

# Neutronenstreuung : eine Einführung anhand aktueller Fragestellungen

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WS 2007/08  
Universität zu Köln



- von der Strukturbestimmung an Pulvern  
bis zur Analyse magnetischer Anregungen  
an Einkristallen

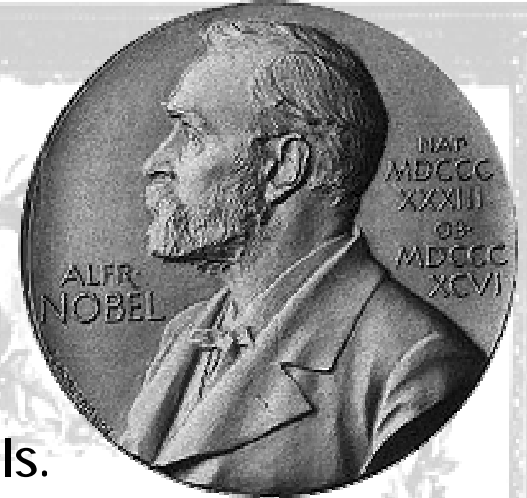
0. Einführung : Neutron : Geschichte Eigenschaften und Quellen

# Wilhelm Conrad Röntgen 1845-1923

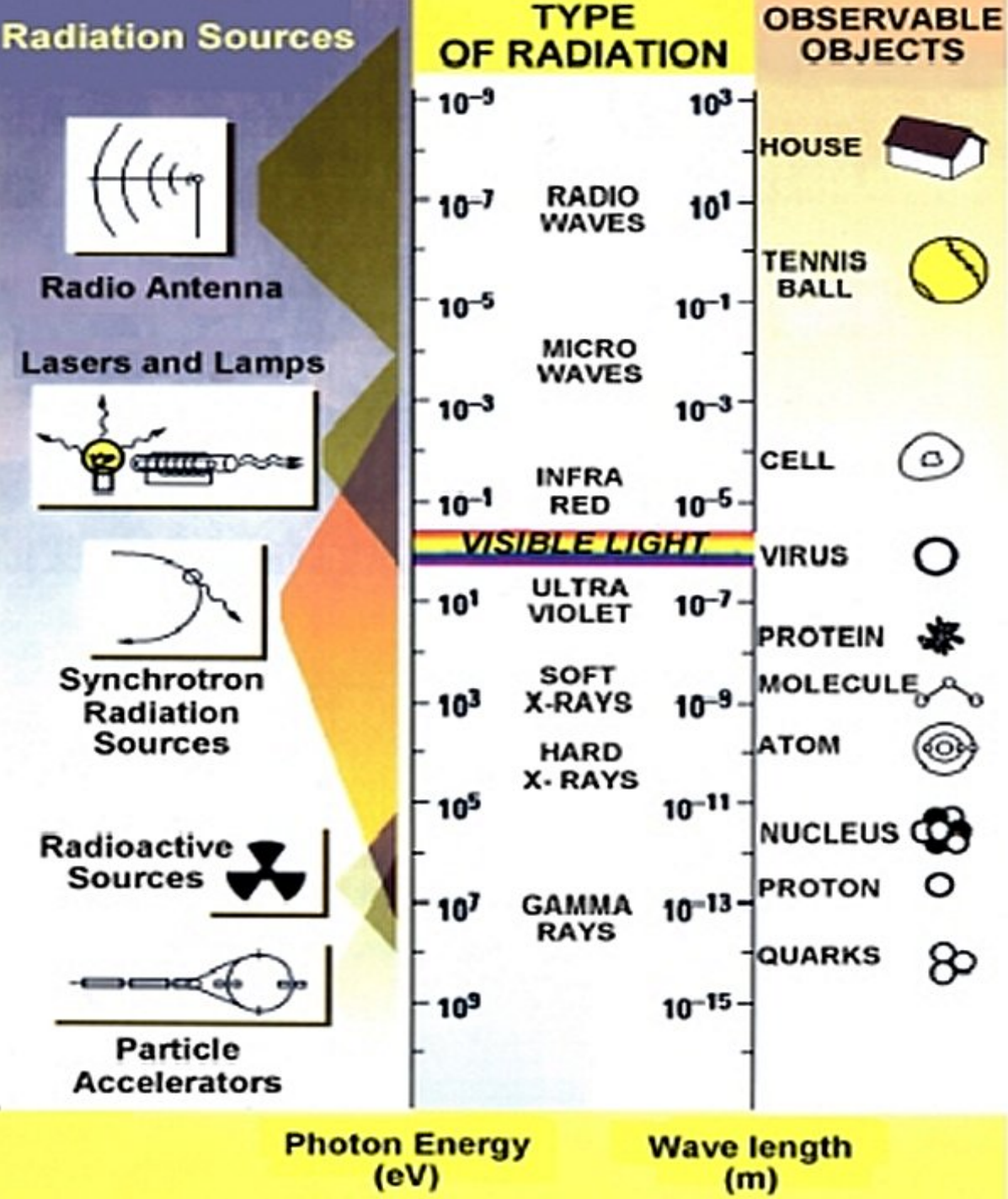


1895: Discovery of  
X-Rays

# Nobel Prizes for Research with X-Rays



- 1901 W. C. Röntgen in Physics for the discovery of x-rays.
- 1914 M. von Laue in Physics for x-ray diffraction from crystals.
- 1915 W. H. Bragg and W. L. Bragg in Physics for crystal structure determination.
- 1917 C. G. Barkla in Physics for characteristic radiation of elements.
- 1924 K. M. G. Siegbahn in Physics for x-ray spectroscopy.
- 1927 A. H. Compton in Physics for scattering of x-rays by electrons.
- 1936 P. Debye in Chemistry for diffraction of x-rays and electrons in gases.
- 1962 M. Perutz and J. Kendrew in Chemistry for the structure of hemoglobin.
- 1962 J. Watson, M. Wilkins, and F. Crick in Medicine for the structure of DNA.
- 1979 A. McLeod Cormack and G. Newbold Hounsfield in Medicine for computed axial tomography.
- 1981 K. M. Siegbahn in Physics for high resolution electron spectroscopy.
- 1985 H. Hauptman and J. Karle in Chemistry for direct methods to determine x-ray structures.
- 1988 J. Deisenhofer, R. Huber, and H. Michel in Chemistry for the structures of proteins that are crucial to photosynthesis.



Wavelength  
 $\approx$   
 Object Size  
 $\approx$   
 Angstroms  
 for Condensed  
 Matter Research

$$\lambda [\text{\AA}] = \frac{12.398}{E_{ph} [\text{keV}]}$$

# Nobelpreise Neutronen

The Nobel Prize in  
Physics 1935



James Chadwick



"for the discovery  
of the neutron"

The Nobel Prize in  
Physics 1994

"In simple terms,  
*Clifford G. Shull (1915-2001)*  
has helped answer the question of  
**where atoms are,**

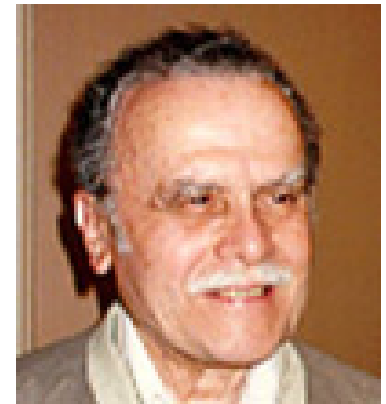
and

*Bertram N. Brockhouse*  
(1918-2003)  
the question of **what atoms do**",  
(Nobel citation)

"for pioneering contributions to the  
development of neutron scattering  
techniques for studies of  
condensed matter"

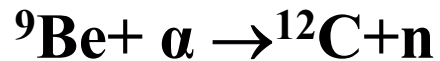
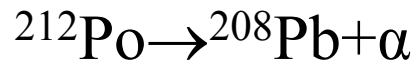


"for the  
development of  
the neutron  
diffraction  
technique"

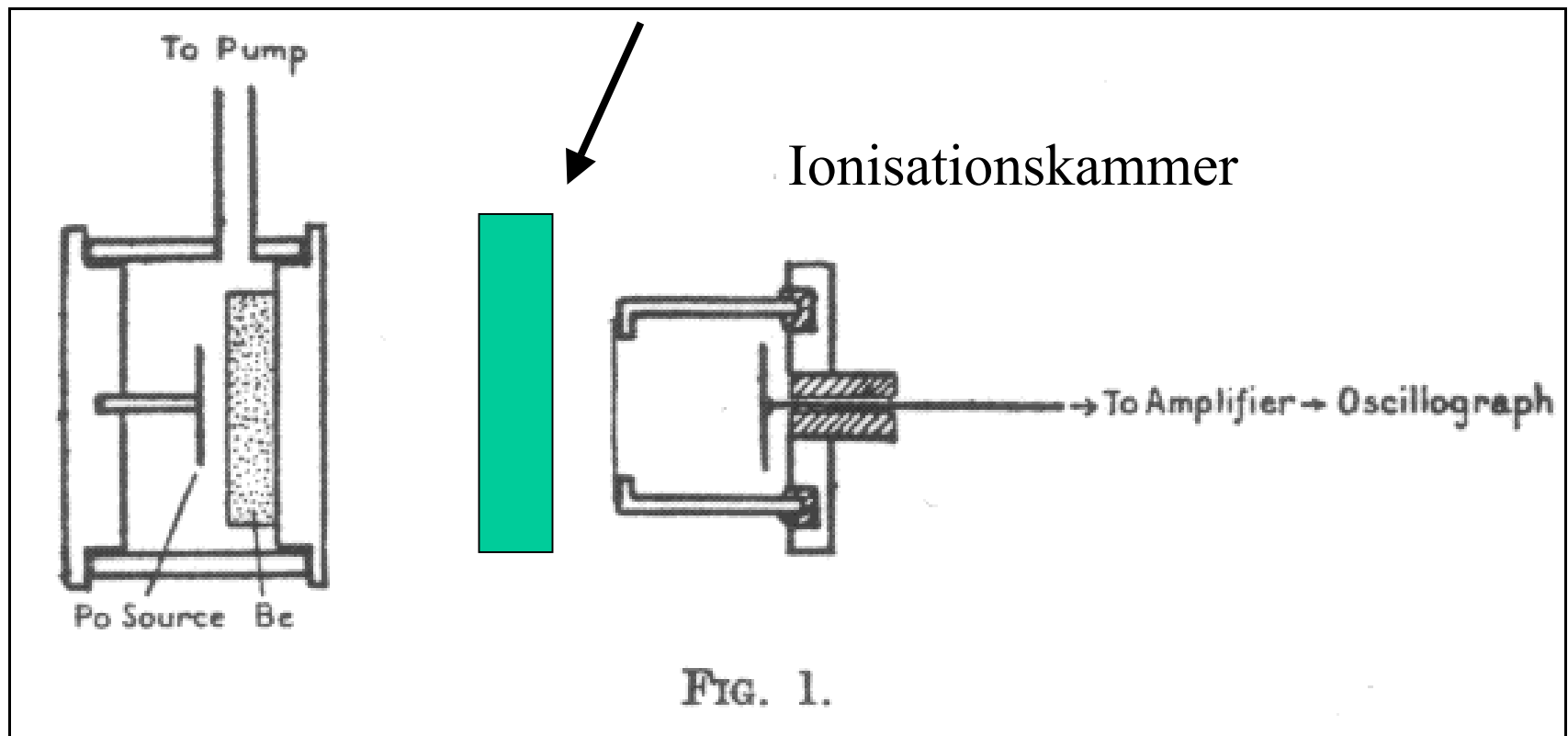


"for the  
development of  
neutron  
spectroscopy"

# 1932: Entdeckung von Chadwick



verschiedene Absorber (Paraffin (H), He, Li)





Letters to the Editor

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Possible Existence of a Neutron

It has been shown by Chadwick and others that boron when bombarded by  $\alpha$ -particles of penetrating power a millionth of gram penetrating power, which have absorption coefficient in lead of about  $10^{10}$  cm<sup>-1</sup>, liberate ions, electrons and  $\gamma$ -rays, found also according to the calculations predicted by the hypothesis advanced in a recent note in this journal, that the ionization produced when neutron-containing hydrogen was placed in focus of the window. The effect appeared to be due to the emission of particles with velocities up to a maximum of nearly  $2 \times 10^8$  cm. per sec. They suggested that the ionization of energy in the proton was by a process similar to the Compton effect, and estimated that the boron nuclei have the maximum energy of  $10 \times 10^8$  electron volts.

I have made some experiments using the value estimated in this note for the properties of this relative motion to boron. The value would consist of a small ionization chamber connected to an amplifier, and the window production of ions by the entry of a particle, such as a proton or  $\alpha$ -particle, is recorded by the deflection of an electrometer. These experiments have shown that the relative motion produced from hydrogen, helium, lithium, boron, carbon, oxygen, etc., all give the same results. The particles emitted from hydrogen, helium, or oxygen range and ionizing power. The process with oxygen up to about  $10 \times 10^8$  cm. per sec. The particles from the other elements have a large ionizing power, and appear to be in such a small range as to be absorbed.

If we consider the direction of the proton in a Compton effect from a quantity of  $10 \times 10^8$  electron volts, then the oxygen need more energy by a smaller amount should have an energy not greater than about 10,000 e.v., should produce not more than about 10,000 ions, and have a range in air as R.T.P. of about 10 cm. Actually some of the most active ionization produced at least 10,000 ions. In collaboration with Dr. Rutherford, I have observed the result seems to be an ionization chamber, and that ions, estimated readily, are consistent in such as  $10^8$  cm. as R.T.P.

These results, and others I have obtained in the course of the work, are very difficult to explain on the assumption that the particles from boron are  $\alpha$ -particles, electrons,  $\gamma$ -rays, and neutrons, as is to be expected in the literature. The different particles, however, it is to be assumed that the relative motion of particles of about  $10^8$  cm. per sec. is consistent. The nature of the particles in the  $10^8$  window may be supposed to result in the formation of a  $\alpha$ -particle and the emission of the window. From the energy relations of the process the velocity of the neutrons emitted in the forward direction may well be about  $2 \times 10^8$  cm. per sec. The velocity of the neutrons with the atoms through which it passes give rise to the result shown, and the observed energies of the result shown are in full agreement with this view. Moreover, I have observed that the proton emitted from hydrogen by the relative motion in the opposite direction to that of the moving  $\alpha$ -particle appear to have a much smaller range than those emitted by the forward motion.

The same results are found experimentally in the various experiments.

It is to be supposed that the relative motion of oxygen, then the motion of the  $\alpha$ -particle by the  $10^8$  window will form a  $10^8$  window. The true value of  $10^8$  is known with sufficient accuracy to show that the energy of the particles emitted in the process must be greater than about  $10 \times 10^8$  e.v. It is difficult to make such a question responsible for the effect observed.

It is to be supposed that many of the effects of a neutron in passing through matter should resemble those of a quantum of high energy, and it is not easy to reach the first division before the ionization. It is to be supposed that the motion of the neutron is in focus of the window, with the quantum of energy that only be emitted if the concentration of energy and momentum be concentrated at some point.

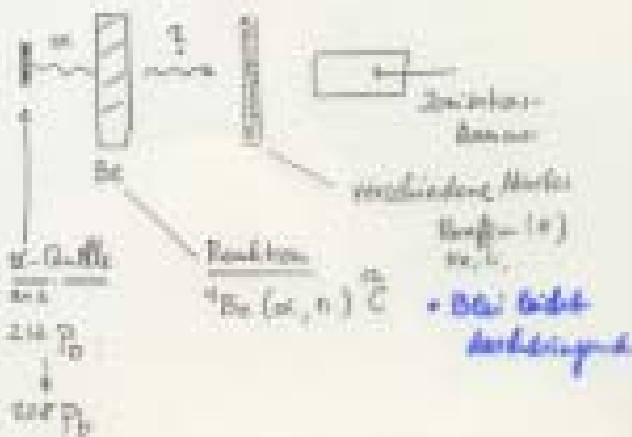
Cambridge Laboratory, Cambridge, Feb. 17.

J. Chadwick

- ① Be<sup>9</sup> +  $\alpha$  He<sup>4</sup> → <sup>12</sup>C + n
- ②  $\alpha$  He<sup>4</sup> + <sup>10</sup>B → <sup>11</sup>B + n

such "Compton Effect" under Rutherford  
 as known experimentally. Note further  
 — felds Erklarung!  
 in Annalen + P. Chadwick

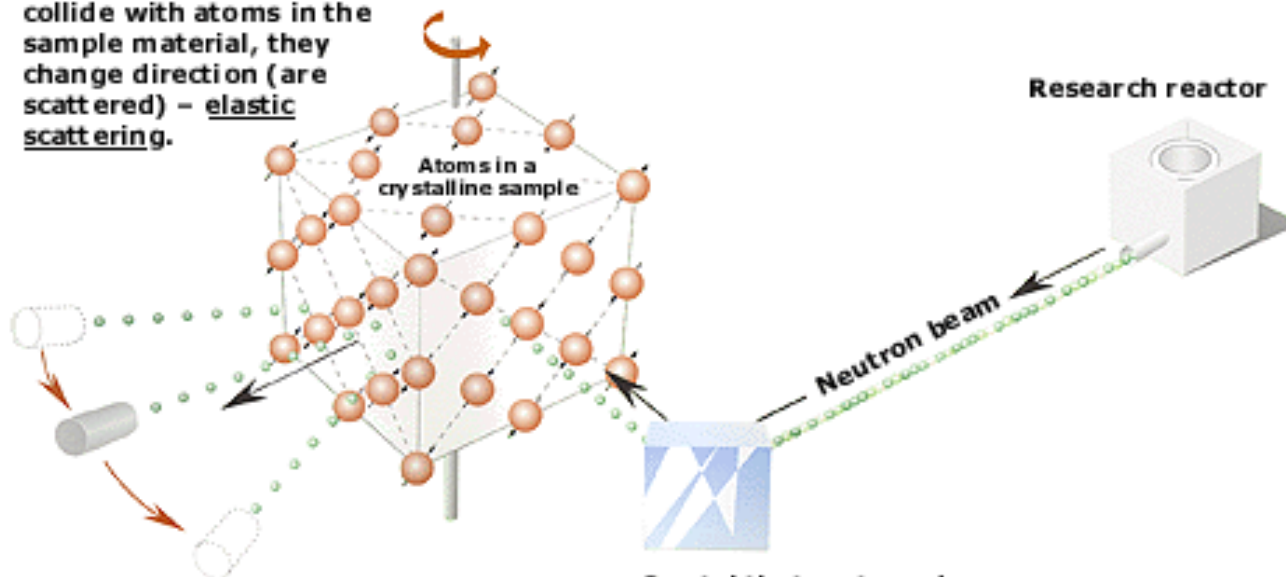
③ Chadwick, Annalen H. 1. und 17-2. 22.



