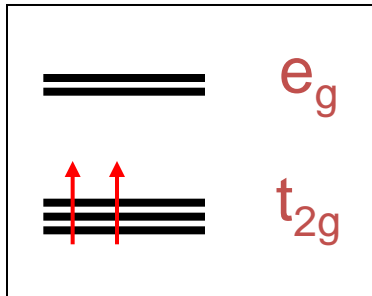
Orbitals degeneracy

„pseudo-Cubic“ Perovskites

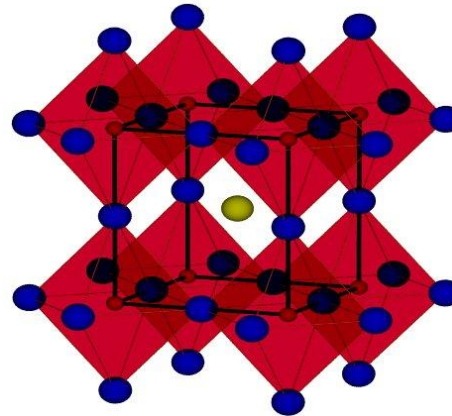
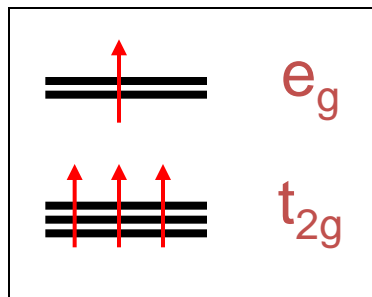
YTiO₃
3d¹



YVO₃
3d²



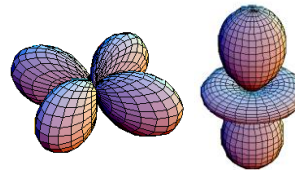
LaMnO₃
3d⁴



Orbitals degeneracy is lifted by:

1. Lattice distortion (Jahn-Teller)
2. orbital and spin dependent superexchange

more details later!



orbitals degeneracy

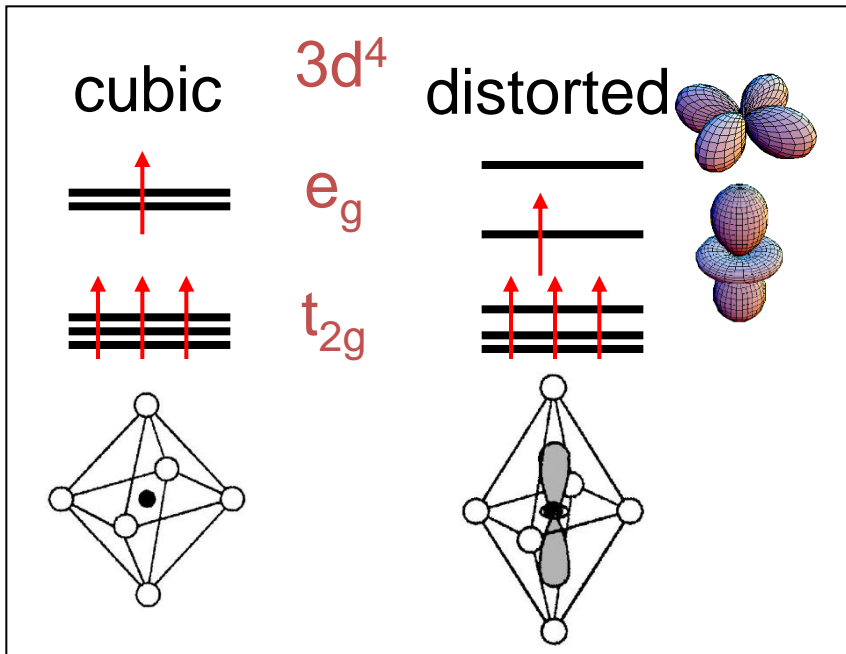
➔ orbital degree of freedom

Differences between Orbitals and Spins

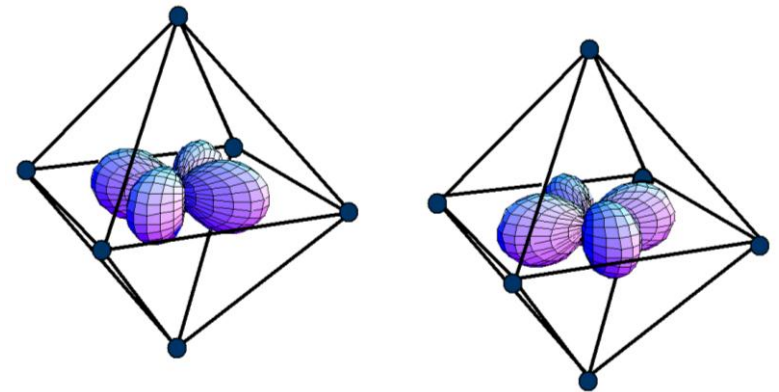
The orbitals are usually strong coupled to the lattice

Distortion

Lifts the orbital degeneracy



Anisotropy

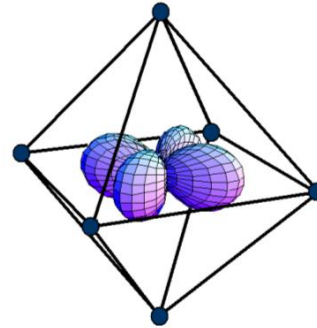
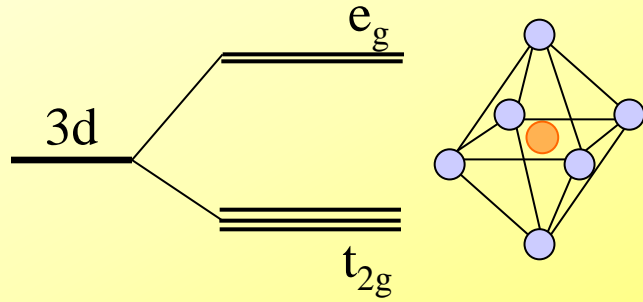


rotated orbitals are not equivalent
Spins in 1. approximation invariant;
e.g. In Ferromagnets alle spins are aligned in the same direction, but this direction is arbitrary

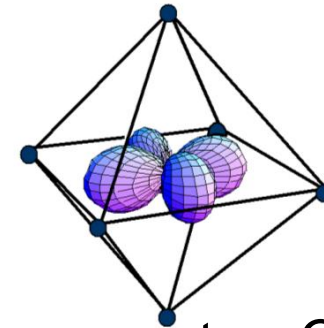
If, however, exchange interaction is dominating, orbitals will be strongly coupled to the spins

Lower Symmetry

cubic



e_g - Orbitals
(x^2-y^2 and $3z^2-r^2$)

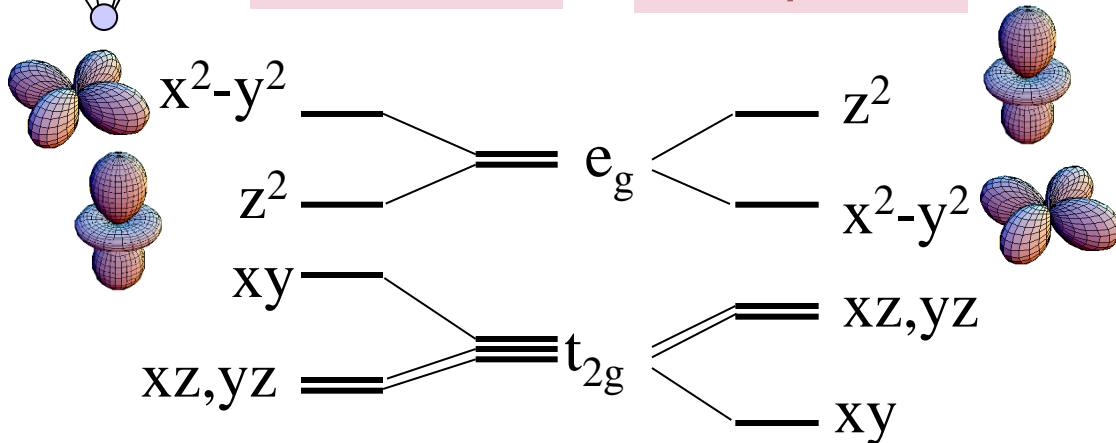
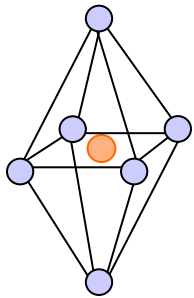


t_{2g} - Orbitals
(xy, xz und yz)

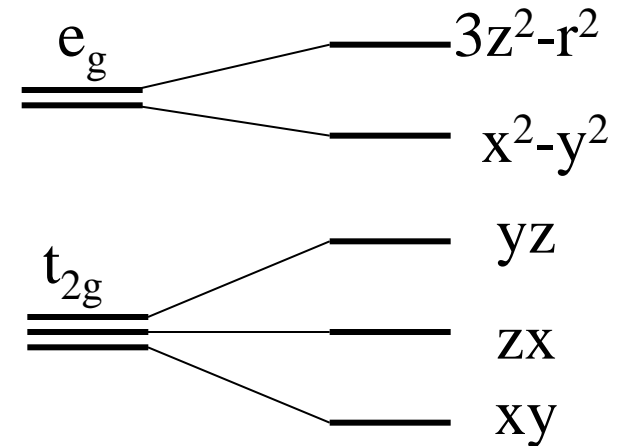
tetragonal

stretch

compress



orthorhombic



Orbital Order

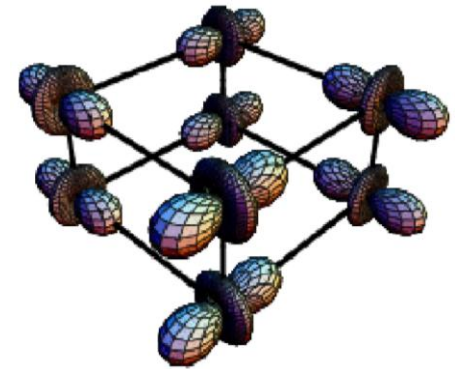
Orbital degeneracy is lifted by:

- 1. Lattice distortion (Jahn-Teller) and/ or**
- 2. orbital and spin dependent superexchange**

Both mechanisms lead to orbital order

orbital order:

**periodic arrangement of specific orbitals
which leads to a long-range ordered pattern**

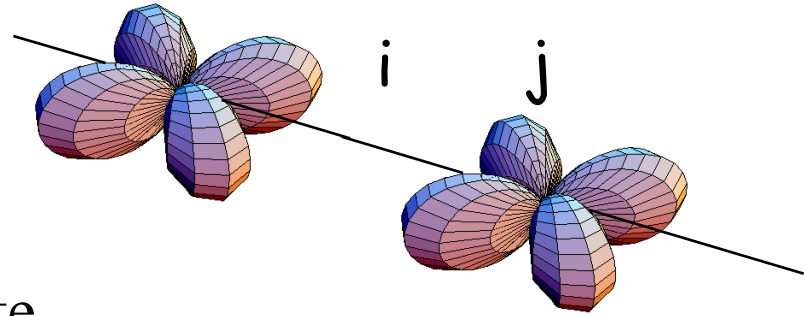


LaMnO_3

Types of orbital ordering

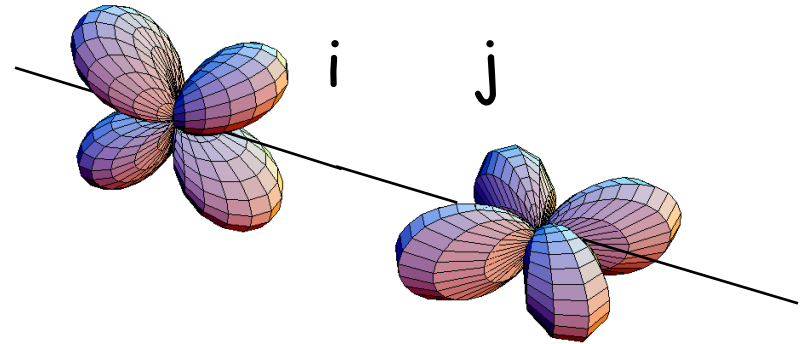
Ferro-orbital ordering:

the same orbital is occupied at each site



Antiferro-orbital ordering:

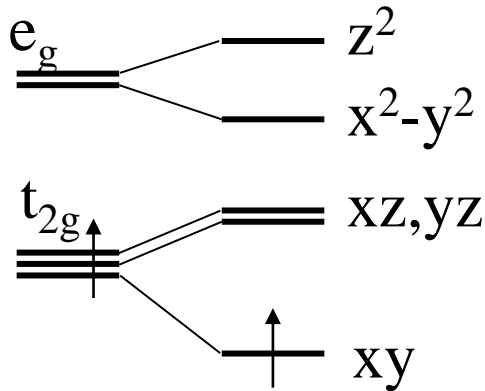
different orbitals are occupied at neighboring sites



Jahn-Teller-Effect

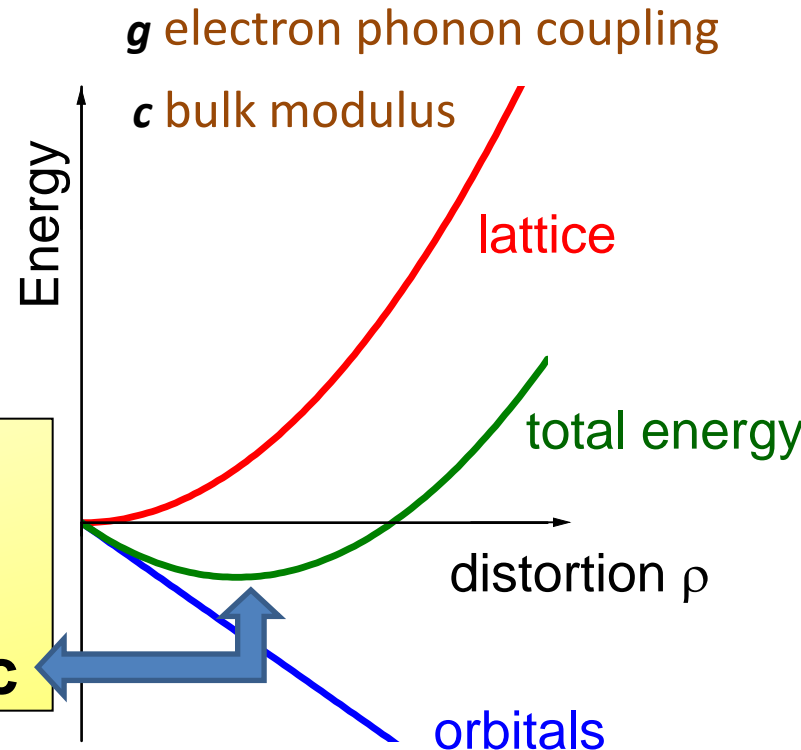
Orbitals degeneracy is lifted by spontaneous distortion

e.g.: 3d¹ in cubic crystal field



Jahn-Teller active ions (orbital degeneracy):
 1, 2, 4, or 5 electrons in t_{2g}-Orbitals; or
 1 or 3 electrons in e_g-Orbitals

$$E(\rho) = \frac{c}{2}\rho^2 - g\rho$$



Distortion of the octahedron leads to:

- Linear reduction of the electron energy
- Quadratic increase of the lattice energy

→ E(ρ) at minimum when distortion $\rho = g/c$

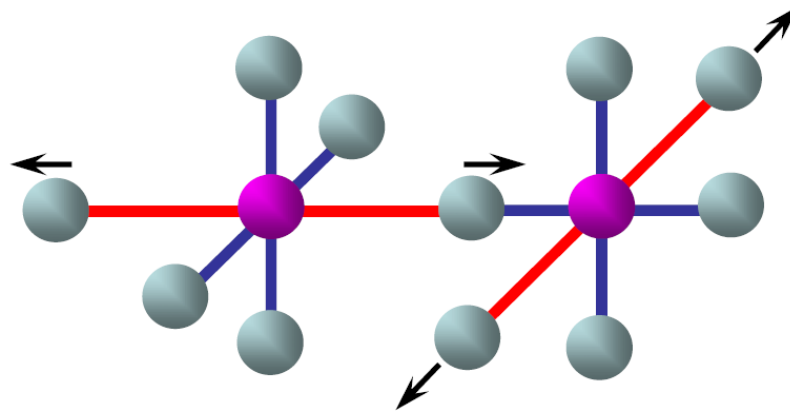
By the distortion, the orbital degree of freedom is „quenched“; but spin degree of freedom remains preserved .

Orbital Order- Cooperative JT effect

Coupling between local distortions on different sites

Jahn-Teller effect mediates *interaction* between orbitals of next neighbors sites, thereby leading to *orbital order*

Cooperative Jahn-Teller effect



stretching of next neighbors Octahedra in direction perpendicular to each other.

2. Orbital and spin dependent superexchange

exchange mechanism which couples Orbital and magnetic order

 **Kugel-Khomskii model**

description and comments, see board!

For more details, see:

K. I. Kugel and D. I. Khomskii, Sov. Phys. JETP 37, 725 (1973);

K. I. Kugel and D. I. Khomskii, Sov. Phys. Ups. 25, 231 (1982)

Orbital order : Kugel-Homskii model

* Consider orbital ordering for the case of double degeneracy ($u \gg t$) and with 1 electron per site, $n=1 \Rightarrow$ only eg electrons, no Jahn-Teller distortion

* localized electrons are characterized by

1. Spin σ ($\sigma^z = \pm \frac{1}{2}$) \Rightarrow double degeneracy = 2

2. Index of occupied orbital ($\alpha = 1$ or 2)
 \Rightarrow extra double degeneracy:

effective pseudo spin operator $\tilde{\tau}^z = \pm \frac{1}{2}$

$$\tilde{\tau}^z = +\frac{1}{2} \text{ for orbital } 1$$

$$\tilde{\tau}^z = -\frac{1}{2} \text{ " " " } 2$$

\Rightarrow orbital degeneracy = $2\tilde{\tau}^z + 1 = \underline{\underline{2}}$

Description of the spin-orbital interaction by the degenerate Hubbard model, which includes the intra-atomic exchange responsible for Hund's rule coupling

$$H = \sum_{\langle i,j \rangle} \sum_{\alpha\beta} t_{ij}^{\alpha\beta} c_{i\alpha}^\dagger c_{j\beta} + U \sum_{i,\alpha\beta} n_{i\alpha} n_{i\beta} - J_H \sum_i \sum_{\alpha \neq \beta} (\frac{1}{2} + 2\vec{S}_i \cdot \vec{S}_{i\beta})$$

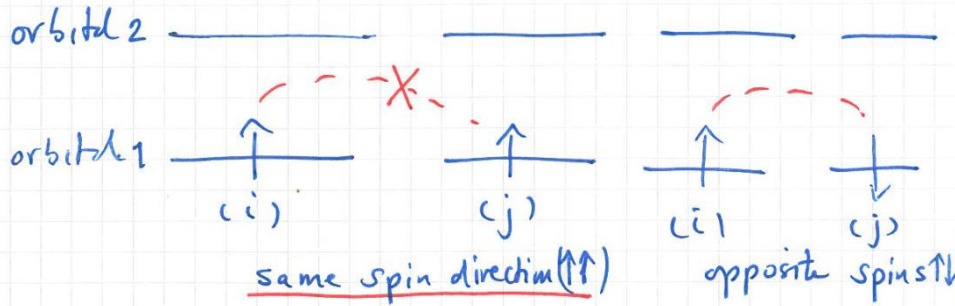
sites i,j spin orbital 1,2 hopping, same spin direction different spins same state (i) ; different orbitals!

$$-J_H \sum_{\alpha \neq \beta} (\frac{1}{2} + 2\vec{S}_{i\alpha} \cdot \vec{S}_{i\beta}) \left\{ \begin{array}{l} \text{for } \uparrow\uparrow \vec{S}_{i1} \cdot \vec{S}_{i2} \Rightarrow \frac{1}{4} \Rightarrow -J_H \\ \text{for } \uparrow\downarrow \vec{S}_{i1} \cdot \vec{S}_{i2} \Rightarrow -\frac{1}{4} \Rightarrow 0 \end{array} \right\} \Rightarrow \text{Difference} = -J_H$$

How do orbital correlations drive magnetism (Kugel-Khomskii model)

assume: only hopping between the same orbitals $\Rightarrow t'' = t'''' = t, t'' = 0$

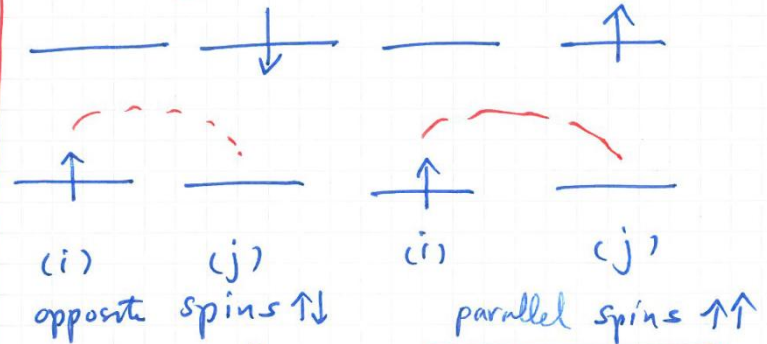
same orbitals



$\Rightarrow \Delta E = 0$
 $\Rightarrow E_{FM} = 0$

$\Rightarrow \Delta E = \frac{-2t^2}{U}$
 $\Rightarrow E_{AF} = \frac{-2t^2}{U}$

different orbitals



$\Rightarrow \Delta E = \frac{-2t^2}{U}$

$\Rightarrow \Delta E = \frac{-2t^2}{U - J_H}$

Lowest Energy \leftarrow

\Rightarrow Results:

ferroorbital ordering \Rightarrow antiferromagnetic order

Results:

antiferro-orbital ordering \Rightarrow ferromagnetic ordering

\Rightarrow

Virtual hoppings lead ~~to both~~ simultaneously to both orbital and spin ordering

effective exchange Hamiltonian (Kugel-Khomskii) : 2 eg electrons
2 orbitals, 2 spin directions $\uparrow \downarrow$

$$H = \sum \left\{ \underset{\text{spins}}{J_s \vec{s}_i \cdot \vec{s}_j} + \underset{\text{orbitals}}{J_\tau \tau_i \cdot \tau_j} + J_{S\tau} (\vec{s}_i \cdot \vec{s}_j) (\tau_i \cdot \tau_j) \right\}$$

Coupling between spin and orbital degrees of freedom.

\Rightarrow total 4 degeneracy :

$$\vec{S} = \frac{1}{2} \begin{cases} S_z = +\frac{1}{2} \uparrow \\ S_z = -\frac{1}{2} \downarrow \end{cases} ; \vec{\tau} = \frac{1}{2} \begin{cases} +\frac{1}{2} \text{ for orbital 1} \\ -\frac{1}{2} \text{ for orbital 2} \end{cases}$$

\Rightarrow 4 states for 2 eg electrons

$$\begin{array}{l} \text{orbit} \quad \left. \begin{array}{l} |1, \uparrow\rangle \\ |2, \uparrow\rangle \end{array} \right\} \Rightarrow \left. \begin{array}{l} |\tau^z = +\frac{1}{2}, S^z = +\frac{1}{2}\rangle \\ |\tau^z = -\frac{1}{2}, S^z = +\frac{1}{2}\rangle \end{array} \right\} \end{array}$$

$$|1, \downarrow\rangle \Rightarrow$$

$$|2, \uparrow\rangle \Rightarrow$$

Directional nature of orbitals and exchange interactions

⇒ orbital occupancy determine the sign and magnitude of the exchange interactions

⇒ magnetic ground state !

Example 5 :

Examples:

orbital occupancy determine the sign and magnitude of exchange interaction, see board!