# Neutrons show where atoms are



When the neutrons collide with atoms in the sample material, they change direction (are scattered) – elastic scattering.



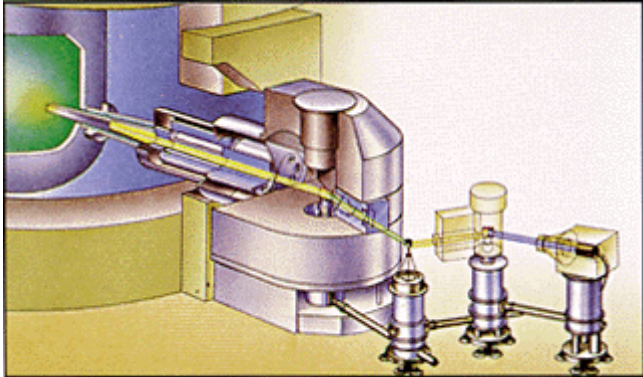
Detectors record the directions of the neutrons and a diffraction pattern is obtained.

The pattern shows the positions of the atoms relative to one another.

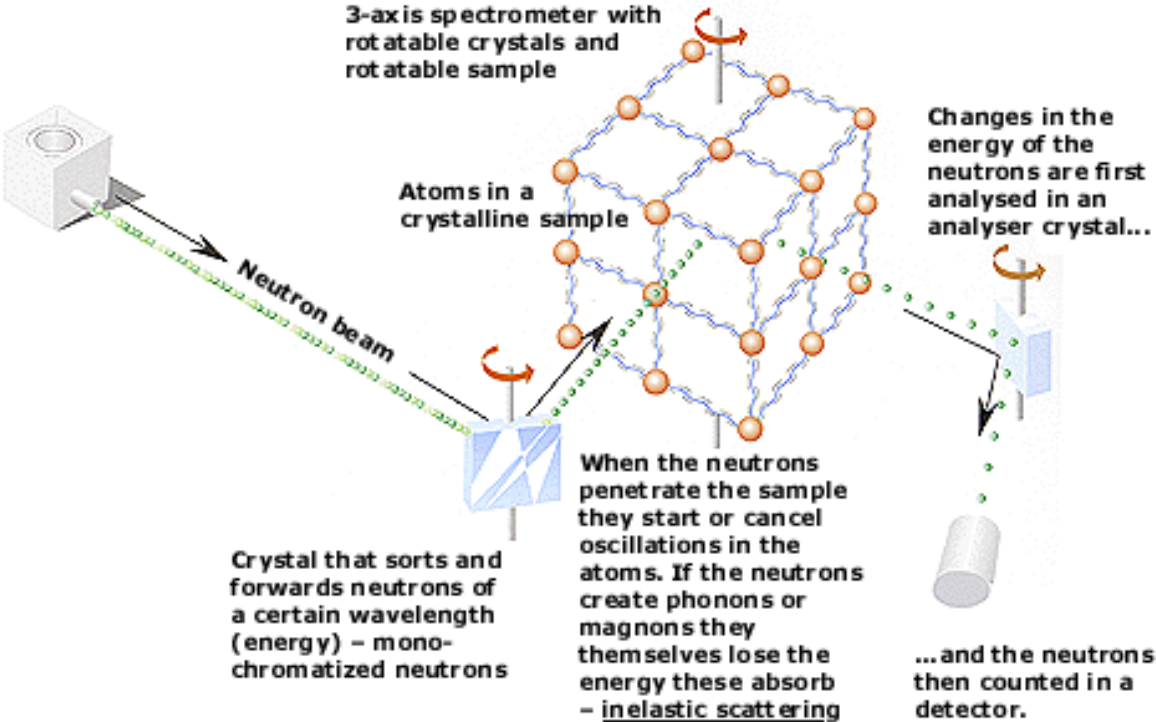
Crystal that sorts and forwards neutrons of a certain wavelength (energy) – monochromatized neutrons



# Neutrons show what atoms do



3-axis spectrometer



**1942** – Nobel Laureate Enrico Fermi led the group that first demonstrated a controlled nuclear reaction (on December 2, 1942, shown below).

On November 4, 1943, Fermi witnessed the initial operation of the Graphite Reactor in Oak Ridge.



## **1943-**

Designed using the results of the Chicago experiment, the Graphite Reactor produced small amounts of plutonium, setting the stage for large-scale plutonium production by reactors in Hanford, Washington. It was the world's first isotope-production reactor.



## **1943-**

Eugene Wigner, later an ORNL director who won a Nobel Prize for physics, predicted that radiation damage could be a problem for reactors. The behavior of irradiated materials was studied at ORNL, which became a leading materials research laboratory.

# Graphit-Reaktor in Oak-Ridge (Tennessee)



**1945** – Ion-exchange chromatography used at the Graphite Reactor enabled the discovery of promethium (element 61).



**1945** – The first neutron-scattering studies using a reactor were performed at the Graphite Reactor by Ernie Wollan (left) and Clifford Shull (right), who won a Nobel Prize for physics in 1954 for advancing the understanding of the positions of atoms and molecules in materials.

Wollan, Shull, Koehler the pioneers ! ! !  
today **High Flux Isotope Reactor (HFIR) in 1966, ORNL**

## Neutron Diffraction Study of the Magnetic Properties of the Series of Perovskite-Type Compounds $[(1-x)\text{La}, x\text{Ca}]\text{MnO}_3^\dagger$

E. O. WOLLAN AND W. C. KOEHLER  
*Oak Ridge National Laboratory, Oak Ridge, Tennessee*

(Received May 9, 1955)

A study has been made of the magnetic properties of the series of perovskite-type compounds  $[(1-x)\text{La}, x\text{Ca}]\text{MnO}_3$ . The investigations have been made primarily by neutron diffraction methods, but x-ray diffraction measurements of lattice distortions and ferromagnetic saturation data are also included. This series of compounds exhibits ferromagnetic and antiferromagnetic properties which depend upon the relative trivalent and tetravalent manganese ion content. The samples are purely ferromagnetic over a relatively narrow range of composition ( $x \sim 0.35$ ) and show simultaneous occurrence of ferromagnetic and antiferromagnetic phases in the ranges ( $0 < x < 0.25$ ) and ( $0.40 < x < 0.5$ ). Several types of antiferromagnetic structures at  $x=0$  and  $x > 0.5$  have also been determined. The growth and mixing of the various phases have been followed over the whole composition range, the ferromagnetic and antiferromagnetic moment contributions to the coherent reflections have been determined, and Curie and Néel temperatures have been measured. The results have been organized into a scheme of structures and structure transitions which is in remarkable accord with Goodenough's predictions based on a theory of semicovalent exchange.

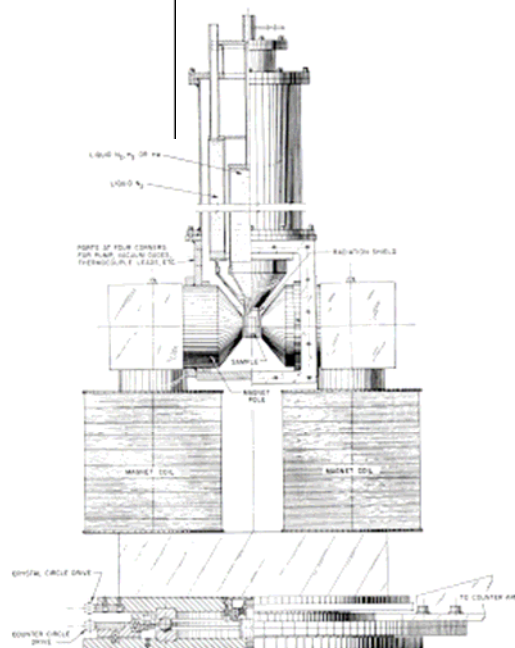


FIG. 1. Neutron spectrometer with crystal and magnet.

# Neutron Diffraction Study of the Magnetic Properties of the Series of Perovskite-Type Compounds $[(1-x)\text{La}, x\text{Ca}]\text{MnO}_3^\dagger$

E. O. WOLLAN AND W. C. KOEHLER

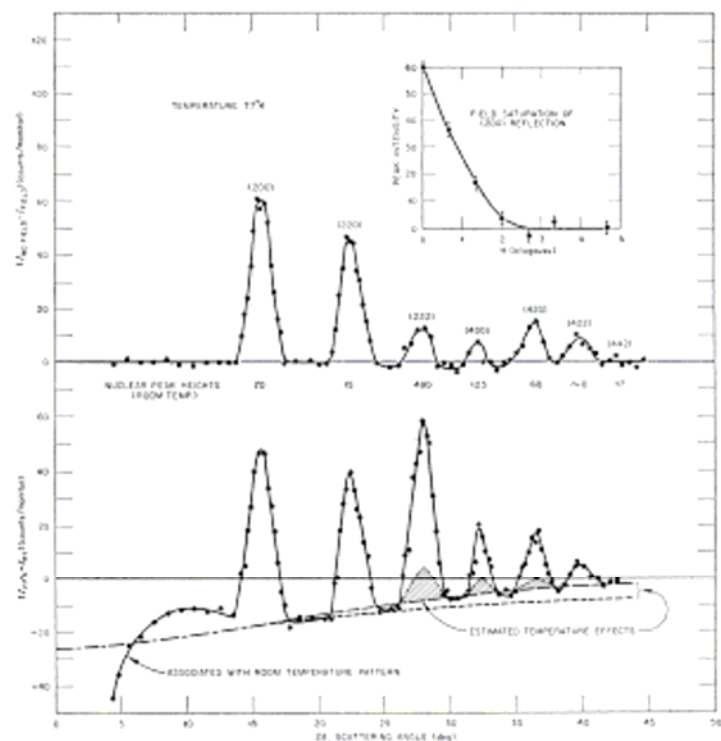


Fig. 9. Forsterite patterns for  $[(0.85\text{ La}-0.15\text{ Ca})\text{MnO}_3]$ , No. 70. Upper curve gives magnetic field difference pattern for maximum (top) to least (bottom) intensity. Lower curve gives comparative data from temperature difference patterns.

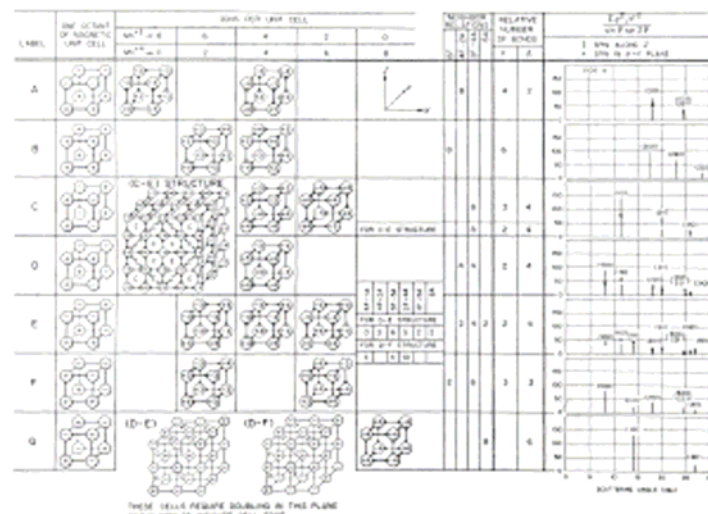


Fig. 10. Scheme of magnetic structures and related information. A, A', C, G, and (C-E) definitely observed and some evidence for B and F. In ordering schemes represent arrangements consistent with certain coupling criteria. Arrowheads are a schematic representation of Goodenough's superexchange coupling.



# Inelastische Neutronenstreuung : Bertram Brockhouse Chalk River Kanada (in den 50'ern)

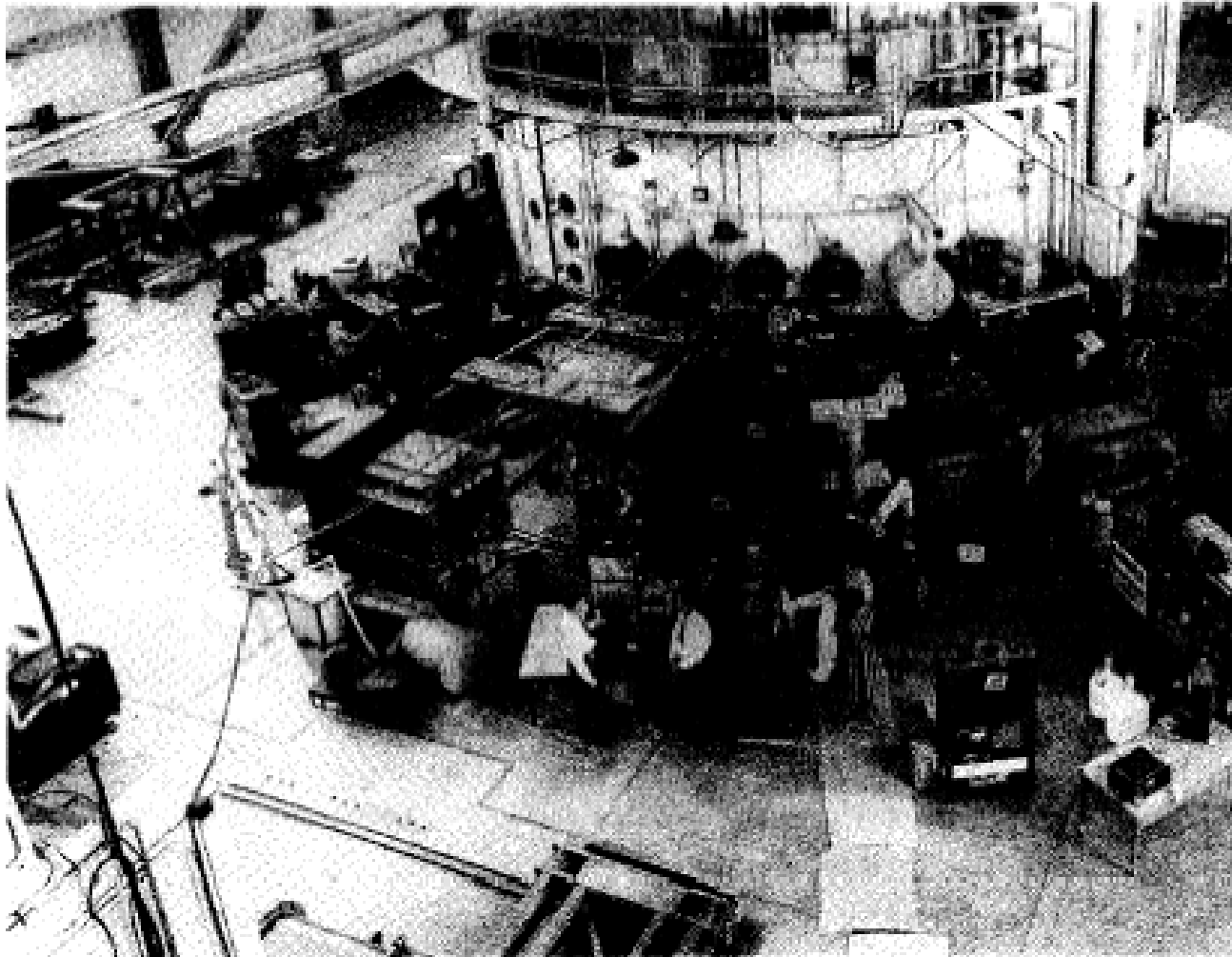


Fig. 1 The main floor of the NRU Reactor at the Chalk River Laboratory about 1950. The powder diffractometer constructed by Donald Hurst and associates is visible near the centre of the photograph. Most of the other equipment is concerned with nuclear physics or with the physics of the neutron itself. For reasons of space, each apparatus is located at the end of a long tube. (AECL photo)

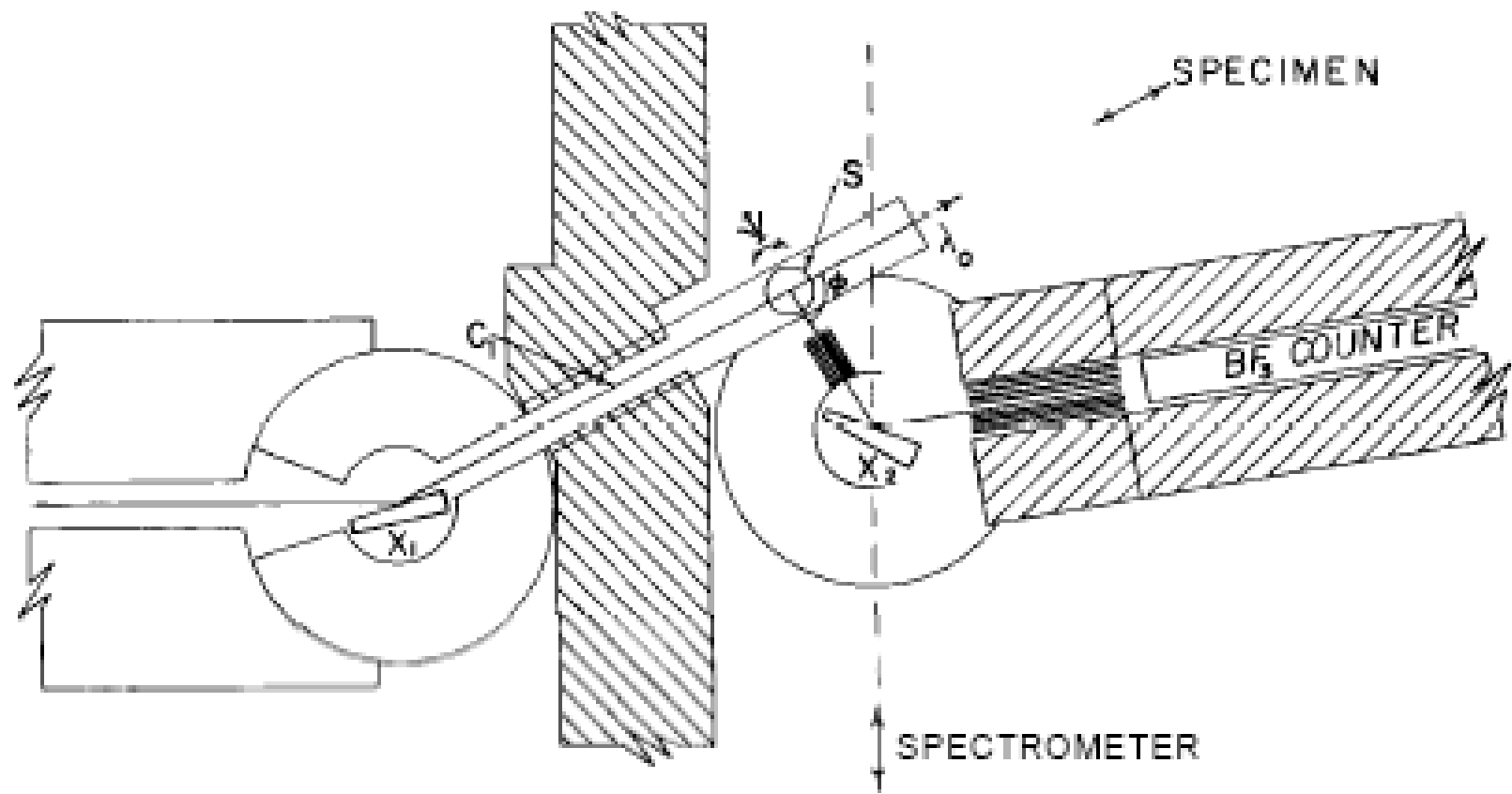
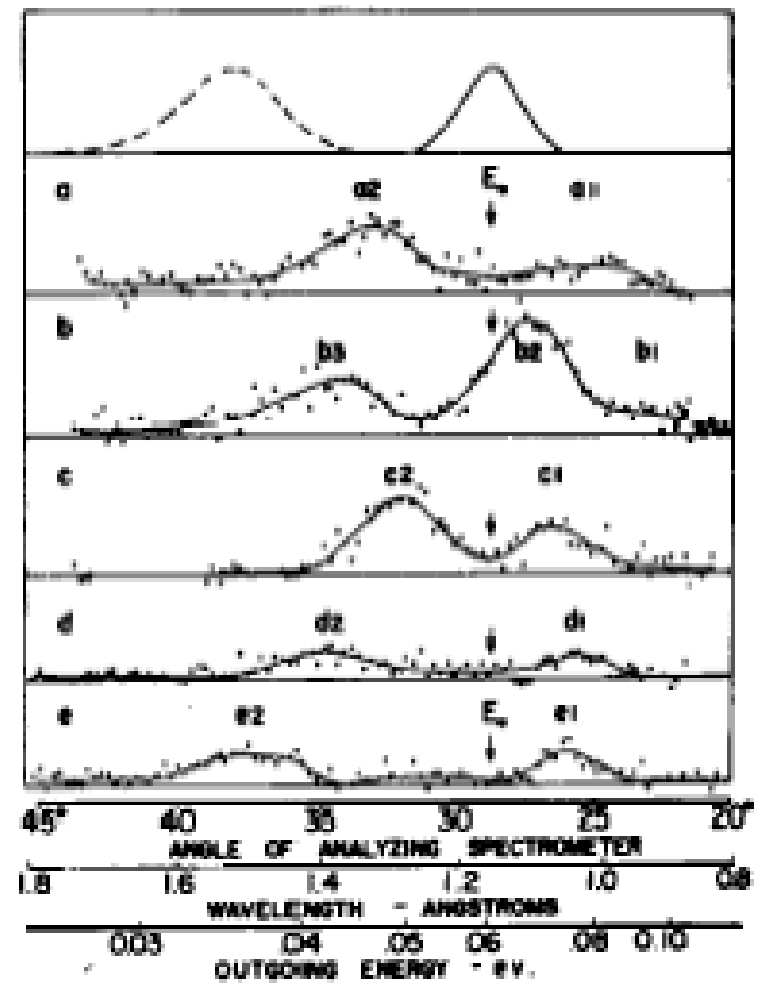
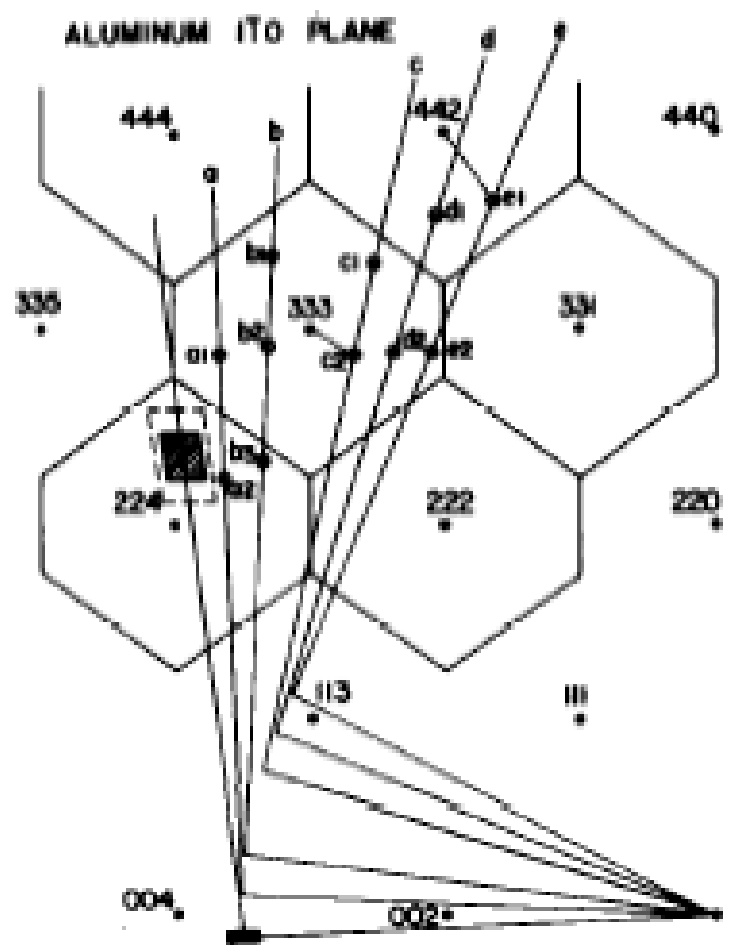


Fig. 3: The first crude version of the triple-axis crystal spectrometer [24 - 26]. Monoenergetic neutrons are selected by the large single crystal monochromator (X1) and impinge on the specimen (S), which is located on a table whose orientation ( $\psi$ ) in the horizontal plane can be selected. This table can be moved along the direction of the incident beam as desired. The analyzing spectrometer, which employs crystal X2, is a diffractometer (of especially large aperture) which can be translated as a unit; the angle ( $\theta$ ) through which the examined neutrons are scattered is determined by triangulation.



Fig. 9: Photograph (1959) of the original triple-axis spectrometer at the NRU reactor at Chalk River. The bank of 52 rotary switches, preset to go through an energy scan of up to 26 points, can be seen in the upper-centre. One of the three variables involved would traverse linearly through the domain of settings desired, while the other two advance nonlinearly according to the settings of the appropriate switches. (AECL photo)



# Neutrons



Neutrons are **NEUTRAL** particles. They

- are highly penetrating,
- can be used as nondestructive probes, and
- can be used to study samples in severe environments.



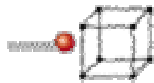
Neutrons have a **MAGNETIC** moment. They can be used to

- study microscopic magnetic structure,
- study magnetic fluctuations, and
- develop magnetic materials.



Neutrons have **SPIN**. They can be

- formed into polarized neutron beams,
- used to study nuclear (atomic) orientation, and
- used for coherent and incoherent scattering.



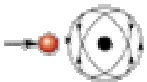
The **ENERGIES** of thermal neutrons are similar to the energies of elementary excitations in solids. Both have similar

- molecular vibrations,
- lattice modes, and
- dynamics of atomic motion.



The **WAVELENGTHS** of neutrons are similar to atomic spacings. They can determine

- structural sensitivity,
- structural information from  $10^{-13}$  to  $10^{-4}$  cm, and
- crystal structures and atomic spacings.



Neutrons "see" **NUCLEI**. They

- are sensitive to light atoms,
- can exploit isotopic substitution, and
- can use contrast variation to differentiate complex molecular structures.

$$M_n = 1.674928 \cdot 10^{-27} \text{kg}$$

$$= 1.001 M_{\text{Proton}}$$

$$\tau = 885 \text{ s } (\beta \text{ decay})$$

$$n \rightarrow p + e^- + \nu_e + 0.78 \text{ MeV}$$

$$n: \quad E = h^2 / 2M_n \lambda^2 = 81.1 \text{ meV} / \lambda^2$$

$$\text{photon: } E = hf = hc / \lambda = 12398 \text{ eV} / \lambda$$

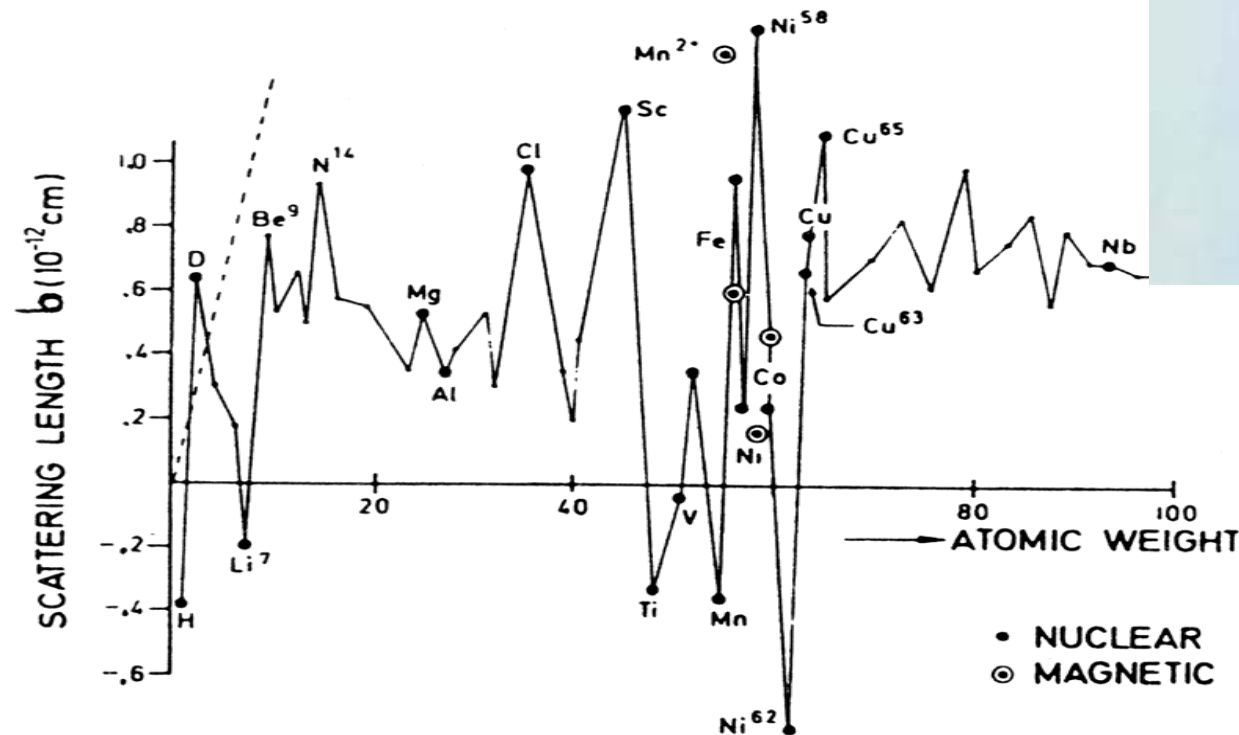
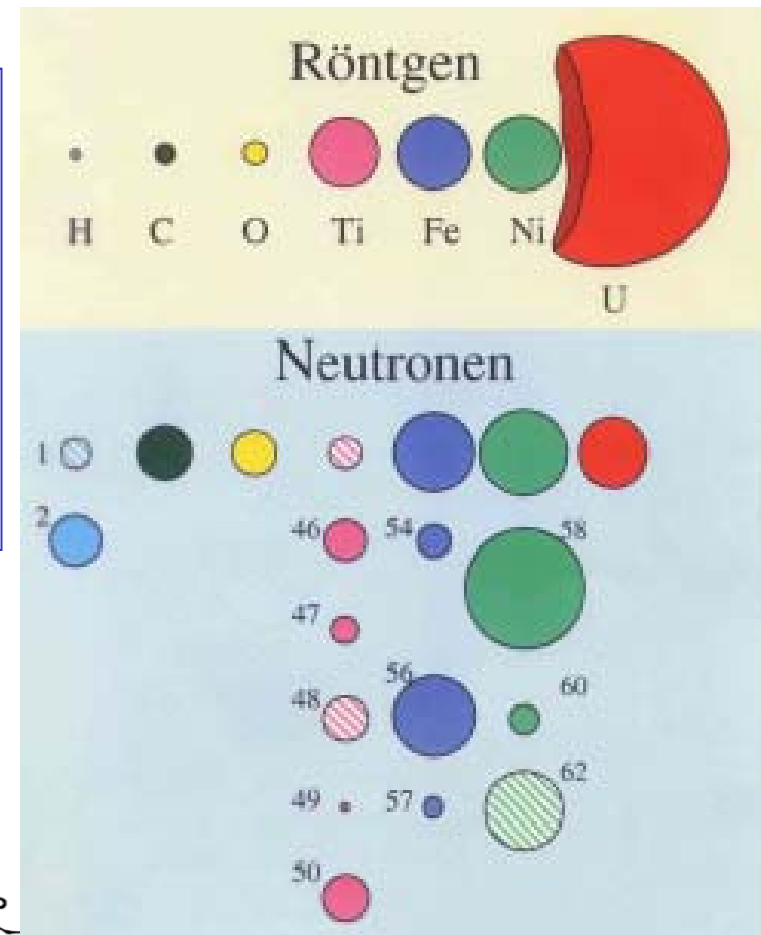
$$k = 2\pi / \lambda \quad p = h / \lambda$$

units:

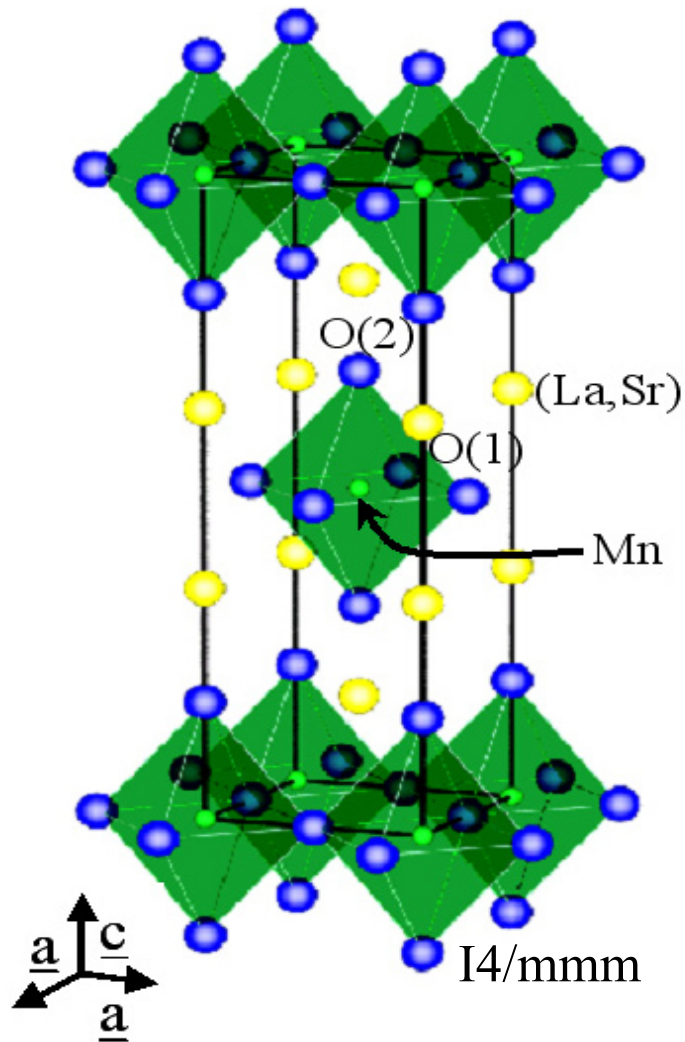
$$1 \text{ meV} = 11.6 \text{ K} = 8.066 / \text{cm} = 0.241 \text{ THz}$$

	$\lambda[\text{\AA}]$	$k[1/\text{\AA}]$	$v(\text{m/s})$	E	best $\Delta E/E$
Photon light	5000	$10^{-3}$	$3 \cdot 10^8$	eV	$10^{-8}$
X-ray	1	1	$3 \cdot 10^8$	keV	$10^{-6}$
electron	1	1	$6 \cdot 10^7$	150eV	$10^{-5}$
neutron	1	1	400	meV	$10^{-6}$

- Neutronen sind neutral ( $Q < 10^{-20}e$ )
- Wechselwirkung mit den Atomkernen
  - lokal
  - nicht direkt von Z abhängig
- **Bestimmung leichter und schwerer Atome !!!**



Neutrons show  
where atoms are

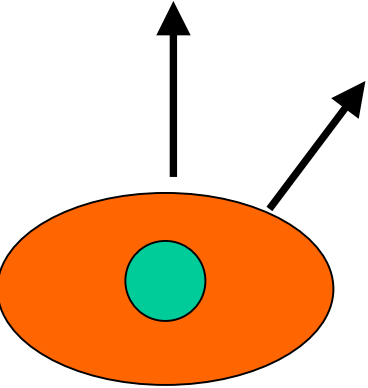


-Sauerstoffpositionen in Oxiden mit  
Seltenen Erden oder La →  
besser mit Neutronen !