$KCuF_3$: JT distortion \Rightarrow elongation along the ab plane

Cubic perovskite

Cu^{2+} ($3d^9$) : e_g system

y^2-z^2/x^2-z^2

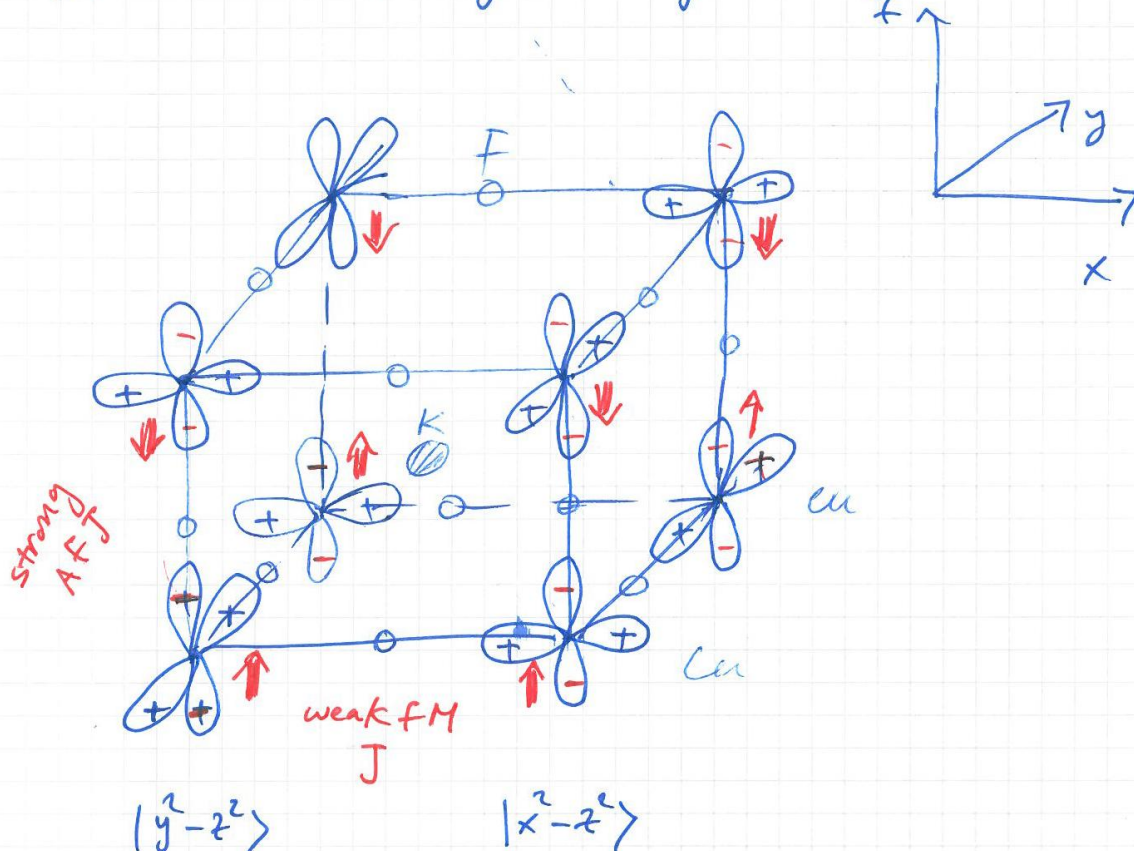


1 d hole (eg)

$3x^2-r^2/3y^2-r^2$



t_{2g}



strong AFM

weak FM J

$|y^2-z^2\rangle$

$|x^2-z^2\rangle$

$J_c/J_a \approx 100 !$

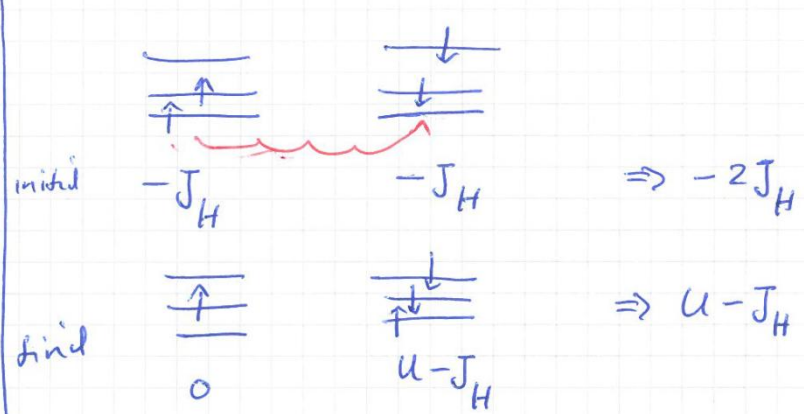
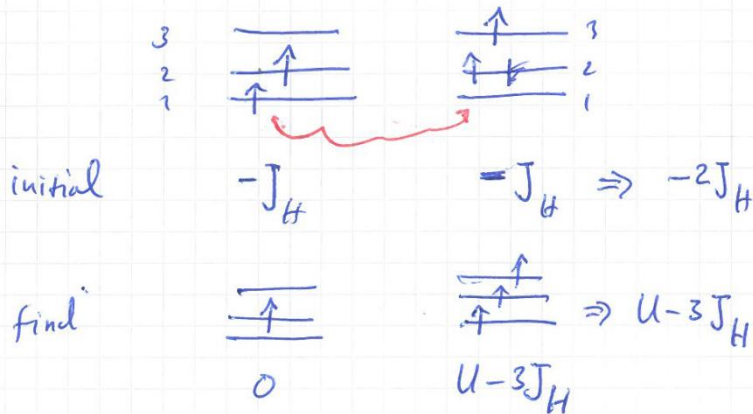
\Rightarrow 1 dimensional AF chain || to z !, despite cubic structure!

Ferromagnetic insulator

LaVO₃ ; V³⁺ (3d²) : t_{2g} system (assuming hopping to same orbital)

parallel spins at i and j (FM pair)

antiparallel spins (AF pair)



$$\Delta E_{\uparrow\uparrow} = (U - 3J_H) - (-2J_H)$$

$$\Delta E_{\uparrow\uparrow} = U - J_H$$

$$\Delta E_{\uparrow\downarrow} = U - J_H - (-2J_H)$$

$$\Delta E_{\uparrow\downarrow} = U + J_H$$

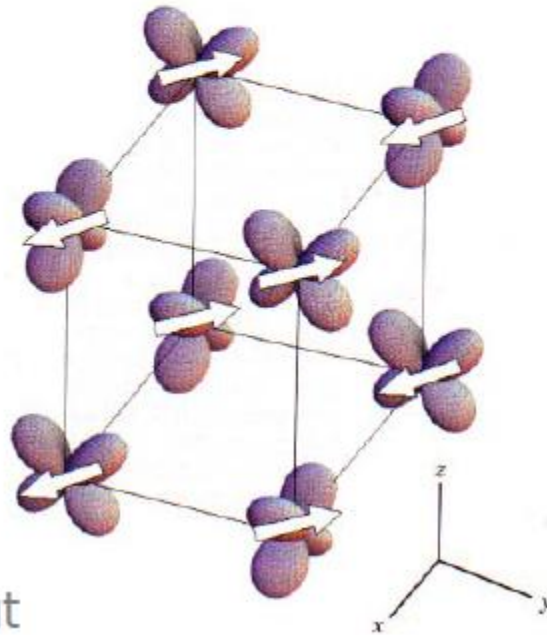
~~Ground FM state is favorable~~ \Rightarrow FM along the z direction
FM chains despite cubic structure!

order patterns: LaVO₃

- La = 3+, O₃ = 6-, thus V = 3+

21	22	23	24	25	26	27	28	29	30
Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn

- V³⁺ means 3d²: a system with partial t_{2g} occupation



d_{xy} left out

orbital-spin pattern:

spin z: FM chains

spin xy: AFM

orbital: d_{yz} and d_{xz}
alternate in x, y and z

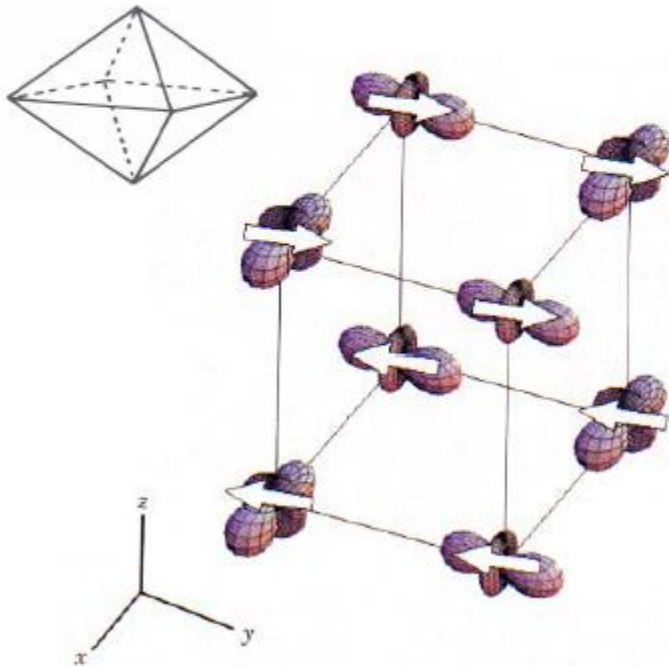
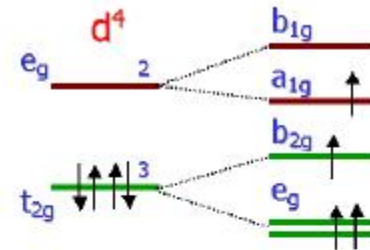
anisotropic despite
cubic structure !

order patterns: LaMnO₃

● La = 3+, O₃ = 6-, thus Mn = 3+

21	22	23	24	25	26	27	28	29	30
Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn

● Mn³⁺ means 3d⁴: a Jahn-Teller system.



orbital degeneracy lifted by JT:

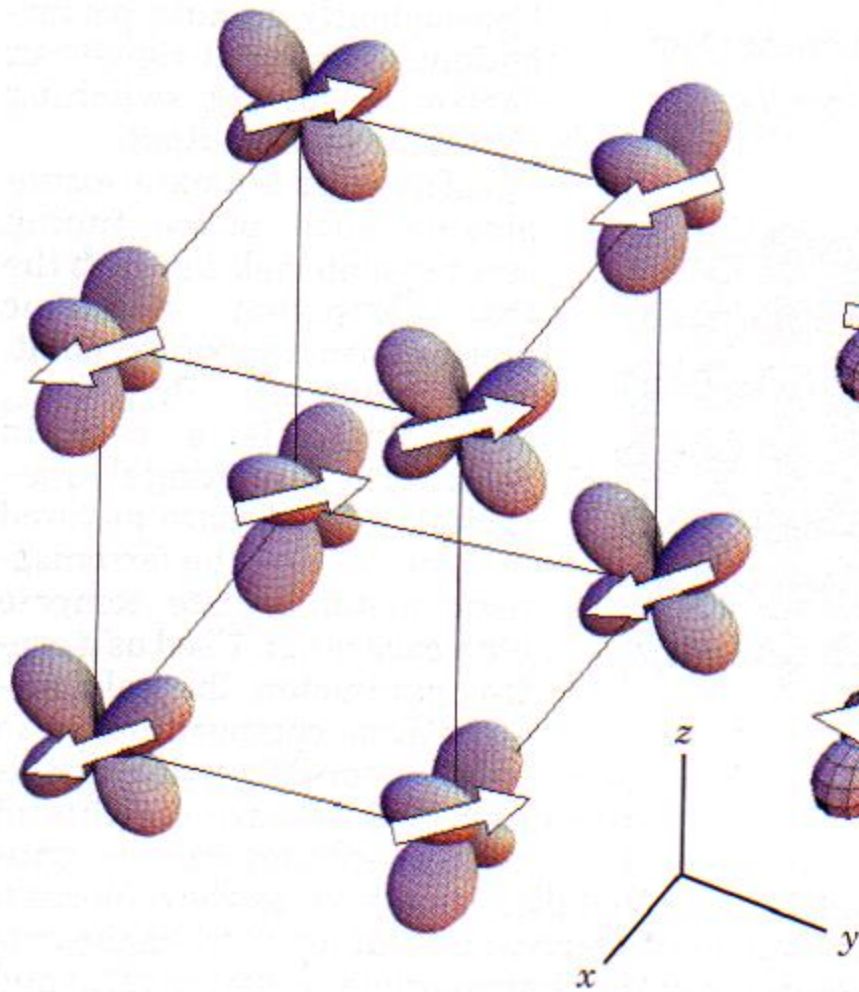
favours occupation of either
 $3d_{z^2-r^2}$ or $3d_{x^2-y^2}$

linear combination
 alternating:

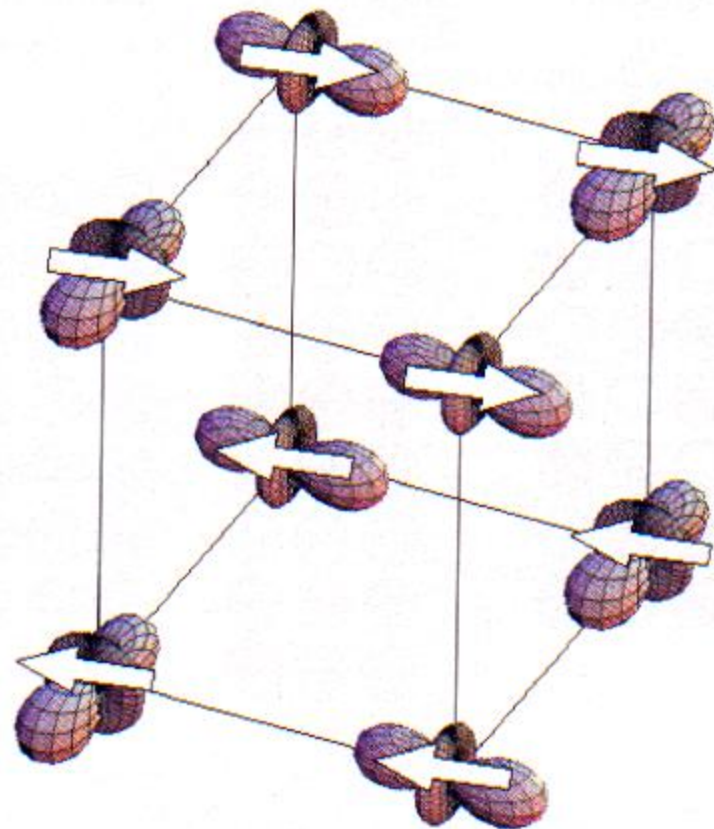
$3d_{x^2-r^2}$ and $3d_{y^2-r^2}$

e_g – Orbitals
 (x^2-y^2 and $3z^2-r^2$)

a LaVO_3

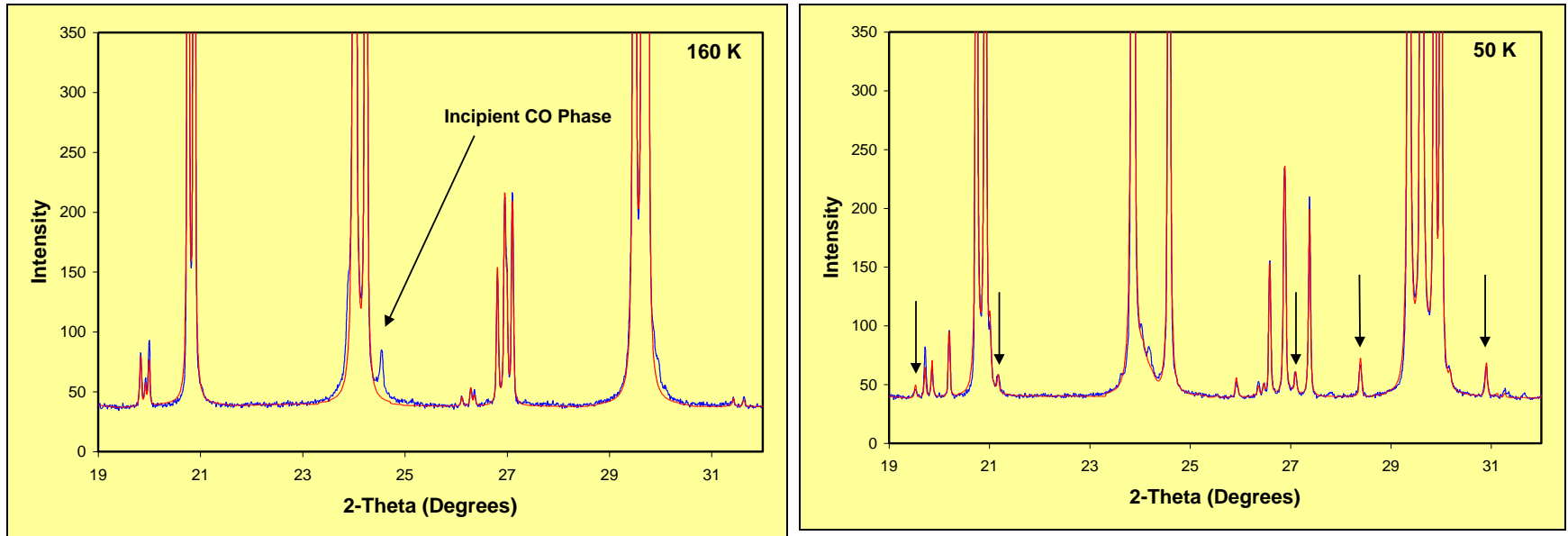


b LaMnO_3



Experimental methods for investigation of Orbital Ordering

Charge Ordering in $\text{Nd}_{0.5}\text{Sr}_{0.5}\text{MnO}_3$

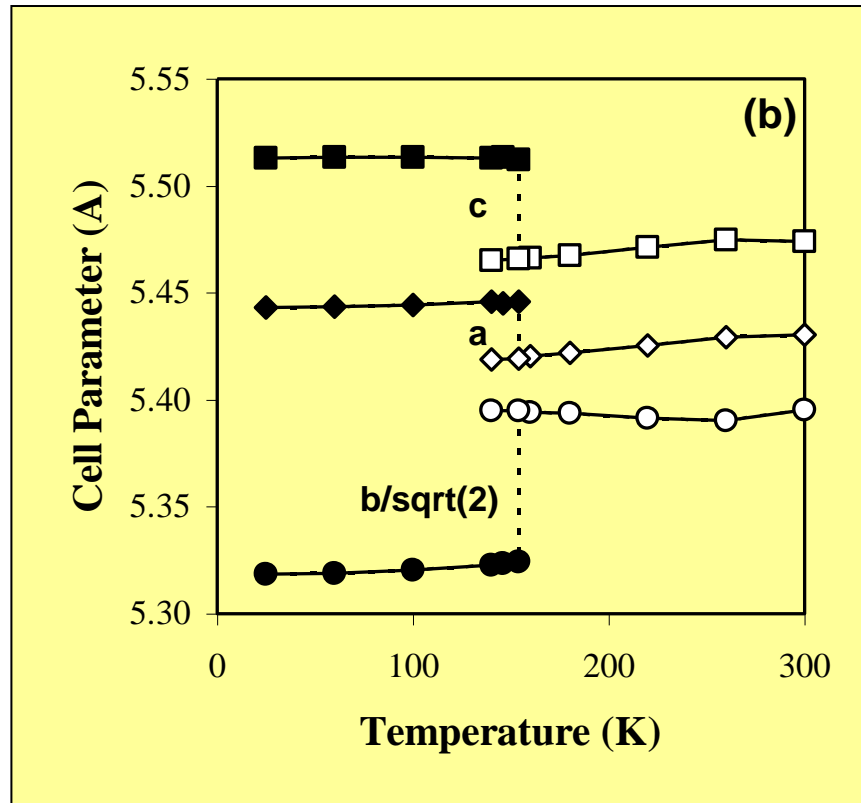


Synchrotron X-ray Powder Diffraction Data (NSLS-X7A)

A series of weak superstructure reflections arise (1% intensity at the strongest) that indicate doubling of the a -axis.

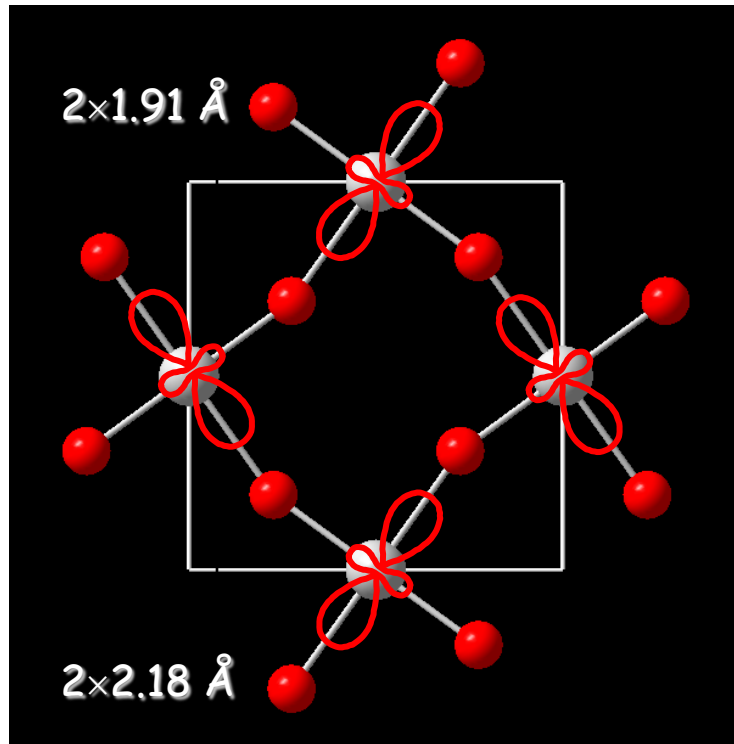
Woodward, Cox, Vogt, Rao, Cheetham, *Chem. Mater.* **11**, 3528-38 (1999).

Orbital Ordering in $\text{Nd}_{0.5}\text{Sr}_{0.5}\text{MnO}_3$



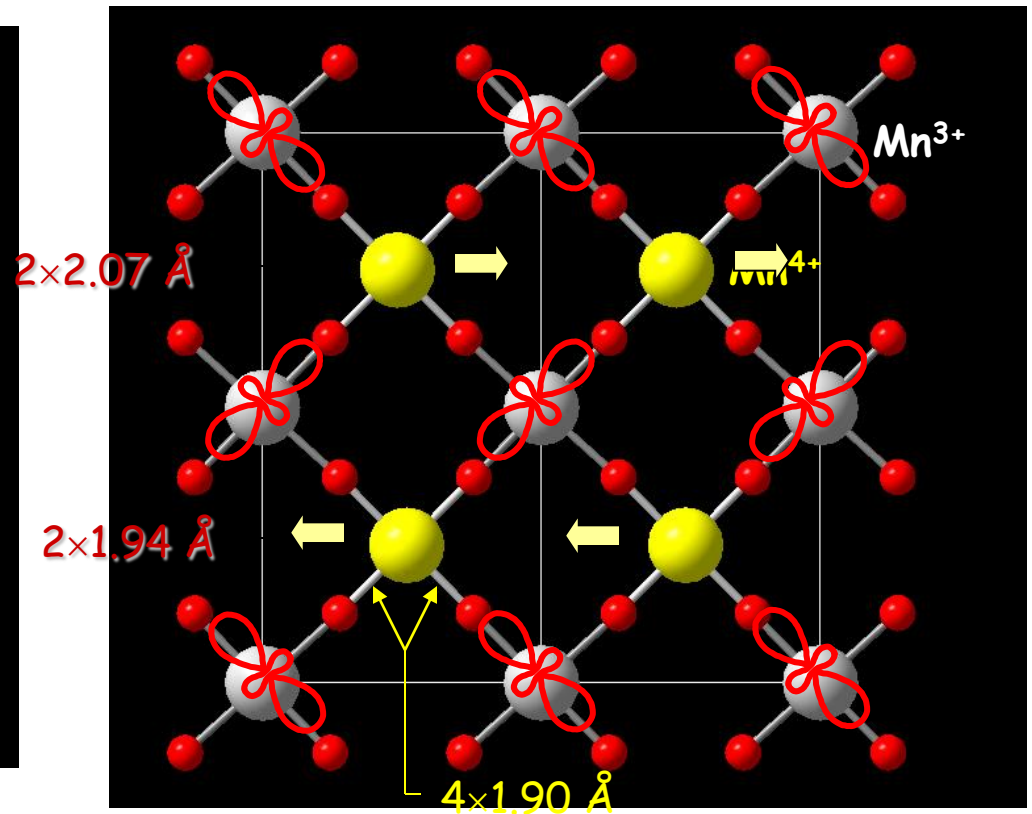
Upon cooling below 150 K, the a & c -axes expand and the b -axis contracts. This is the signature of orbital ordering

Examples of Orbital Ordering



LaMnO₃ (298 K)

Rodriguez-Carvajal, et al.
Phys. Rev. B **57**, R3189
(1998).



NdSrMn₂O₆ (50 K)

Woodward, et al. Chem.
Mater. **11**, 3528-38
(1999).