# Analyse der Intensitäten

**Intensitäten der Reflexe**  
**→ komplette Strukturinformation**



$$F_G = \int_{\text{Zelle}} dV n(\mathbf{r}) \exp(-i\mathbf{G} \cdot \mathbf{r}) \quad n(\mathbf{r})\text{-Streudichte; Strukturfaktor}$$
$$= \sum_j f_j \cdot \exp(-i\mathbf{G} \cdot \mathbf{r}_j)$$

$$f_j = \int dV \cdot n_j(\rho) \exp(-i\mathbf{G} \cdot \rho) \quad \text{Atomformfaktor}$$

$$F_{(hkl)} = \sum_j f_j \exp[-i2\pi(x_j h + y_j k + z_j l)]$$

- weitere Verschmierung durch thermische Bewegung oder Unordnung

$$F_{(hkl)} = \sum_j f_j \exp[-i2\pi(x_j h + y_j k + z_j l)] * \exp\left(-\frac{1}{3}\langle u^2 \rangle G^2\right)$$

-aber : man misst  $F_{(hkl)}^2$  Verlust der Phaseninformation  
bei Zentrosymmetrie nur Vorzeichen !!

# Magnetische Eigenschaften

Ladung=0  
elektr. Dipolmoment=0

Spin:  $S = 1/2$

magnetisches Dipolmoment:

$$\begin{aligned}\mu &= 9.6491783 \cdot 10^{-27} \text{ J/T} \\ &= -1.913042 \mu_N\end{aligned}$$

$$\mu_N = \mu_B \cdot m_e/m_p = 0.5446 \cdot 10^{-3} \mu_B$$

„Erklärung“:

$$\begin{array}{ll} \text{DIRAC } \mu_{\text{Proton}} = 1 \mu_N & \text{exp } 2.8 \mu_N \\ \mu_{\text{Neutron}} = 0 & \text{exp } -1.9 \mu_N \end{array}$$

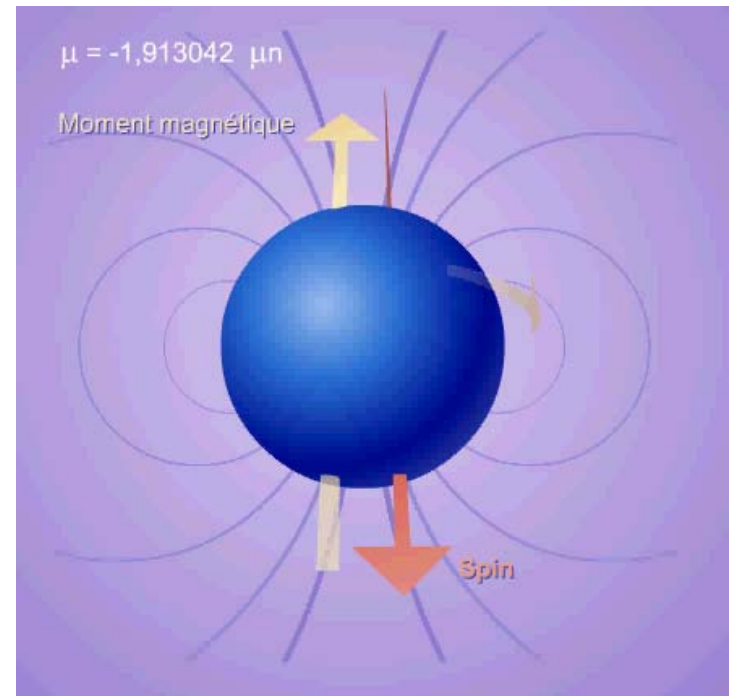
MESON  $p \leftrightarrow n + \pi^+$      $n \leftrightarrow p + \pi^-$

$\mu_{\text{Neutron}} \neq 0$ : Neutron  
ist 20% der Zeit  $p + \pi^-$ , wobei  
 $\pi^-$  Bahnmoment von  $10 \mu_N$   
hat.

QUARK u d d

# Magnetic properties of the neutron

$$\vec{\mu}_n = -\gamma\mu_B \frac{m_e}{m} \vec{\sigma}$$



# Magnetische Wechselwirkung

$$V_m(\mathbf{r}) = -\vec{\mu} \cdot \mathbf{B}(\mathbf{r}) = -\frac{\mu_0}{4\pi} g\gamma \frac{m_e}{m} \mu_B^2 \vec{\sigma} \cdot \nabla \times \left( \frac{\mathbf{S} \times \hat{\mathbf{R}}}{R^2} \right)$$

magnetische Streuung kann so stark wie Kernstreuung sein !

Spin-Dichte ist verteilt → Streuung nimmt mit Q ab

sensitiv zu magnetischem Dipolmoment (Spin) senkrecht zu Q !

# Anwendungen von magn.Einkristall-Beugung

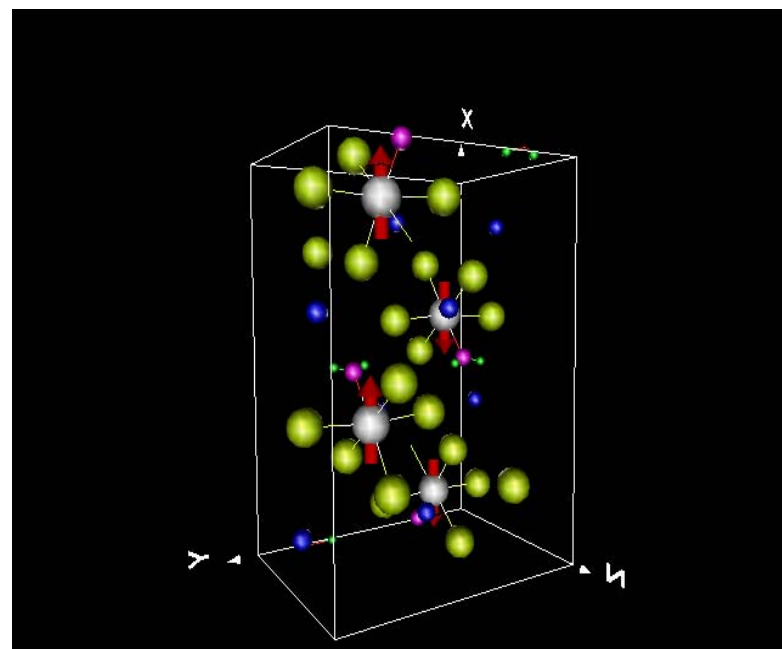
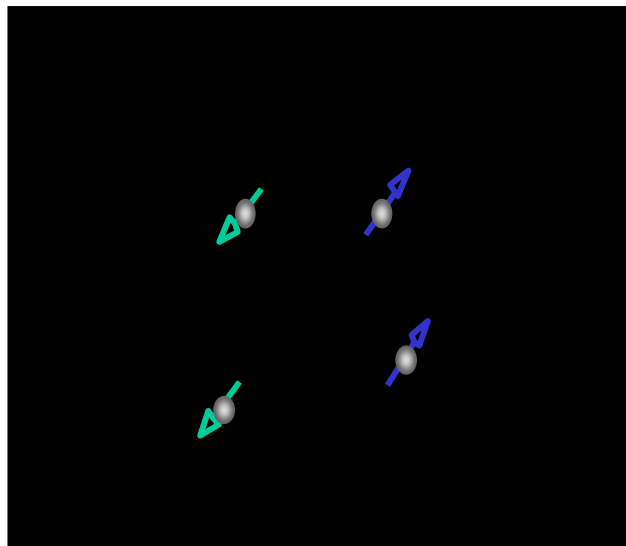
- Moments are very small
- Transition temperatures lower than 1 K
- Incommensurate magnetic structures
- Magnetic transitions induced by H or P
- Magnetism of absorbing elements (Gd, Sm)
- Mapping of the density of spin
- Molecular magnetism

# Magnetic structure



- Colinear AF with  $T_N = 14.06$  K
- Easy axis:  $a$
- Ferromagnetic planes  $\perp b$ -axis AF coupled

$$k = (0 \ 0 \ 0)$$

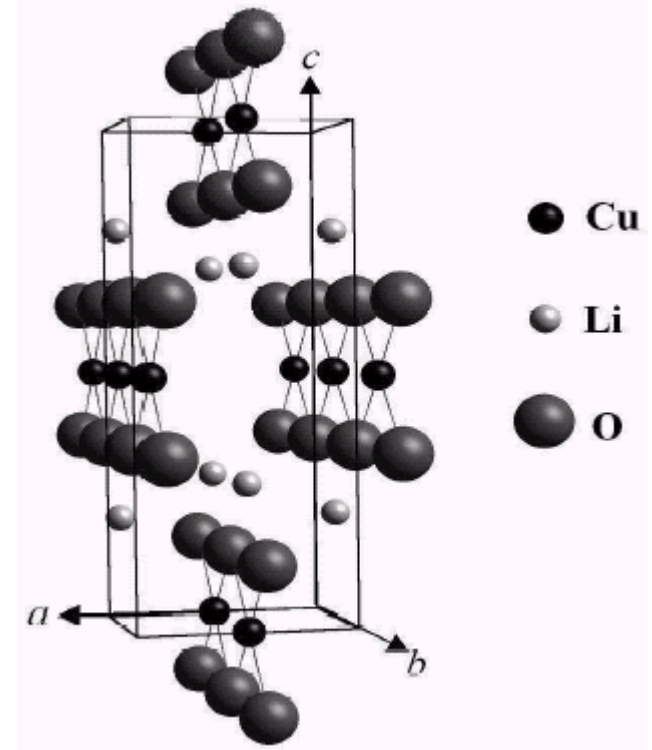
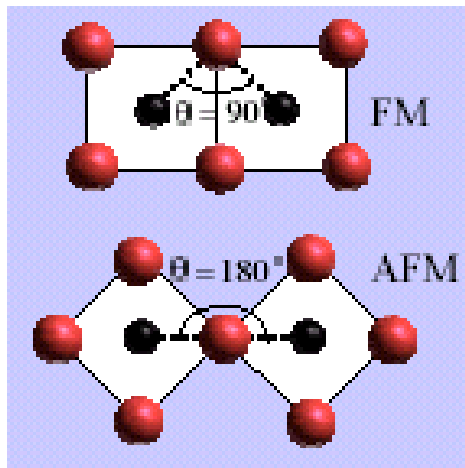


# Oxygen moment in $\text{Li}_2\text{CuO}_2$

E. Chung et al. 2003

1D chains of  $S=1/2$  Cu atoms

$\text{CuO}_4$  plaquettes (like in  $\text{CuGeO}_3$ , High- $T_c$  or chain-ladder system  $\text{La}_{14-x}\text{Ca}_x\text{Cu}_{24}\text{O}_{41}$ )

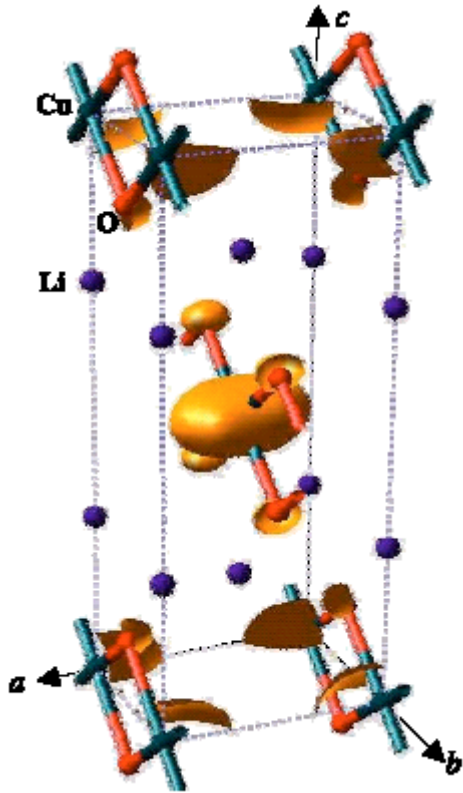


$\text{Li}_2\text{CuO}_2$  crossover between AFM and FM ordering



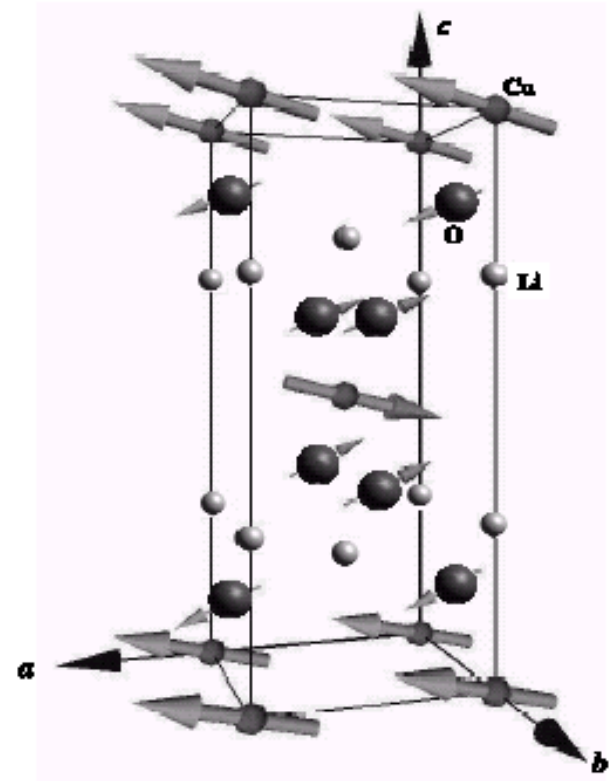
# Oxygen moment in $\text{Li}_2\text{CuO}_2$

AF canted model



Moments are canted in the  $a$ - $c$  plane toward the  $c$ -axis. Canting of the Cu moments is almost fully compensated by counter-canting of the O moments

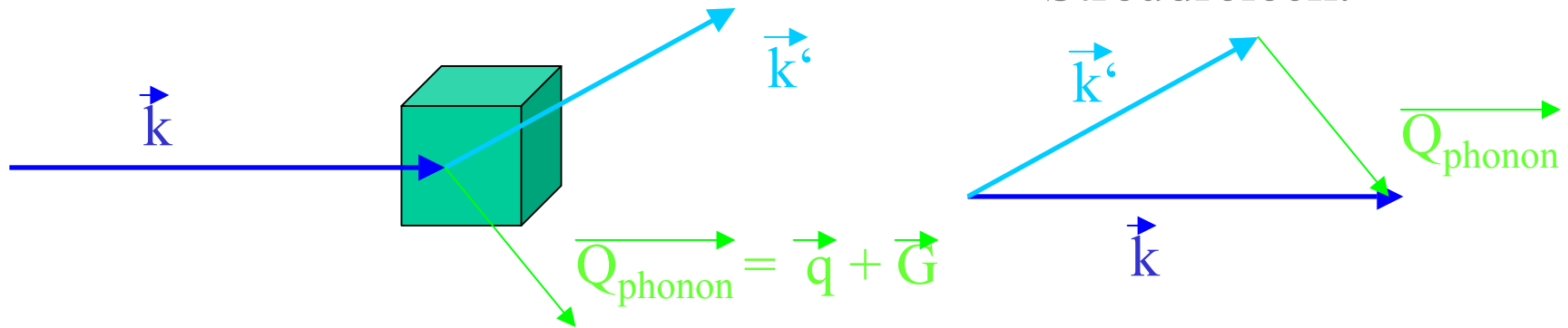
3-D spin density Patterson



Cu moment:  $0.91 \mu_B$   
O moment:  $0.12 \mu_B$

# Inelastische Streuung

Neutrons show what atoms do



Stredreieck:

**Impuls :  $\vec{Q} = \vec{k} - \vec{k}'$**

$$E = \frac{\hbar^2 \cdot k^2}{2 \cdot m}$$

**Energie :  $E = E' + E_{\text{Phonon}}$**

$$E_{\text{Phonon}} = \hbar \cdot \omega = \frac{\hbar^2}{2 \cdot m} \cdot (k^2 - k'^2)$$

**Spin : Gesamt - Spin vorher = nachher**

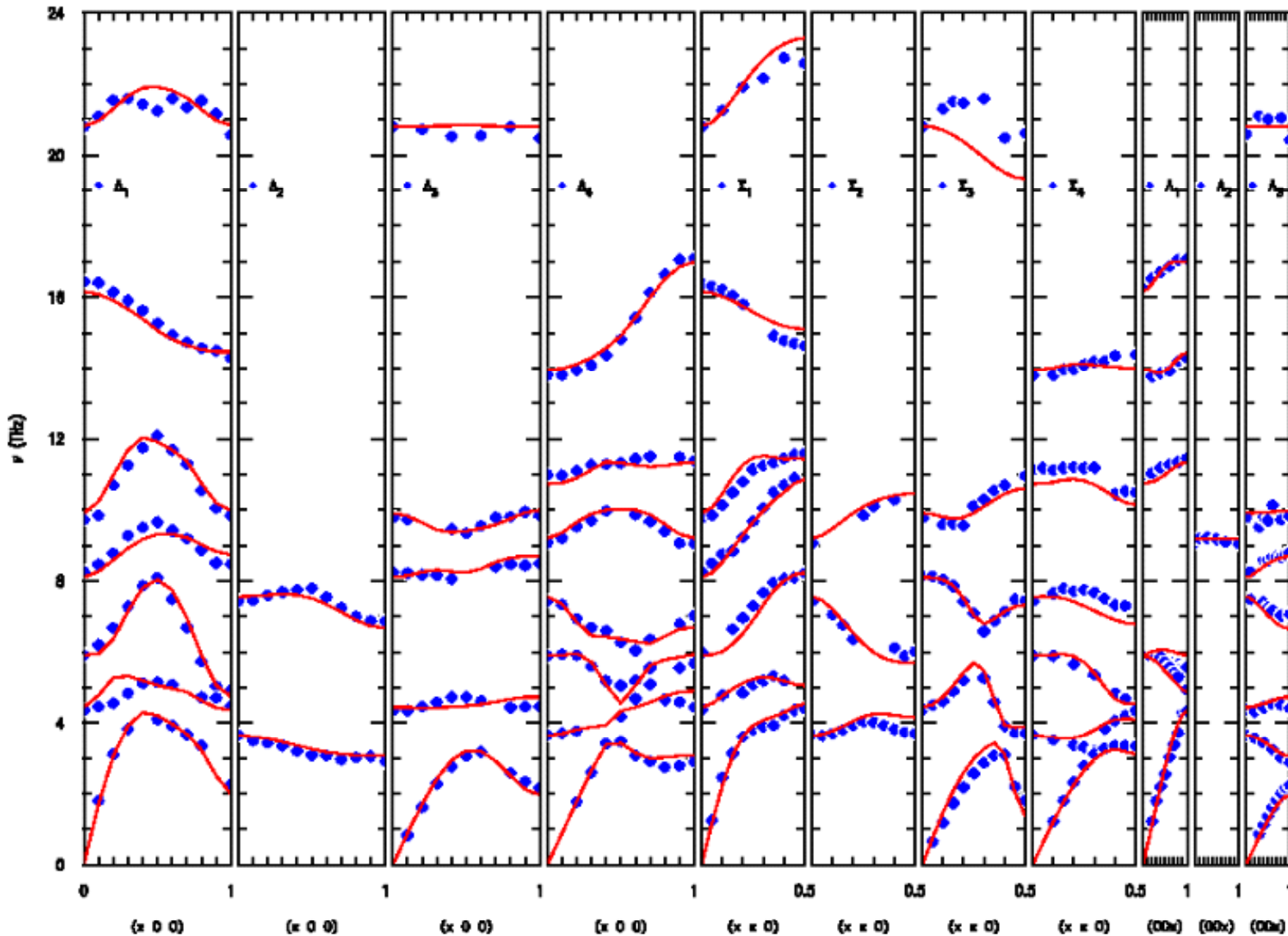
**Beispiel : Thermische Neutronen**

**Energie : 5THz = 20.68meV = 240K = 167cm<sup>-1</sup>**

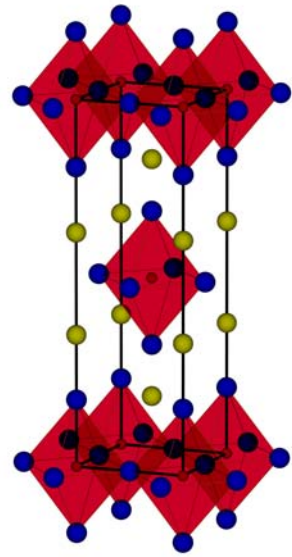
**Wellenlänge : 2.0Å Wellenvektor : 3.14 Å<sup>-1</sup>**

**Geschwindigkeit : 1988 m sec<sup>-1</sup>**

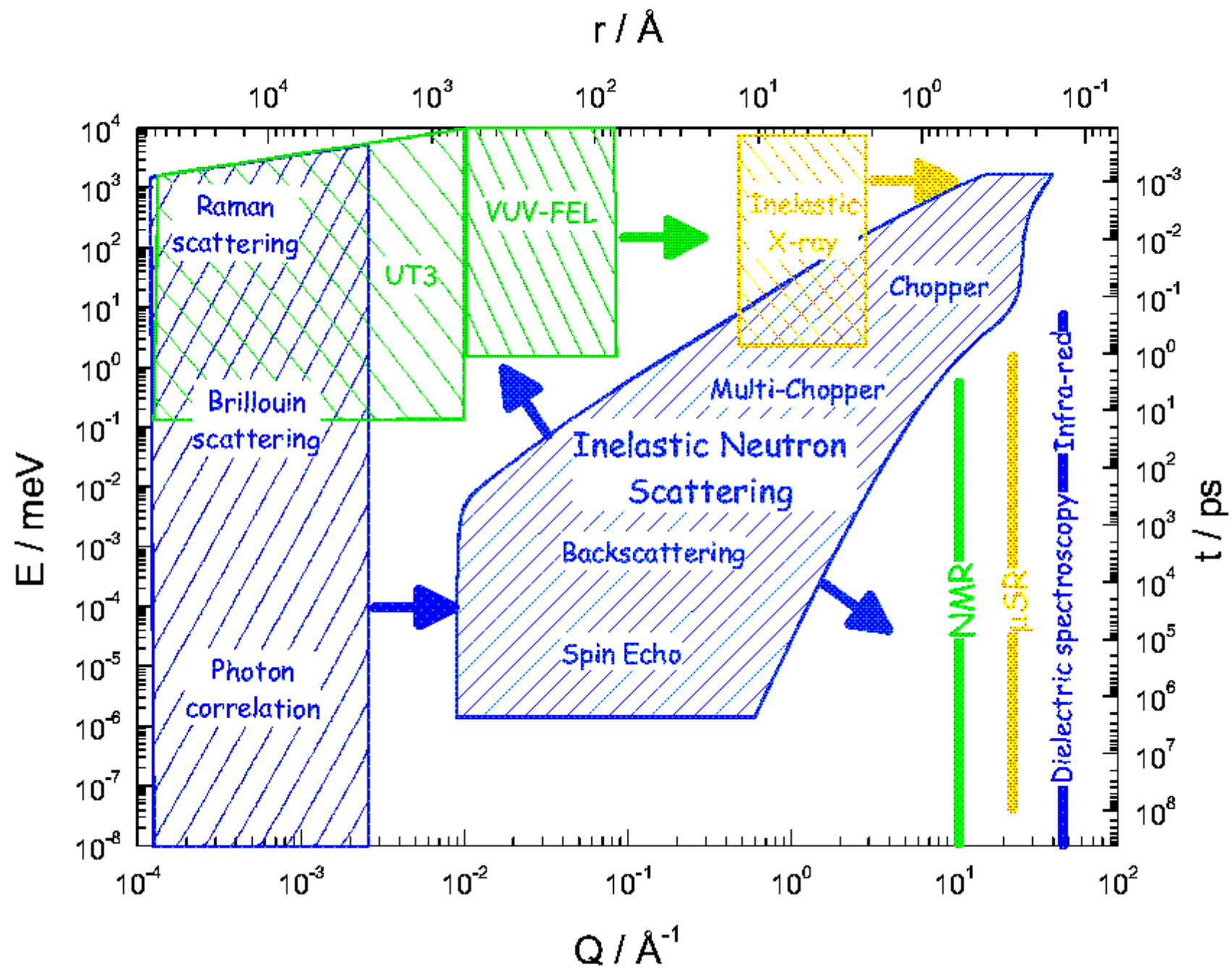
# Gitterdynamik in $\text{Sr}_2\text{RuO}_4$



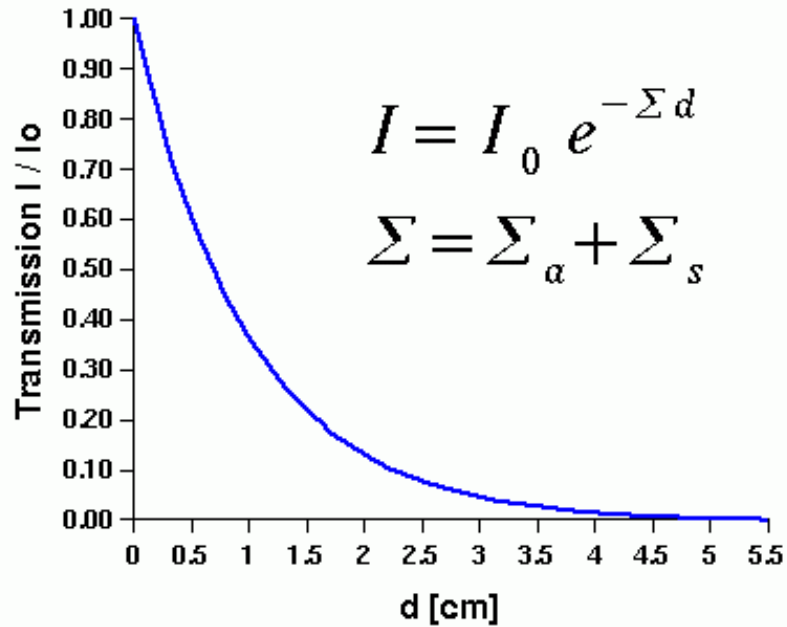
**I4/mmm**  
**7atoms**  
 $\Rightarrow$   
**21 branches**



Braden et al.,  
 PRB 2007.



# Exponential Attenuation Law



$$I = I_0 e^{-\Sigma d}$$

$$\Sigma = \Sigma_a + \Sigma_s$$

Macroscopic Cross Section  $\Sigma$

$$\Sigma = N \sigma \quad [cm^{-1}]$$

$$N = \frac{\rho}{A} N_A \quad [cm^{-3}]$$

$N$  := number density [ $cm^{-3}$ ]

$\rho$  := material density [ $g\ cm^{-3}$ ]

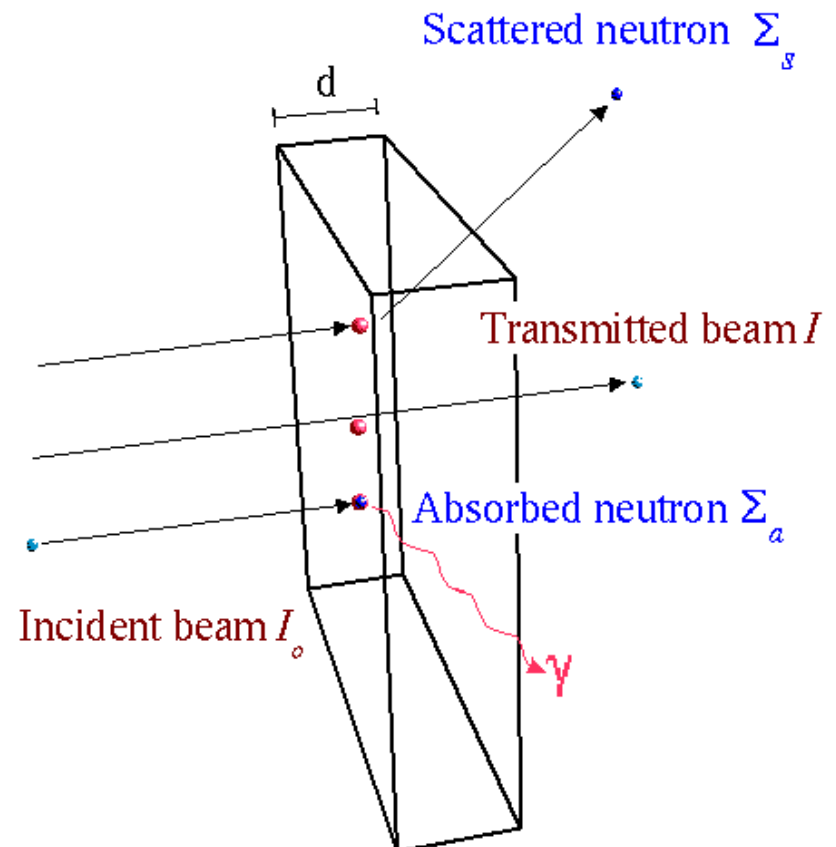
$A$  := atomic weight [ $g\ mol^{-1}$ ]

$N_A$  := Avogadro number  $6.022 \cdot 10^{23}$  [ $mol^{-1}$ ]