Experimental methods for investigation of Orbital Ordering

Resonant Inelastic X-ray Scattering (**RIXS**)

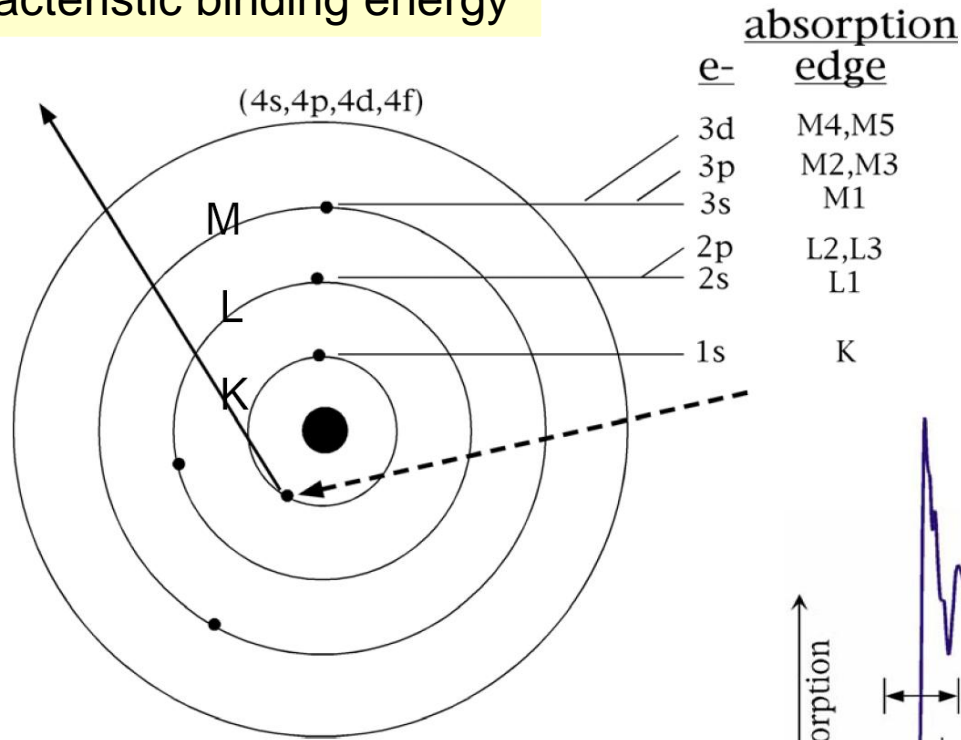
For an excellent review see:

Amen et. al; Rev. Mod. Phys. 83 705 (2011)

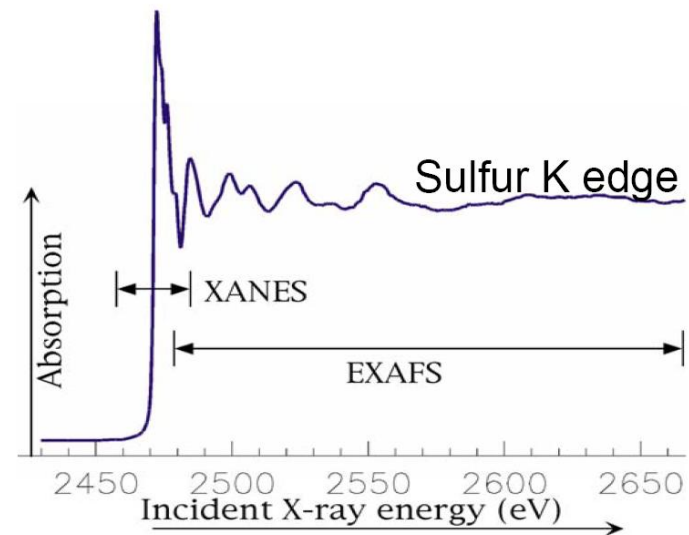
First in short: X-ray absorption spectroscopy

X-ray absorption spectroscopy

Each edge of each element has a characteristic binding energy

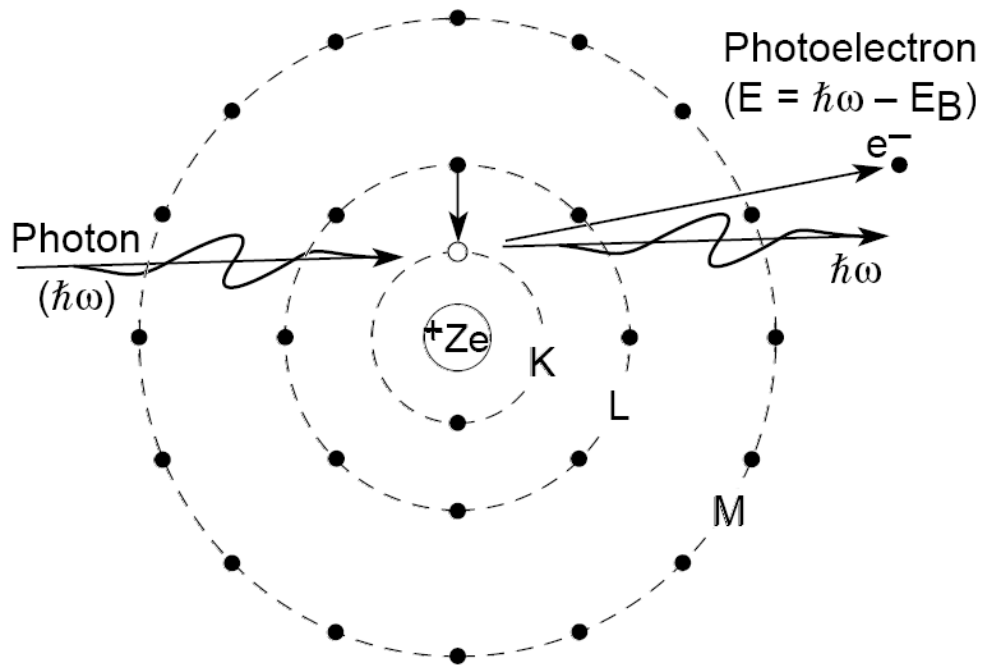


Absorption occurs when the energy of the incident photon is sufficient to eject the electron.

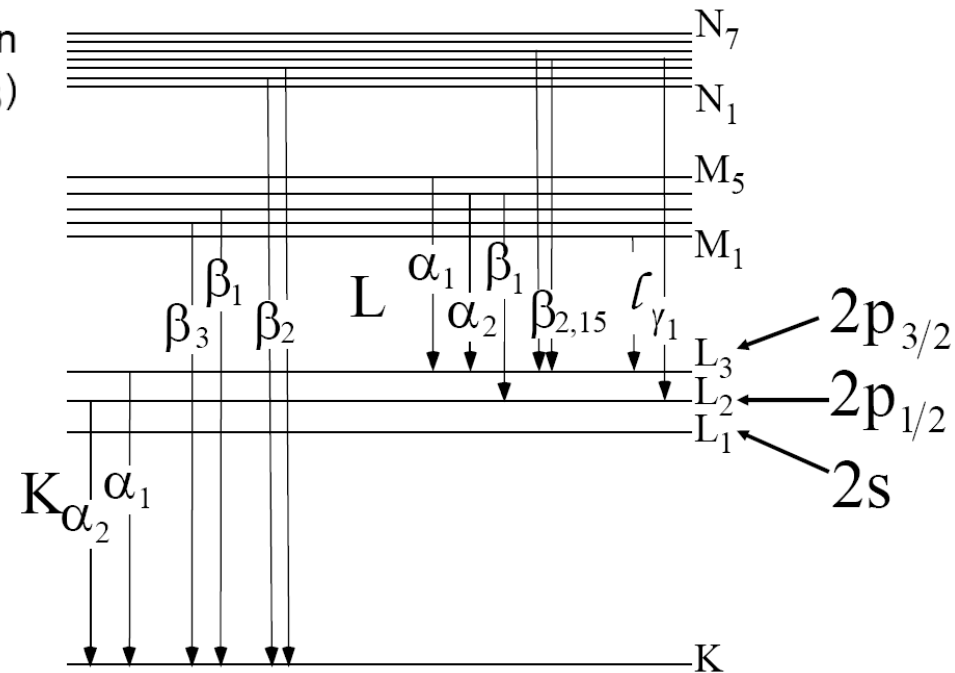


X-ray fluorescence spectroscopy (XFS)

Principle:

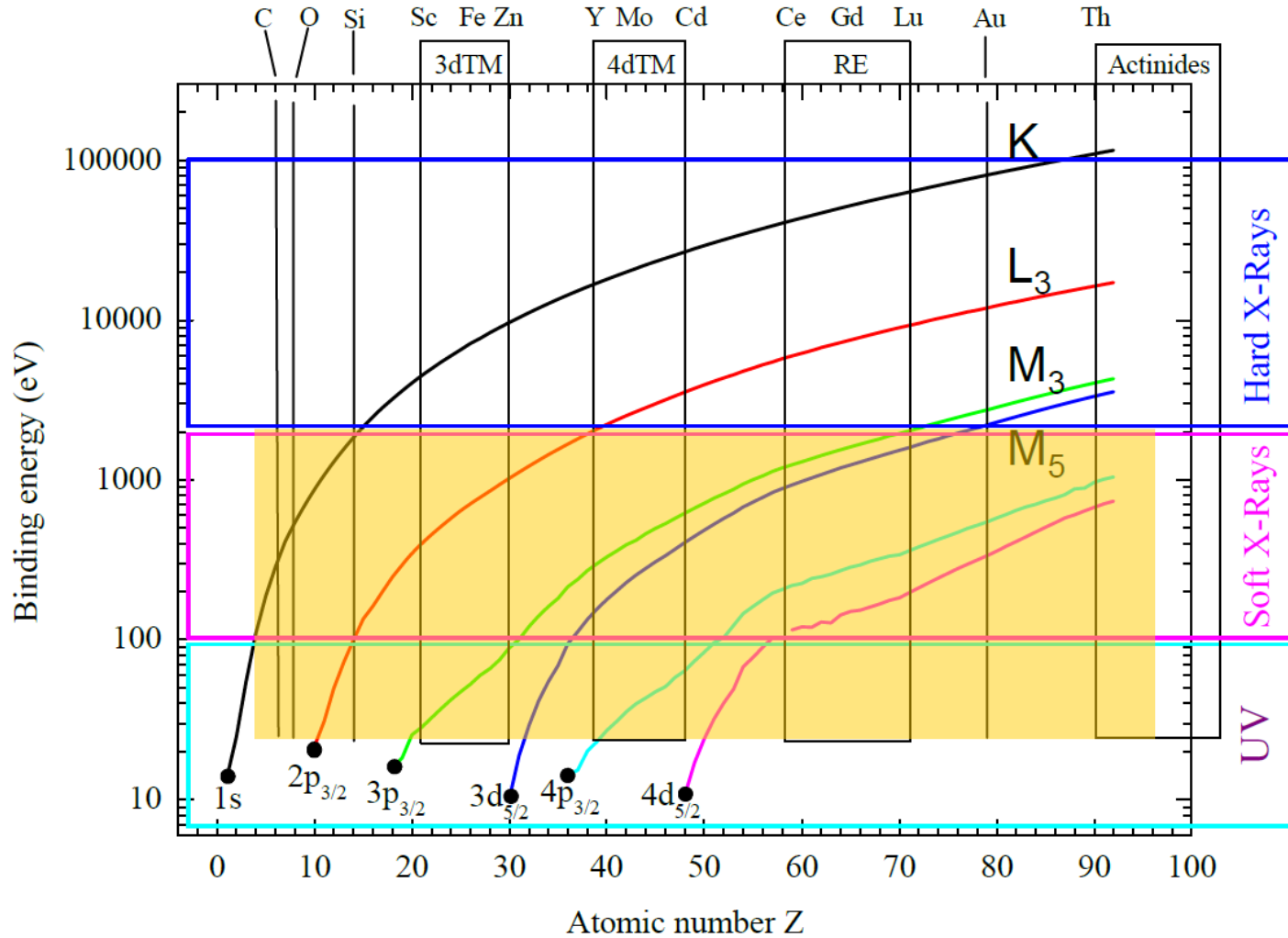


X-ray fluorescence lines:



an element specific technique!

core level energies



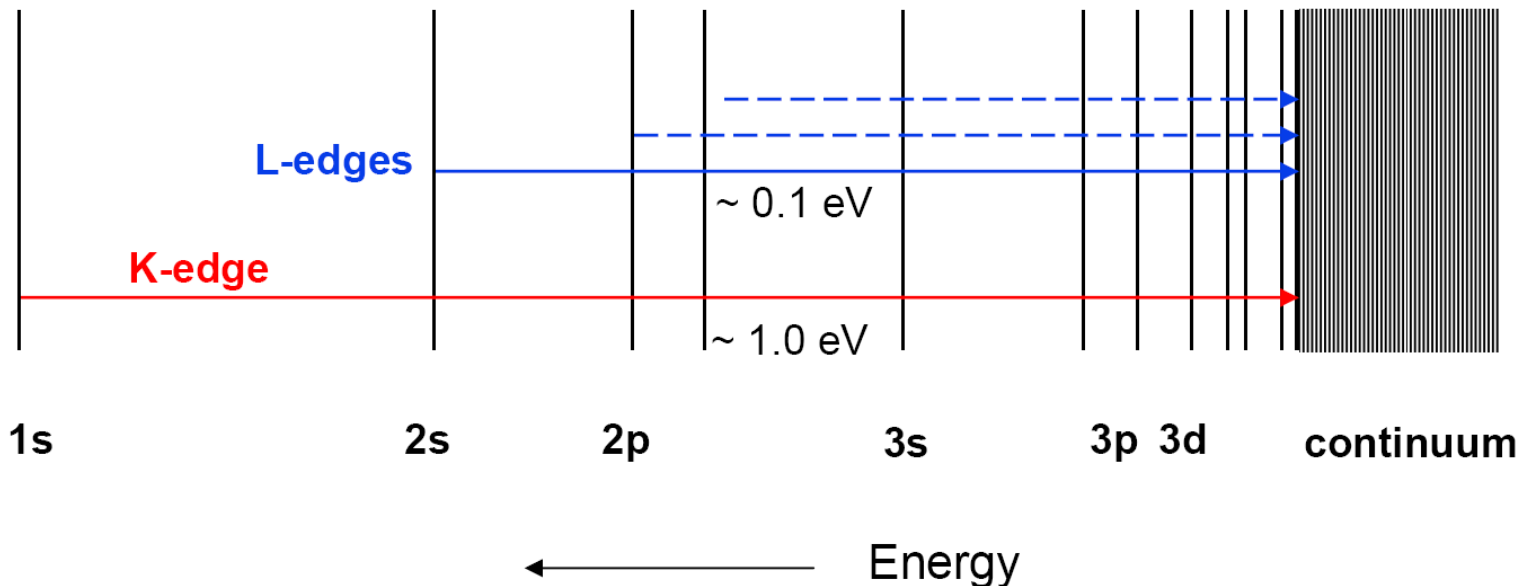
Examples for different edges

Examples:

Cu K-edge ~9000 eV
Cu L-edges ~930 eV
Cu M-edges ~70-120 eV

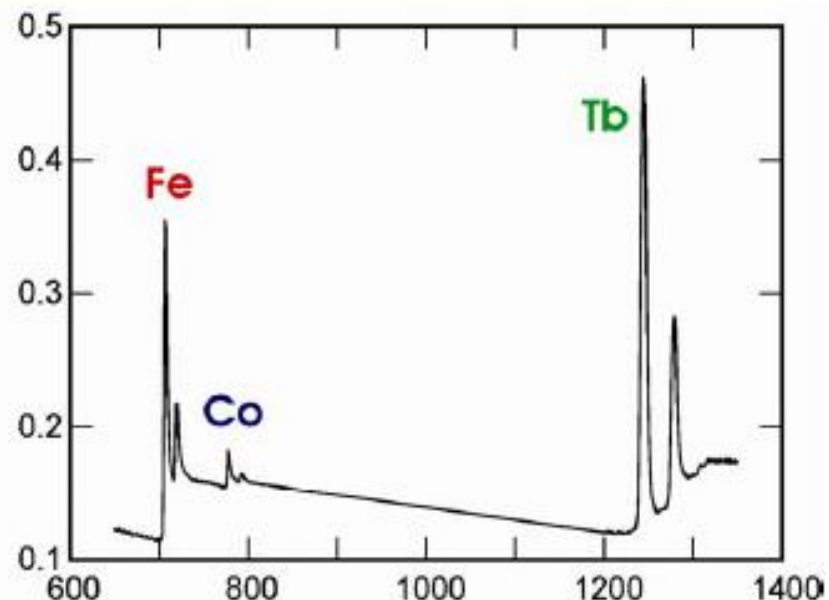
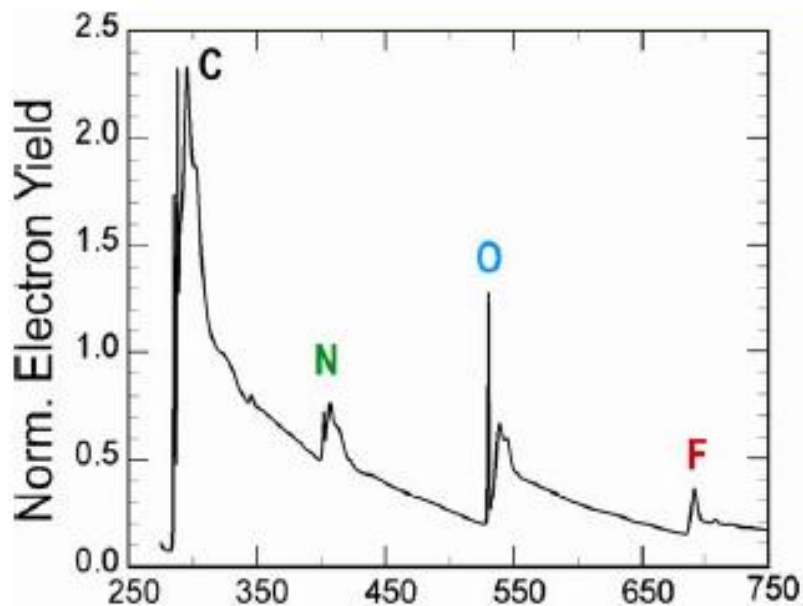
Fe K-edge ~7000 eV
Fe L-edges ~720 eV
Fe M-edges ~50-100 eV

S K-edge ~2472 eV
S L-edges ~200 eV

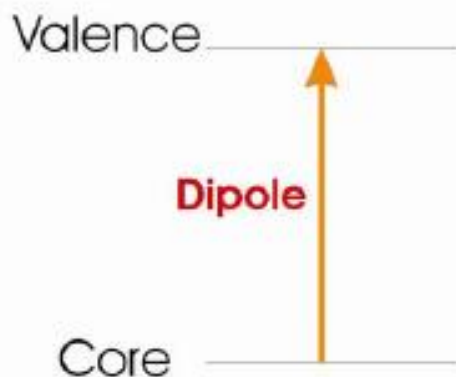


- An edge results when a core electron absorbs energy equal to or greater than its binding energy
- Results in transitions to unoccupied atomic or molecular orbitals
- Edges are labeled according to the shell the **core electron** originates from

Tunable x-rays offer elemental specificity



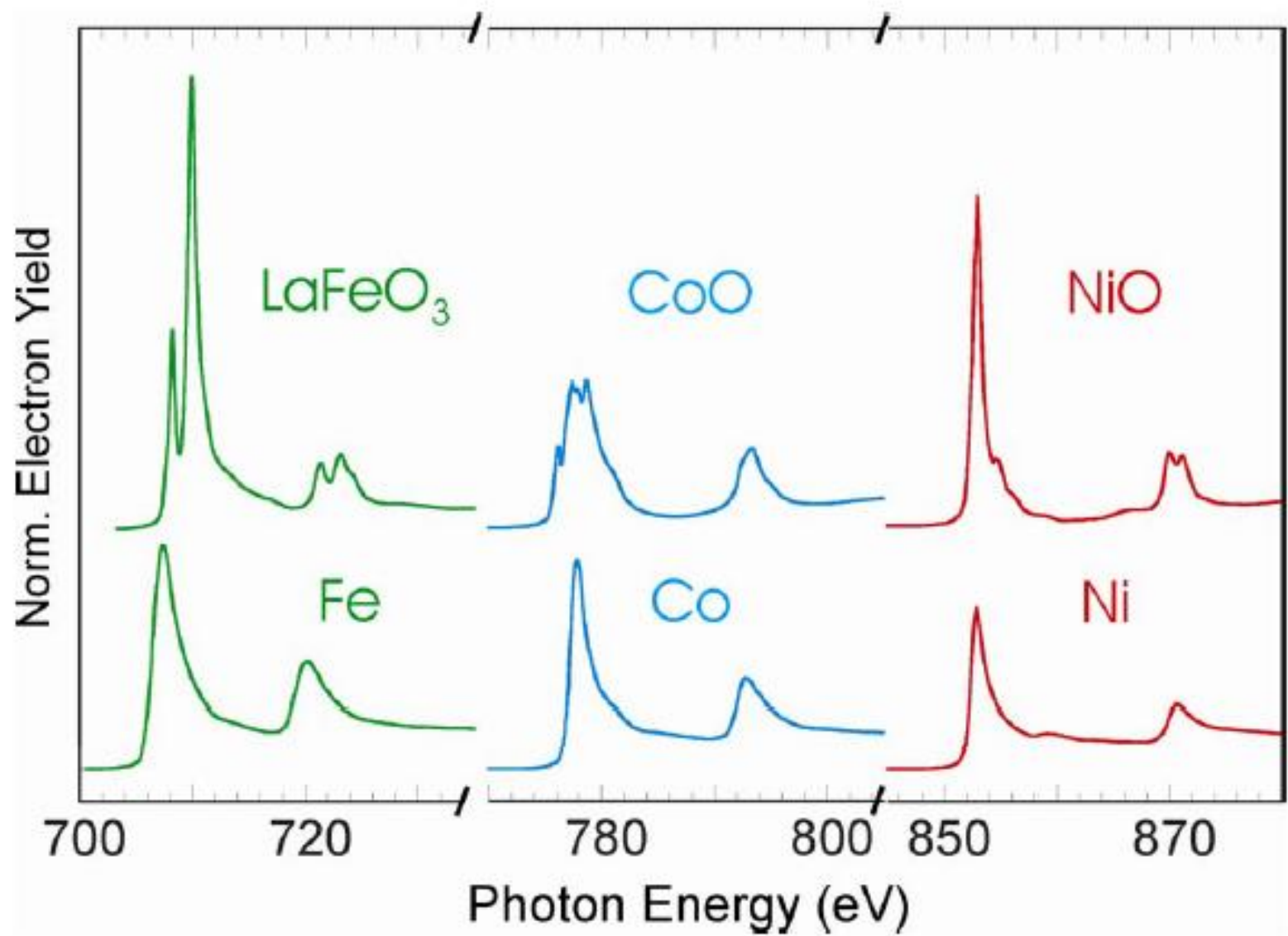
Photon Energy [eV]



Polymers: p-orbitals
 Trans. Met.: d-orbitals
 Rare Earths: f-orbitals

1s K-edge
 2p L edge
 3d M edge

Rich “multiplet structure” reveals local bonding



Orbital Order

Jahn-Teller distortion and/ or Superexchange lead to orbital order

Excitations are different!

Local crystal field excitations vs. dispersing orbitons

Superexchange: spins and orbitals entangle.

Jahn-Teller: spins and orbitals decouple, orbitals frozen out at low temperatures.