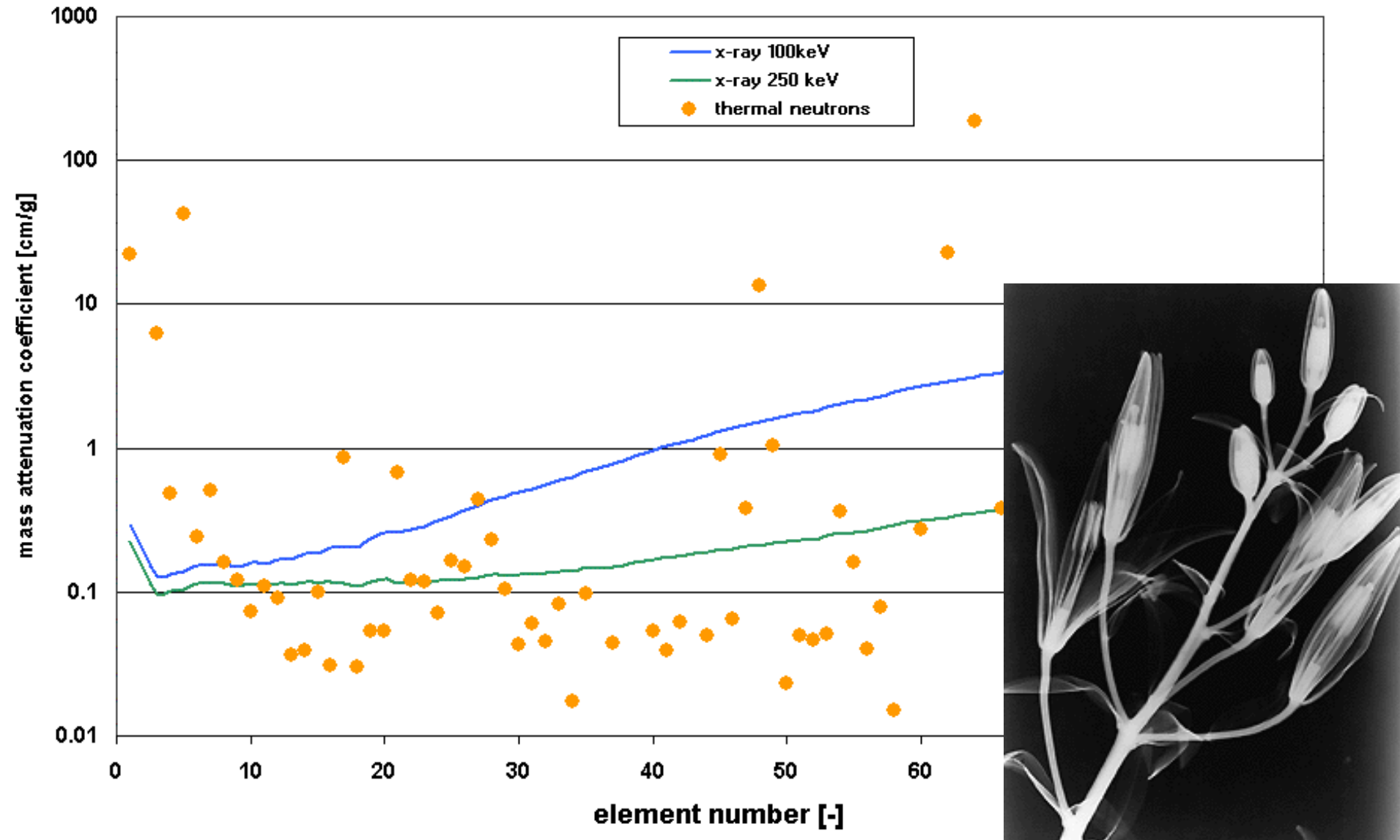
# Wechselwirkung mit Materie

## Narrow Beam Attenuation

$$I = I_0 e^{-\Sigma d}$$

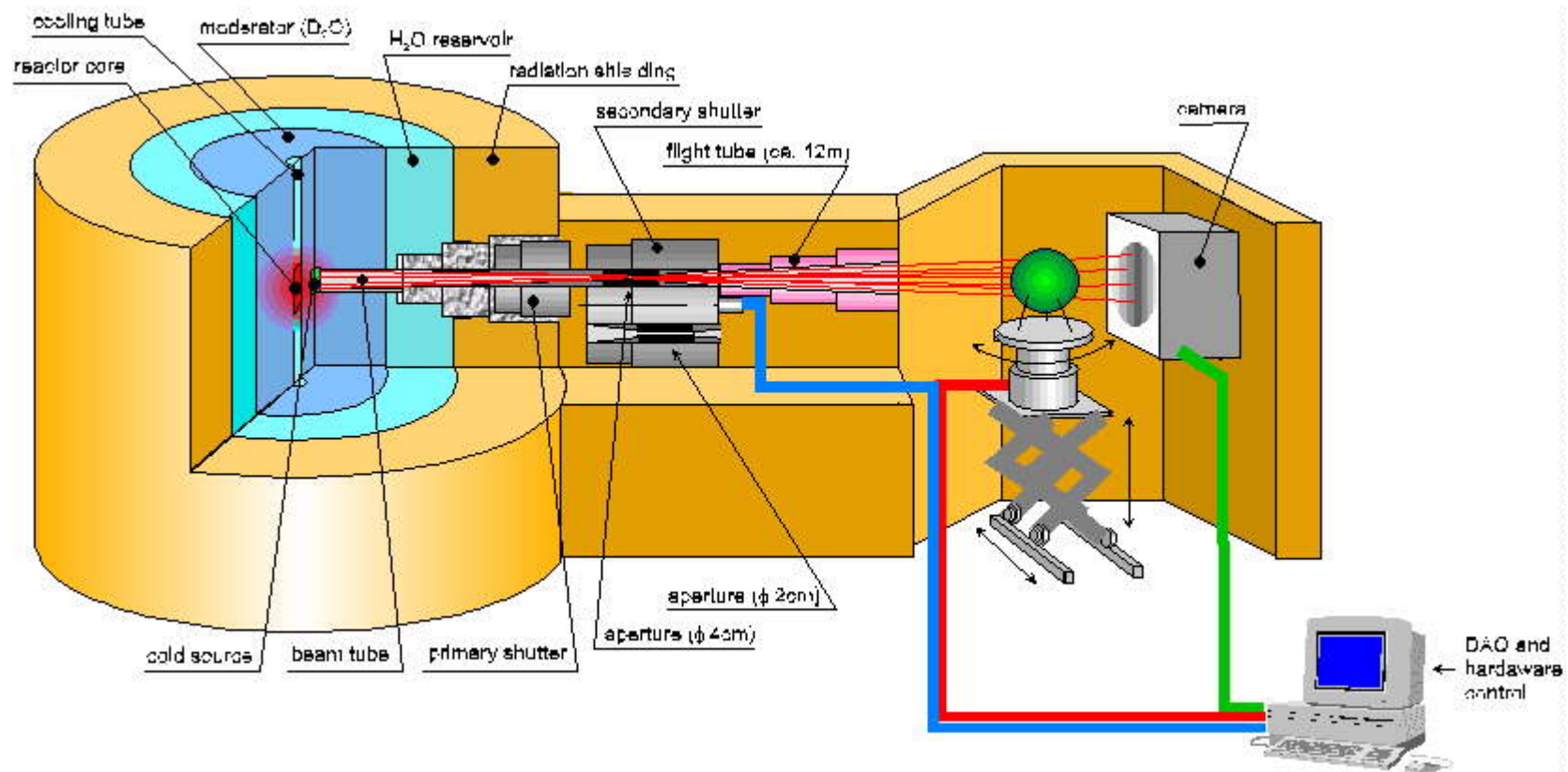


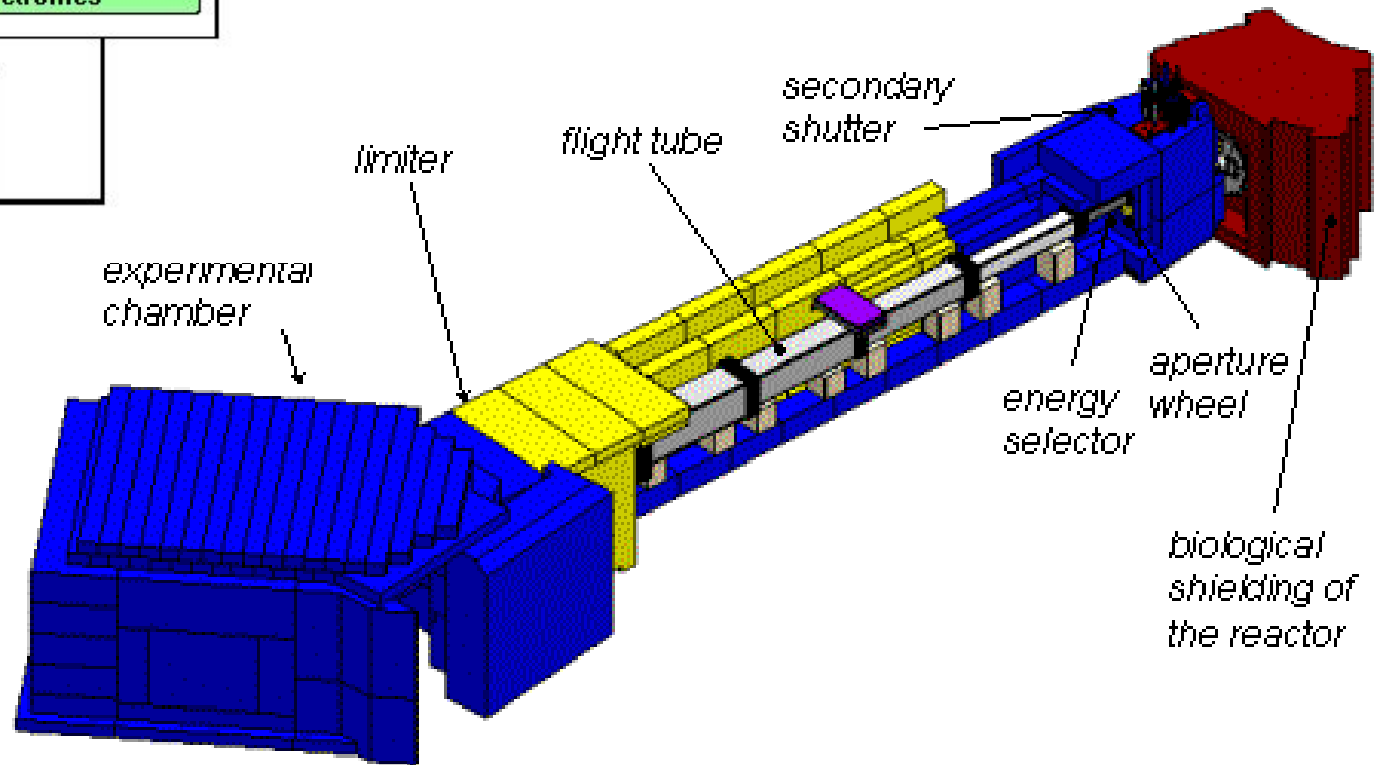
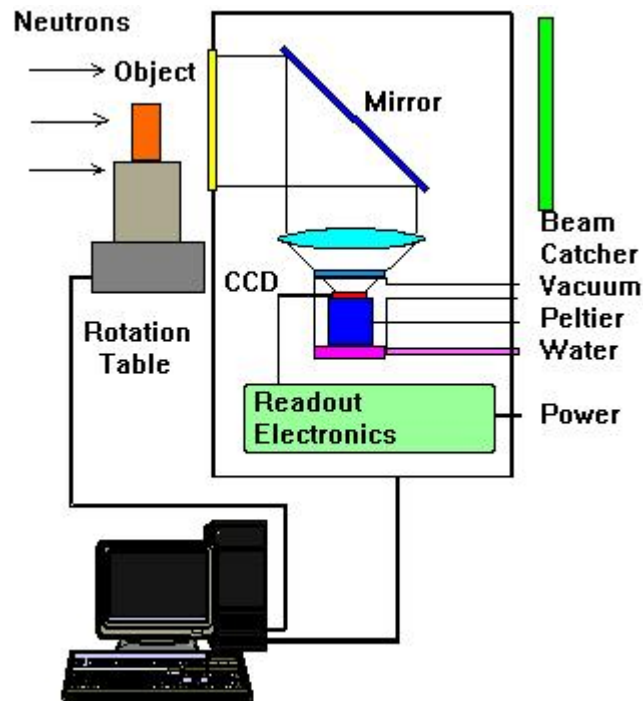
# Schwächungskoeffizienten $\Sigma$



# Antares

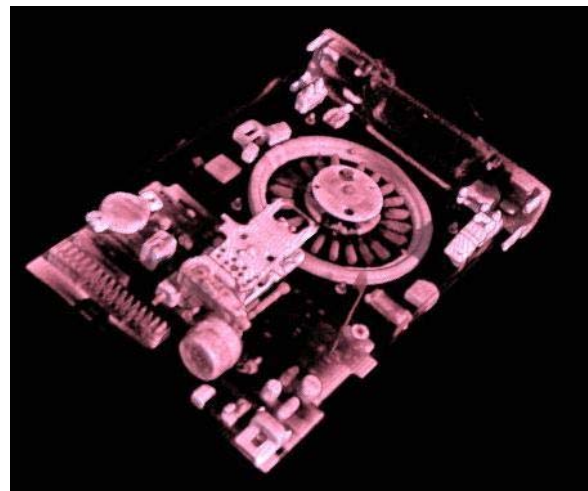
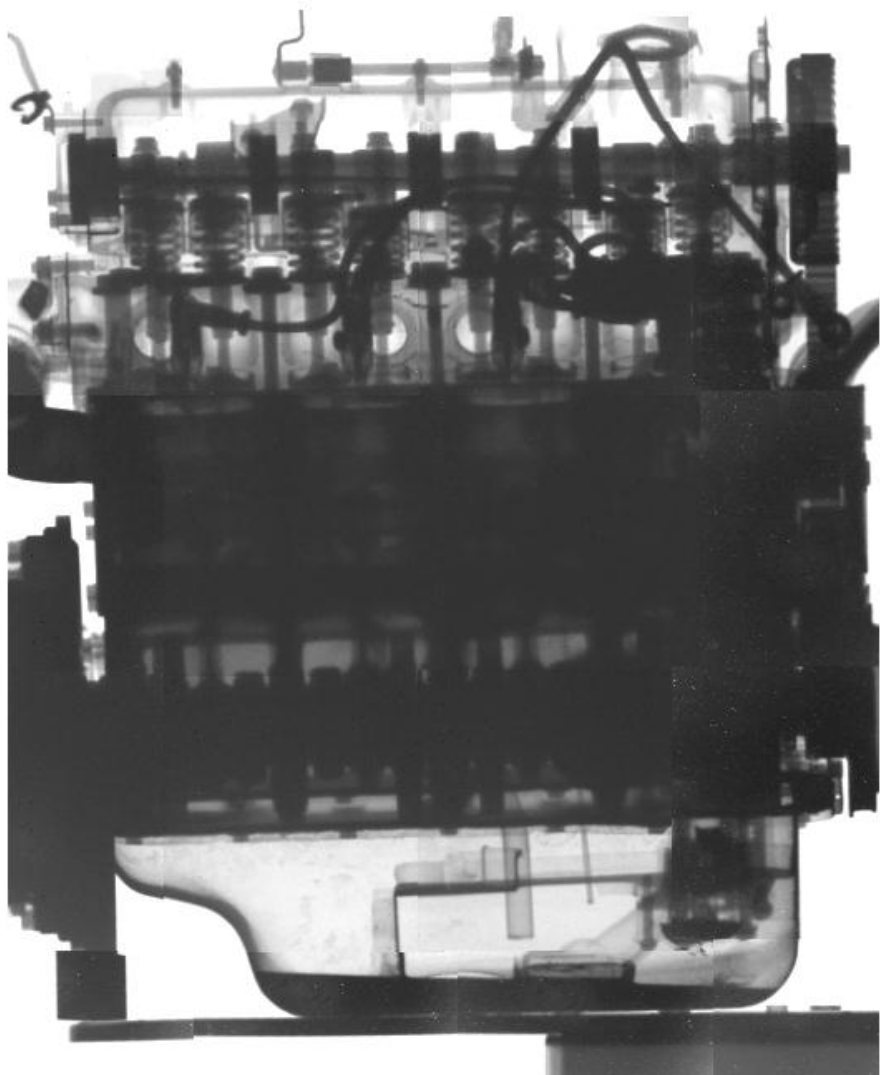
## Advanced Neutron Tomography And Radiography Experimental System



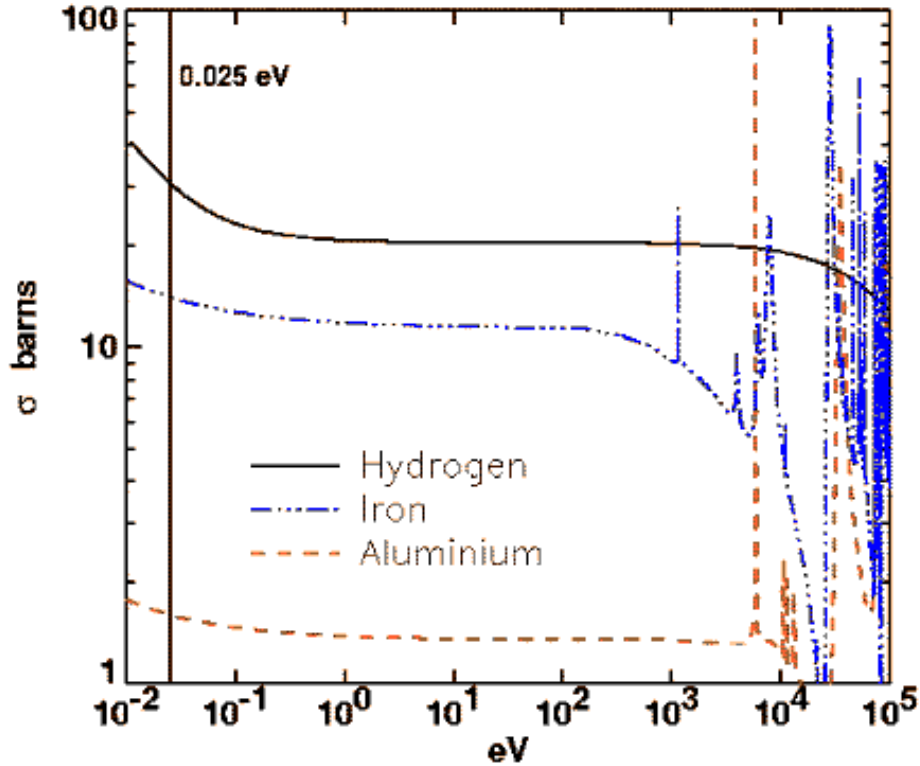




23.10.2007



# Neutron Cross Sections (Energy – Dependence)



## *Thermal Neutron Cross Sections at 0.025eV*

<i>Material</i>	$\rho$	$\sigma$	$\Sigma$
<i>Water</i>	1	103	3.45
<i>Iron</i>	7.9	14	1.18
<i>Aluminium</i>	2.7	1.7	0.1

# Was sind die Charakteristika von Neutronen ?

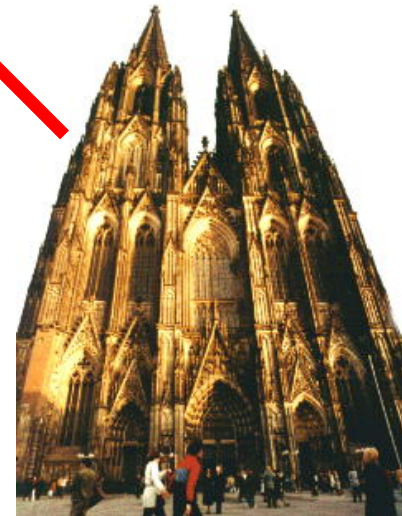
- 1) Masse vergleichbar zu Kern-Massen → elastische und inelastische Streuung
- 2) Wellenlänge 1-5Å im Bereich der Gitterabstände
- 3) nukleare und lokale Wechselwirkung : keine Z-Abhängigkeit  
→  $V(r) = b \cdot \delta(r)$
- 4) Energie im Bereich 1-100meV vergleichbar zu der von Phononen  
(x-ray : Cu-K hat  $8 \cdot 10^6 \text{meV}$  → Energieauflösung besser als  $10^{-7}$  )
- 5) das Neutron hat einen Spin  $\frac{1}{2}$

- Kontrast : leichte schwere Atome
- Magnetismus
- Experimente bei grossem Streuvektor möglich
- Messungen in der ganzen Brillouin-Zone (inelastisch)

## Beispiel : Thermische Neutronen

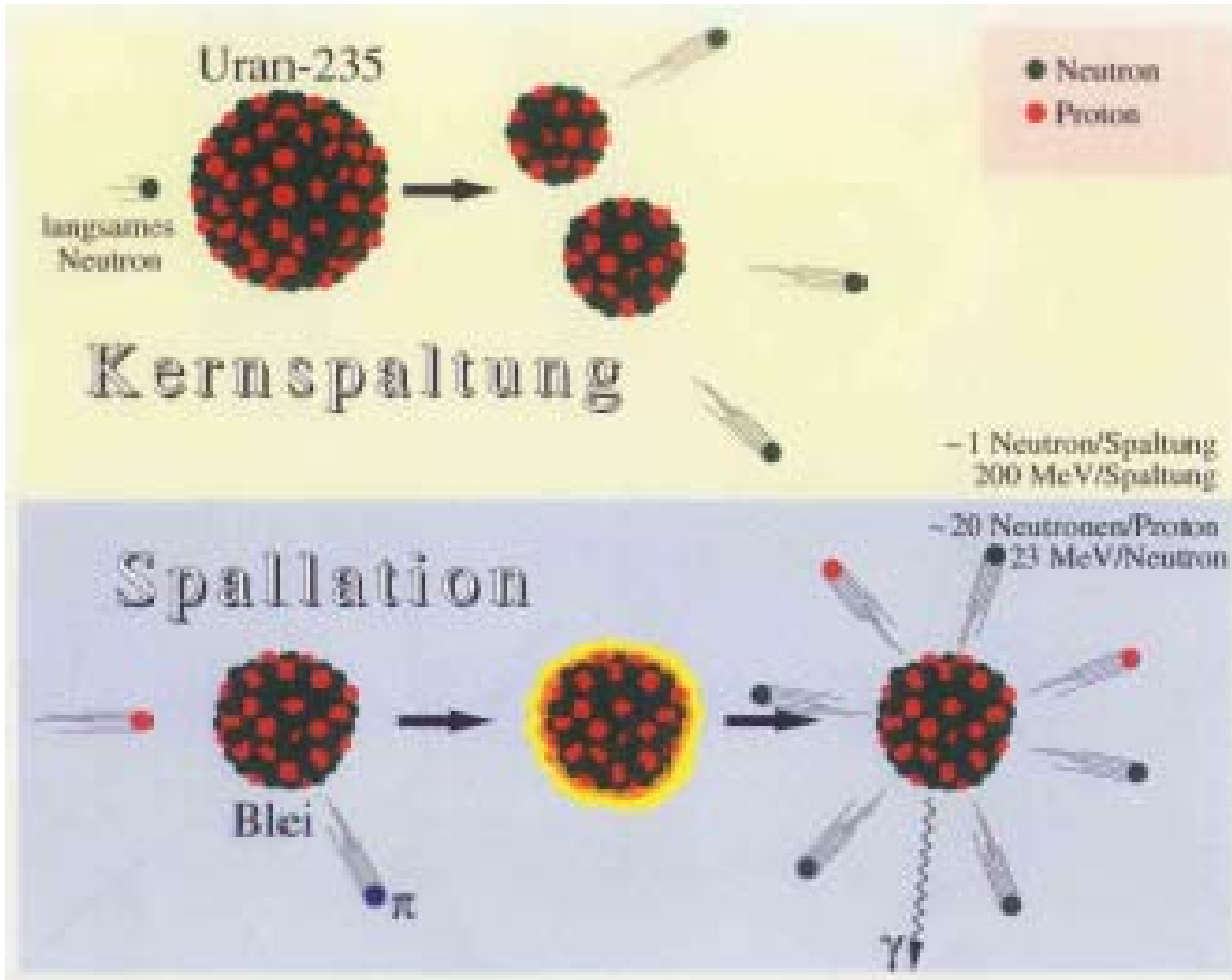
Energie : 5THz = 20.68meV = 240K = 167cm<sup>-1</sup>  
Wellenlänge : 2.0Å Wellenvektor : 3.14 Å<sup>-1</sup>  
Geschwindigkeit : 1988 m sec<sup>-1</sup>

**ABER : Quellen sind teuer und schlecht !!!**



**Liste ist nicht vollständig !!!**

# Neutronen-Quellen







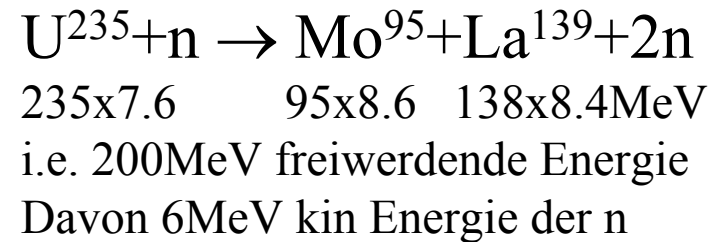
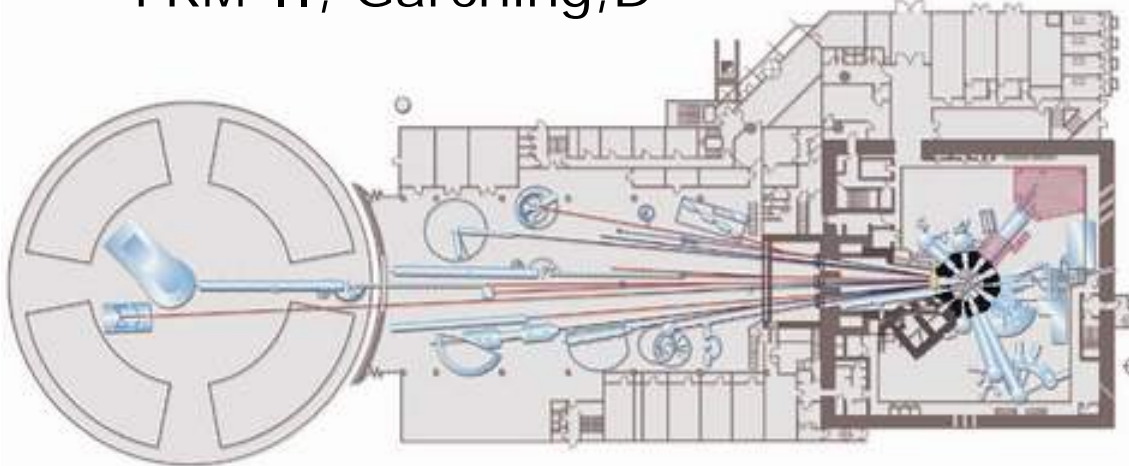
# Reaktor-Neutronenquellen



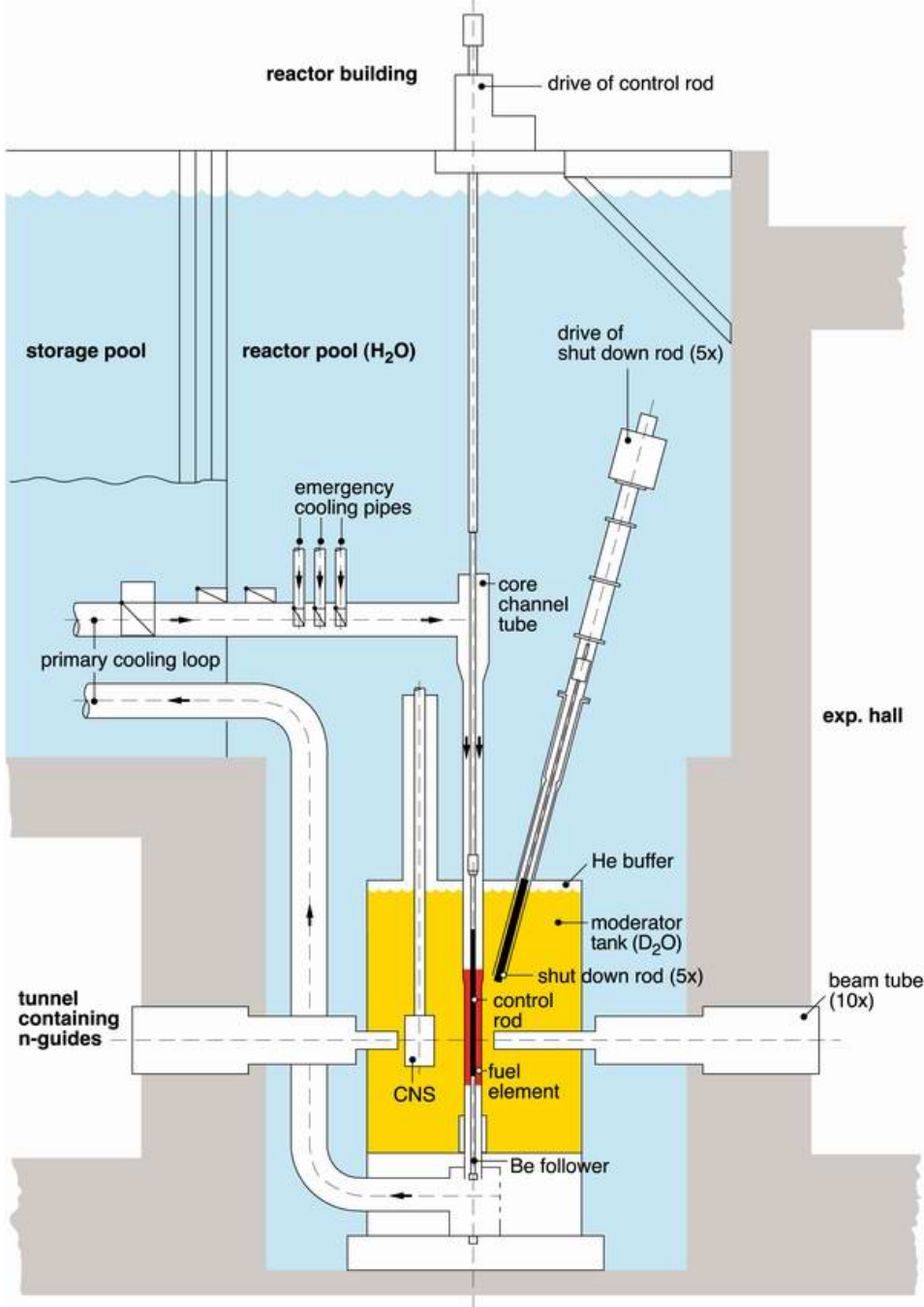
FRM-II, Garching, D



ILL, Grenoble, F







**FRM-II** - thermal power: 20MW

ILL: 56MW

unperturbed thermal

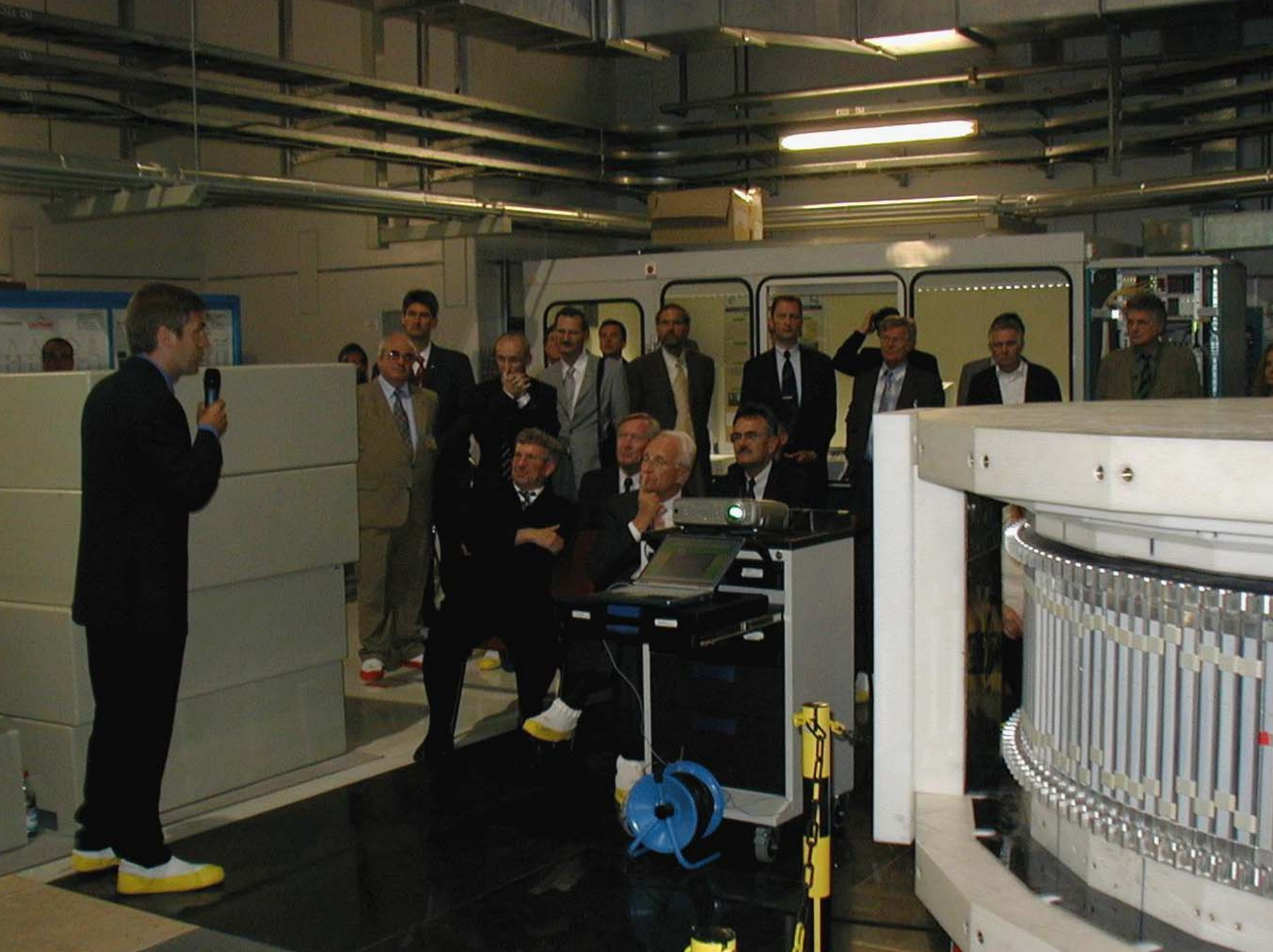
neutron flux:  $8 \times 10^{14} \text{cm}^{-2}\text{s}^{-1}$

core cooled with light water H<sub>2</sub>O

moderator tank: heavy water D<sub>2</sub>O

diameter of approx. 2,5m

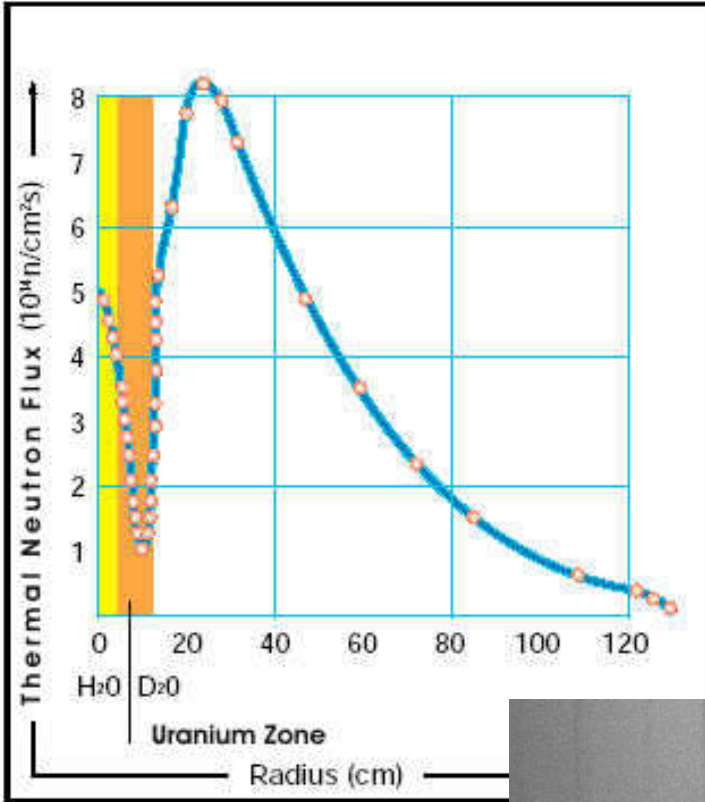




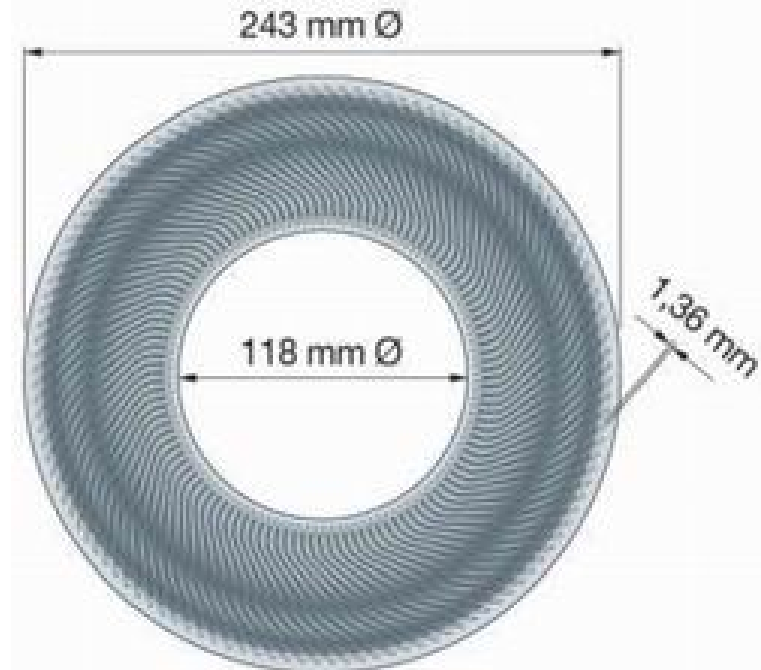
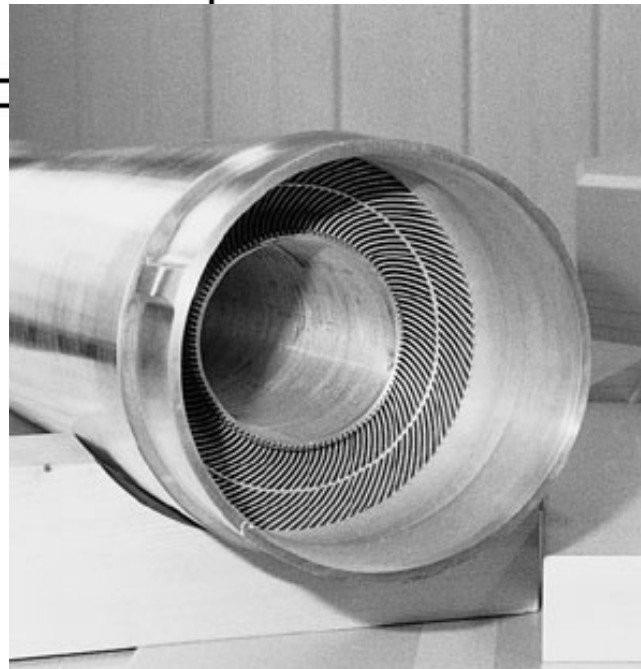