Elementary excitations in condensed matter

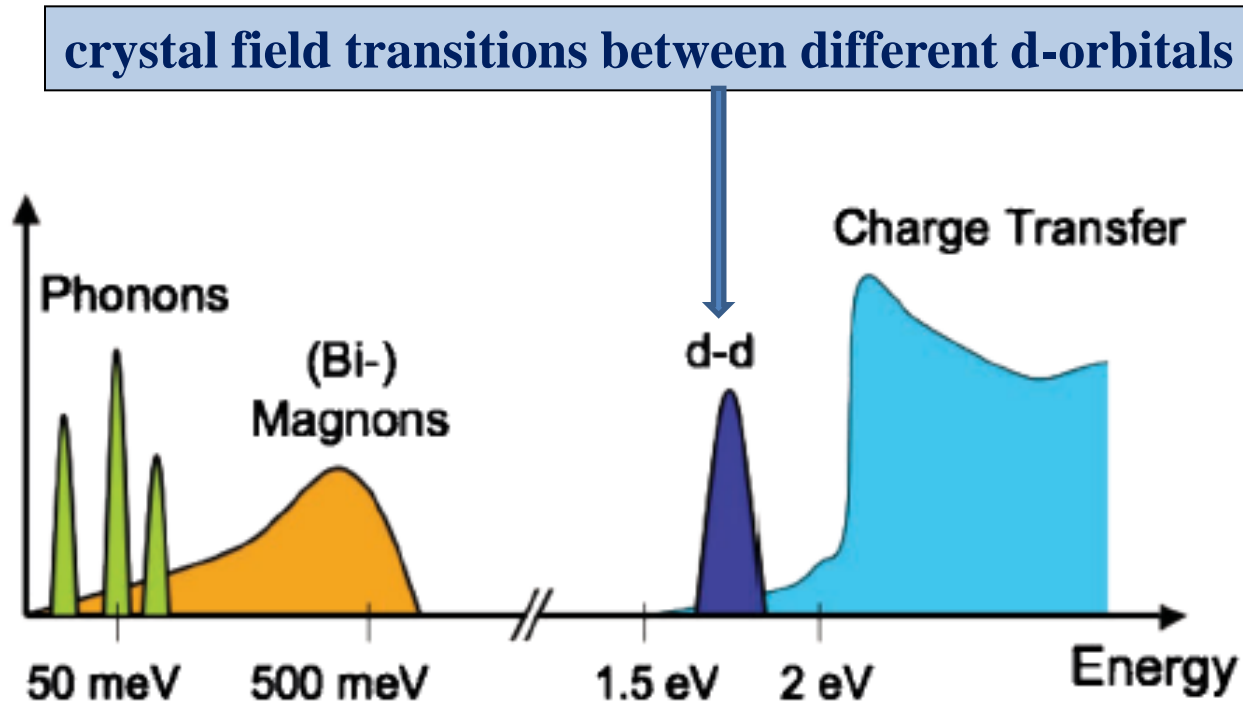


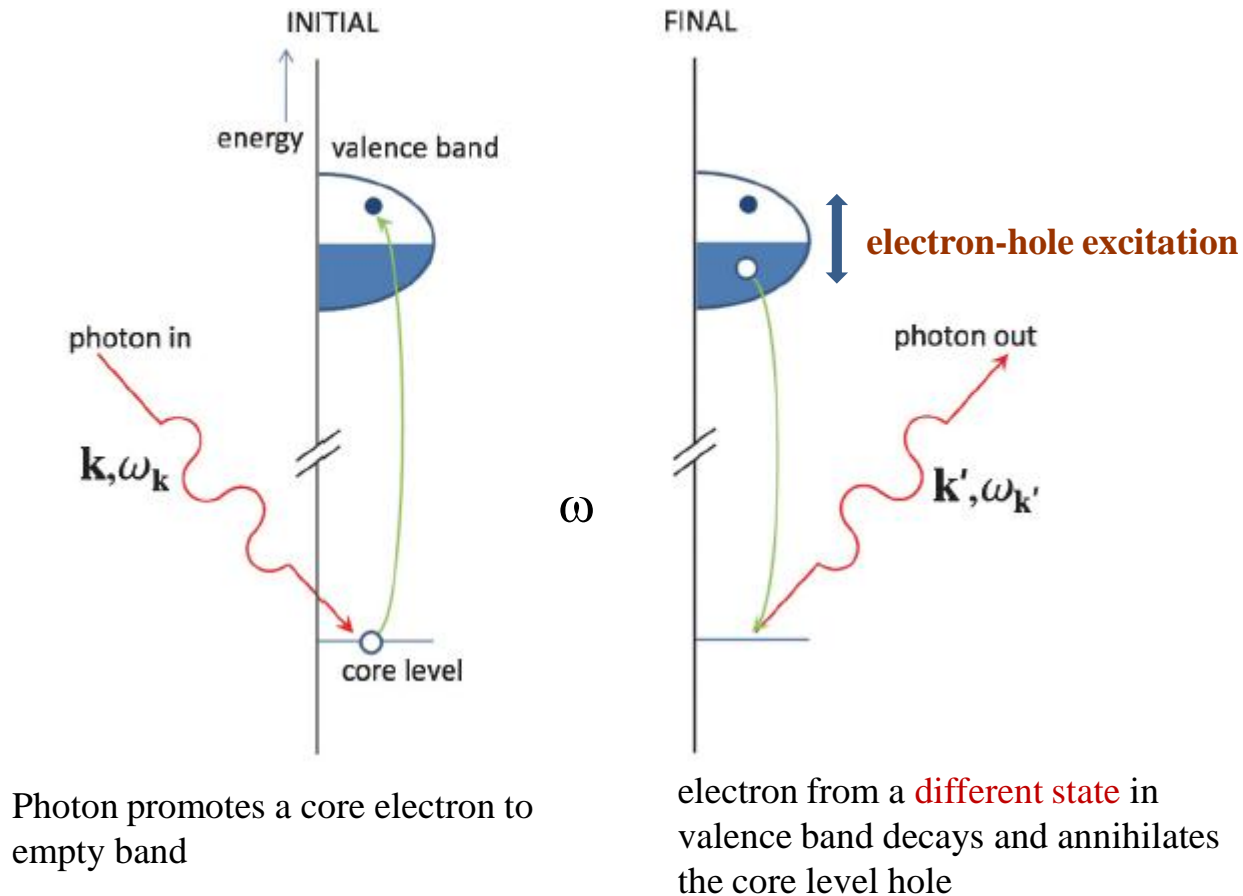
FIG. 6 (color). Different elementary excitations in condensed matter systems and their approximate energy scales in strongly correlated electron materials such as transition-metal oxides.

Resonant Inelastic X-ray Scattering (RIXS)

For an excellent review see:

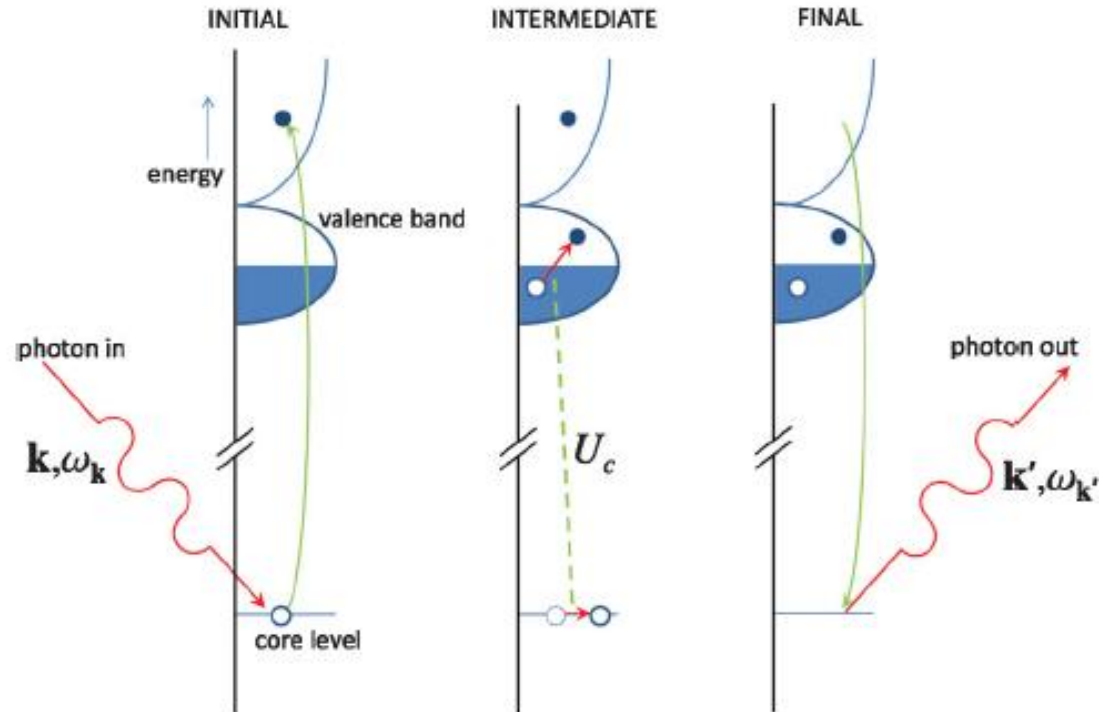
Amen et. al; Rev. Mod. Phys. 83 705 (2011)

Direct RIXS



RIXS creates excitation in the valence band with momentum $\hbar\mathbf{k}' - \hbar\mathbf{k}$ and energy $\hbar\omega_{\mathbf{k}'} - \hbar\omega_{\mathbf{k}}$

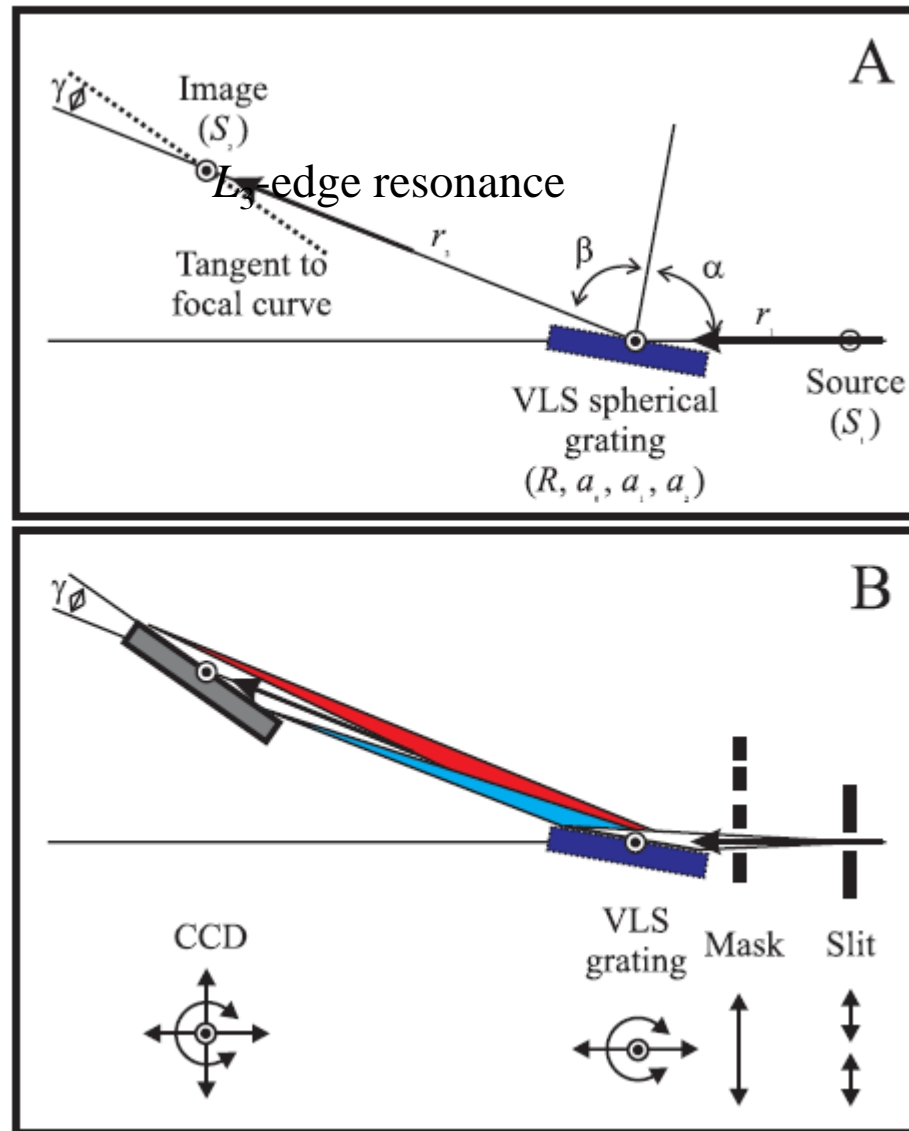
Indirect RIXS

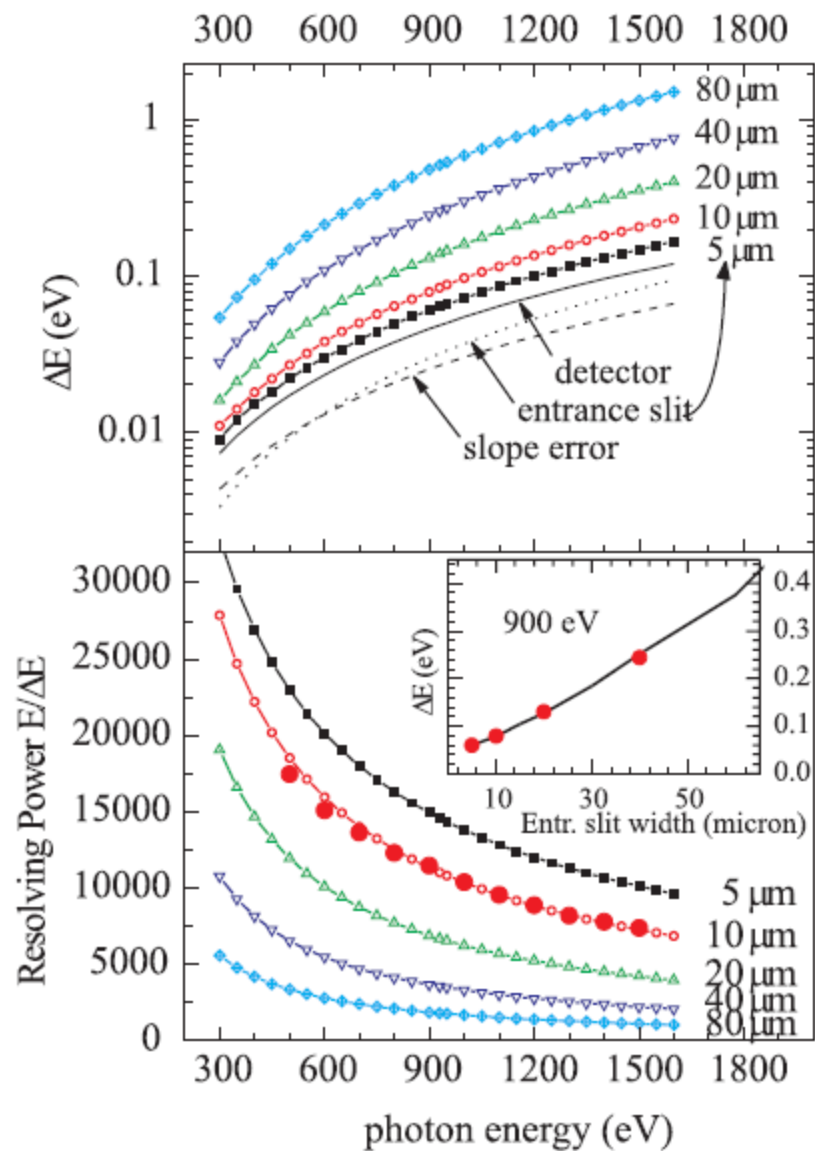


electron is excited from a deep-lying core level into the valence shell. Excitations are created through the Coulomb interaction U_c between the core hole (and in some cases the excited electron) and the valence electrons. The photoelectron subsequently decays leaving behind an excitation in the valence band with momentum $\hbar\mathbf{k}' - \hbar\mathbf{k}$ and energy $\hbar\omega_{\mathbf{k}'} - \hbar\omega_{\mathbf{k}}$

The measured photon energy transfer and momentum transfer in a RIXS experiment is directly related to the energy and momentum of the created excitations (spinons, orbitons)

RIXS Instrument for soft x-rays: SAXS at the Swiss Light Source





RIXS Instrument for hard x-rays: IDB 30 beamline at APS, USA

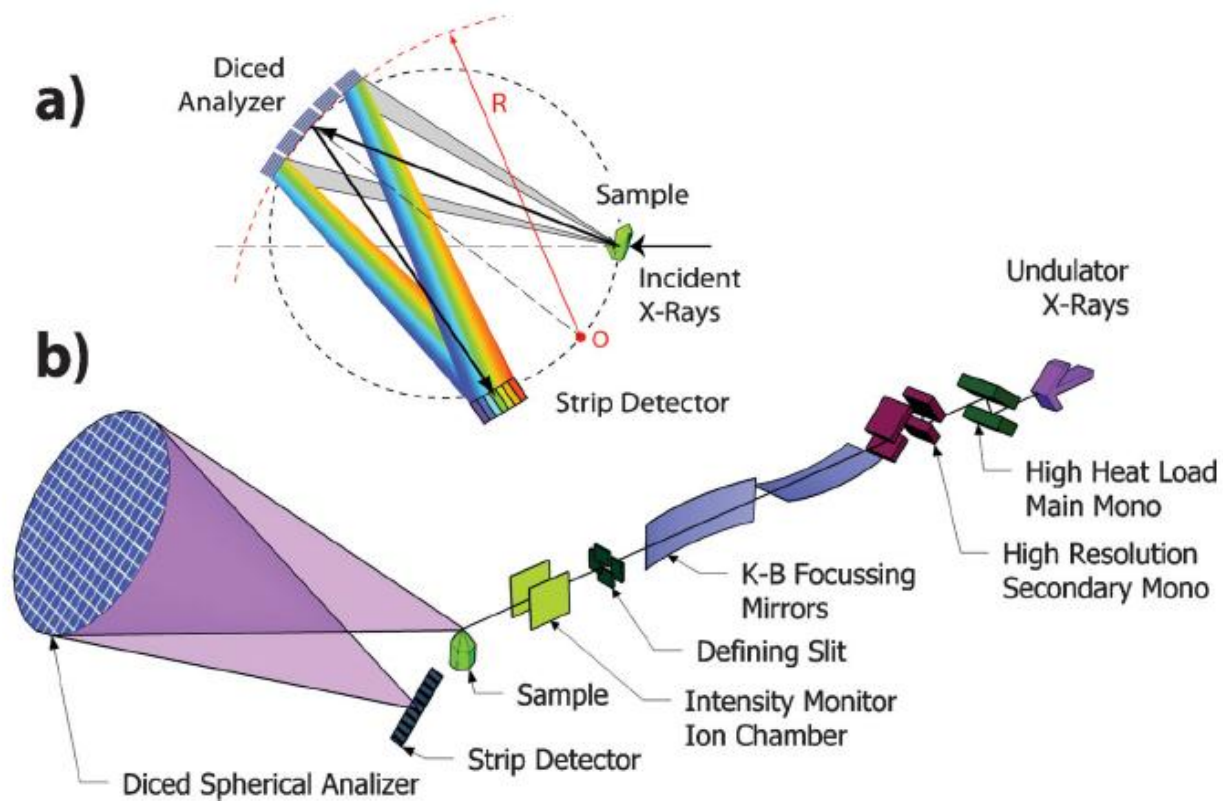
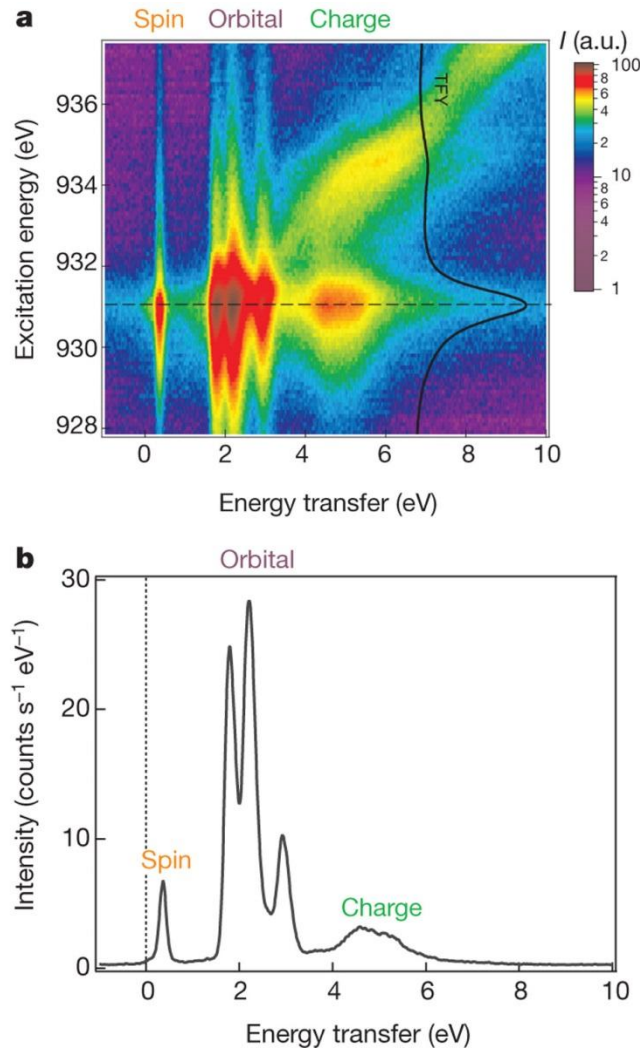


FIG. 15 (color). Schematic of optical components in the 30 IDB inelastic x-ray scattering beam line at the Advanced Photon Source, USA. Figure courtesy of D. Casa (APS).

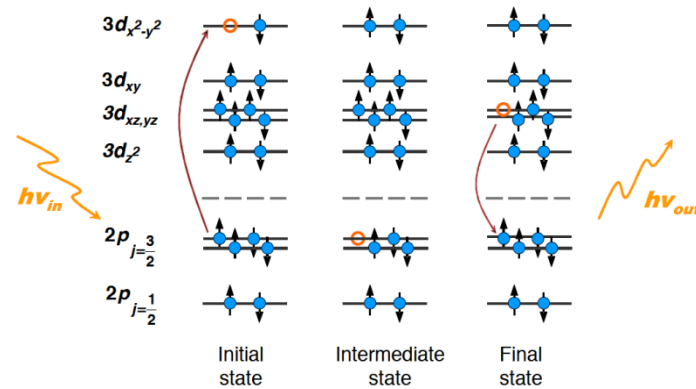
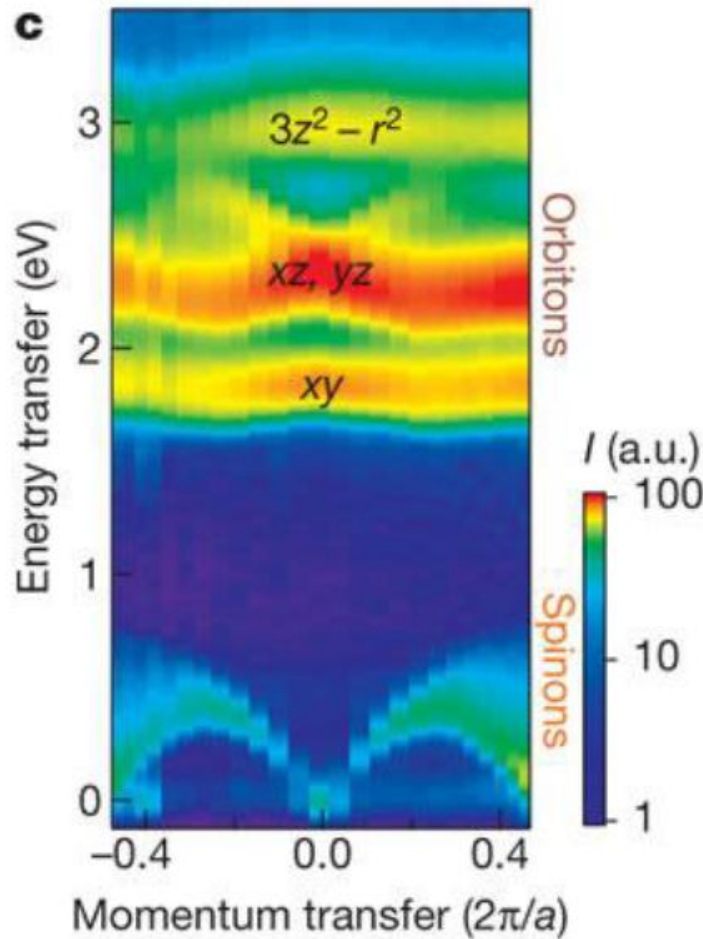
Energy dependence of elementary excitations in Sr_2CuO_3 observed with RIXS at the copper L_3 -edge resonance.



J Schlappa *et al.* *Nature* **485**, 82 (2012)

dispersion of orbital and spin excitations across the first Brillouin zone

J Schlappa *et al. Nature* **485**, 82 (2012)



orbital excitation from the ground-state copper $3d x^2 - y^2$ orbital to an excited copper $3d xy$ or xz orbital