***FRM-II's reactor core:***  
*a single, cylindrical fuel element 700mm height.*

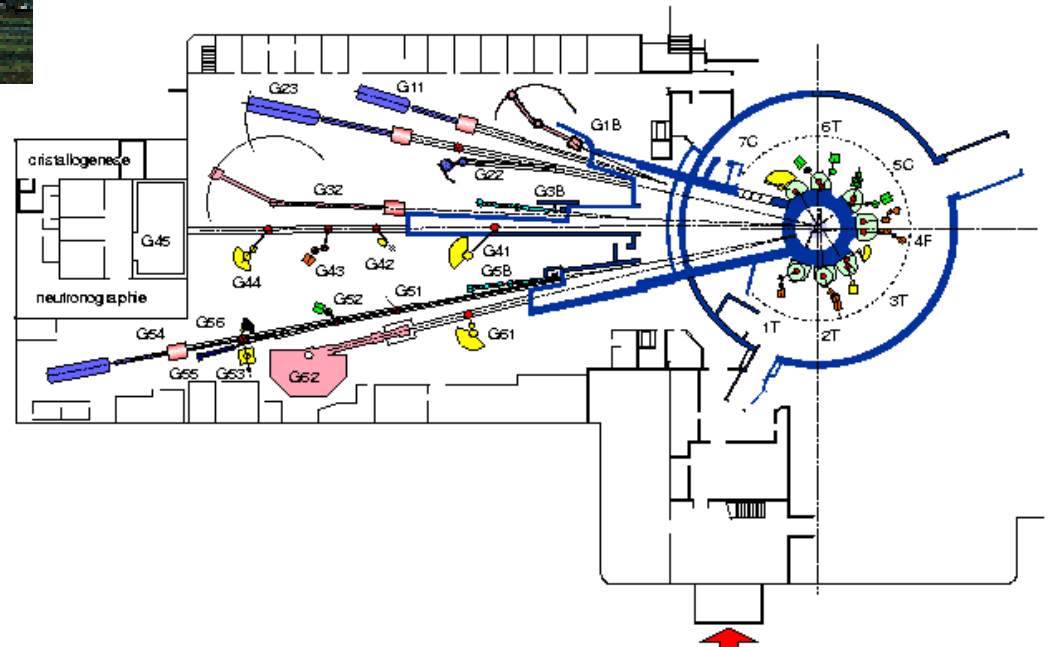
*113 involuted, curved fuel plates  
three layers: fuel  $U_3Si_2$  +  
aluminium powder  
in an aluminium matrix.  
5 elements will be required per year.*



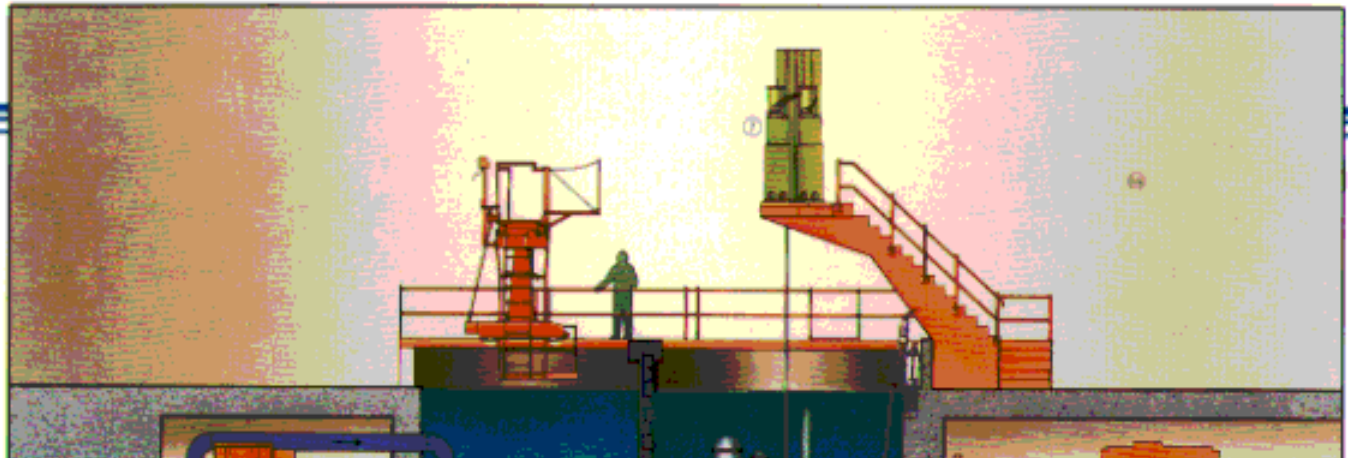
***Thermal neutron flux density as function of the radius (FRM-II)***



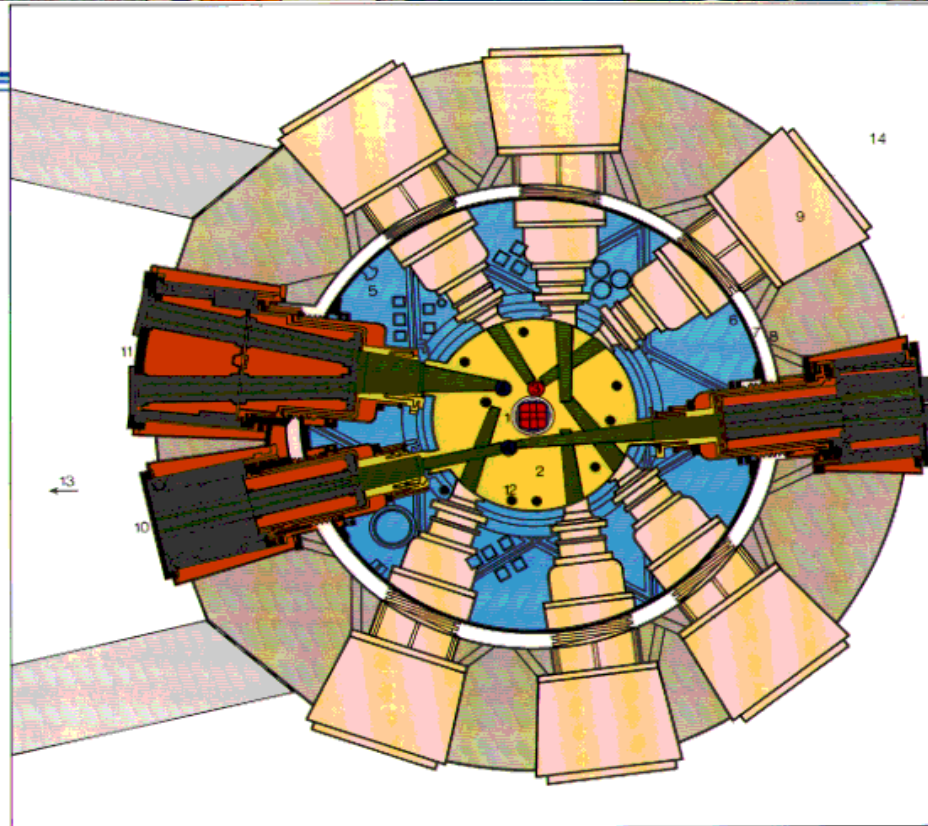
# Laboratoire Leon Brillouin (Saclay, Paris)



1. Cœur
2. Réflecteur d'eau lourde
3. Piscine et canal de transfert
4. Circuit primaire
5. Circuit secondaire
6. Circuit d'eau lourde
7. Mécanismes de commande des barres de contrôle
8. Echangeur
9. Pompe
10. Bâche de vidange de la piscine
11. Source froide
12. Source chaude
13. Canal simple
14. Hall des guides à neutrons
15. Guide à neutrons
16. Hall-pile
17. Protection
18. Monochromateur
19. Protection
20. Spectromètre



1. Cœur
2. Réflecteur d'eau lourde
3. Source chaude
4. Source froide
5. Piscine
6. Cuvelage de la piscine
7. Vide annulaire
8. Doubliante piscine
9. Canal standard
10. Canal simple
11. Canal double
12. Canal vertical
13. Hall des guides à neutrons
14. Hall des expérimentateurs



**Facility:** LLB

**Type:** Reactor

**Flux:**  $3.0 \times 10^{14}$  n/cm<sup>2</sup>/sec

**Operational days/year:** 250 no longer now 114 till 180

**Total number of instruments:** 28

**Number of instruments available to users:** 24

**Type of instruments available to external users:**

6 powder/liquid diffractometers

2 single crystal diffractometers

1 Strain diffractometer

1 Texture diffractometer

3 SANS

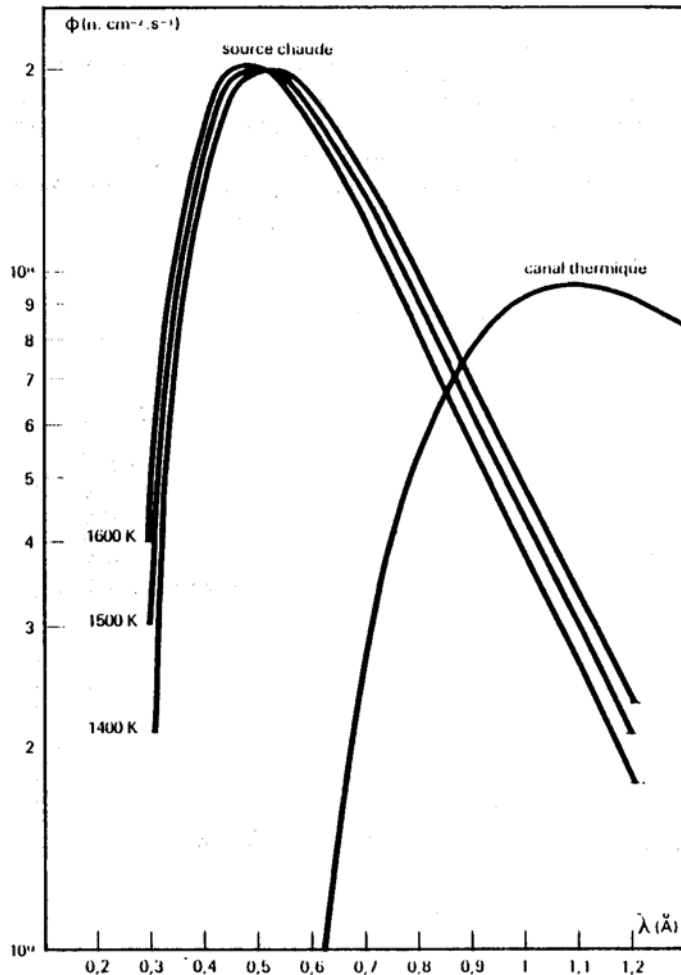
3 reflectometers

5 3-axis spectrometers

1 TOF (MET)

1 spin echo

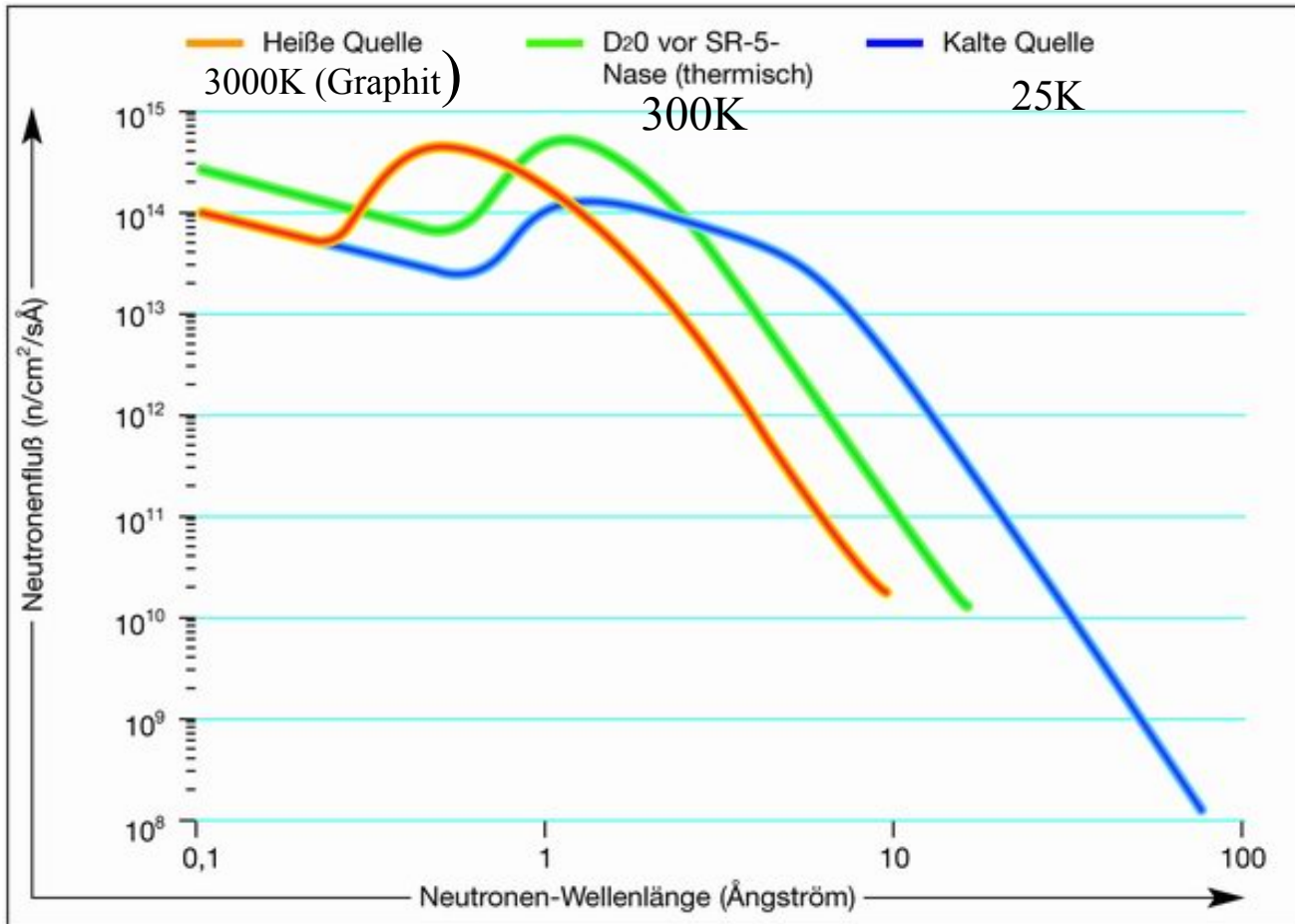
1 polarised neutron instrument



**Figure 3 : Source chaude**  
**Distribution du flux en fonction de la longueur d'onde pour trois températures de sources chaudes.**

# Moderatoren ( $D_2O$ )

MeV  $\rightarrow$  meV durch elastische Streuung an leichten Elementen verlieren die Neutronen Energie und es ergibt sich im Idealfall eine Maxwell Verteilung der n-Geschwindigkeit entsprechend der Moderator Temperatur



Betrachtung eines elastischen Stoßes eines Neutrons mit einem Kern der Massenzahl  $A$  ergibt:

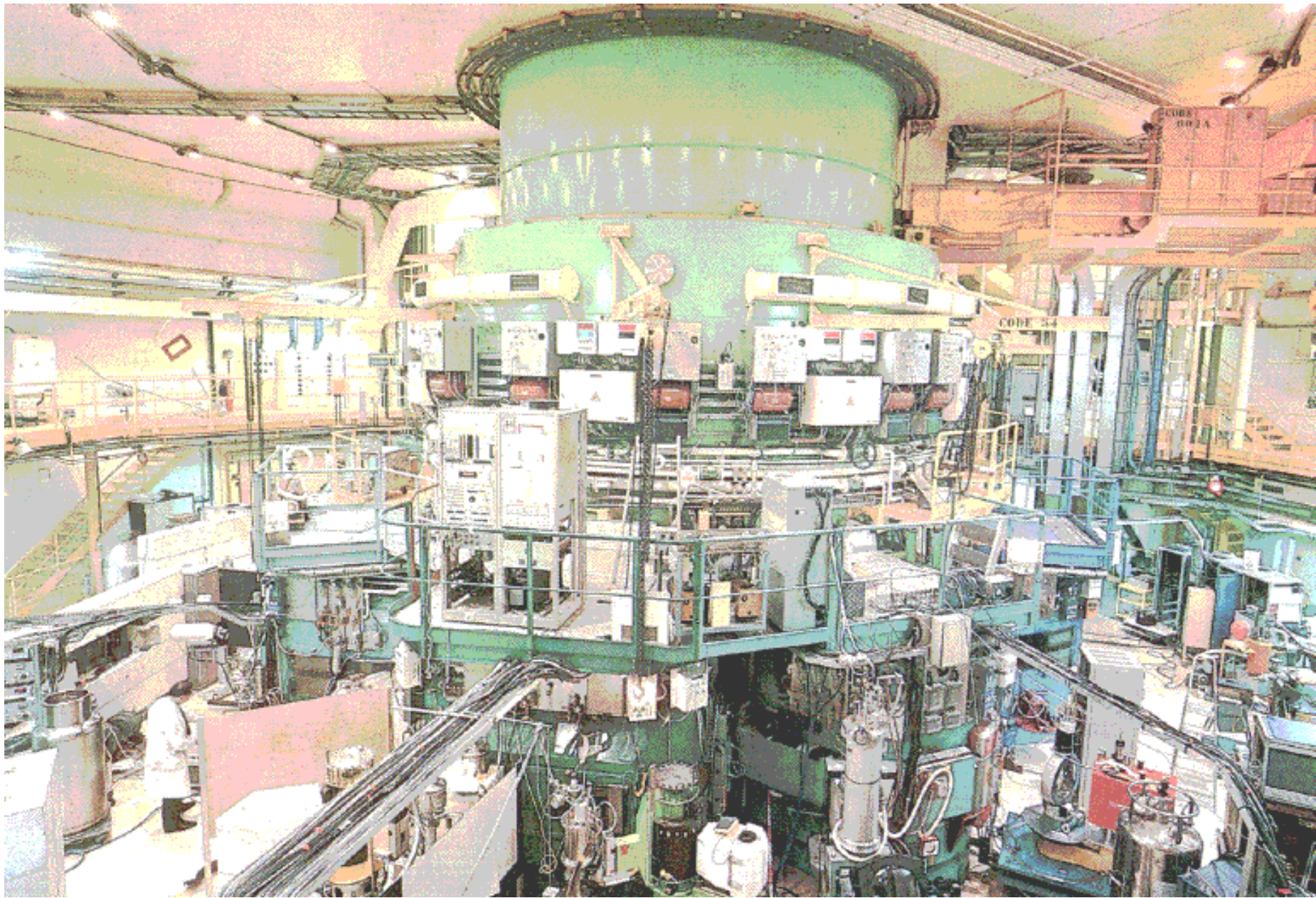
Am besten wäre Wasserstoff Problem

- Absorption -
- daher Deuterium !

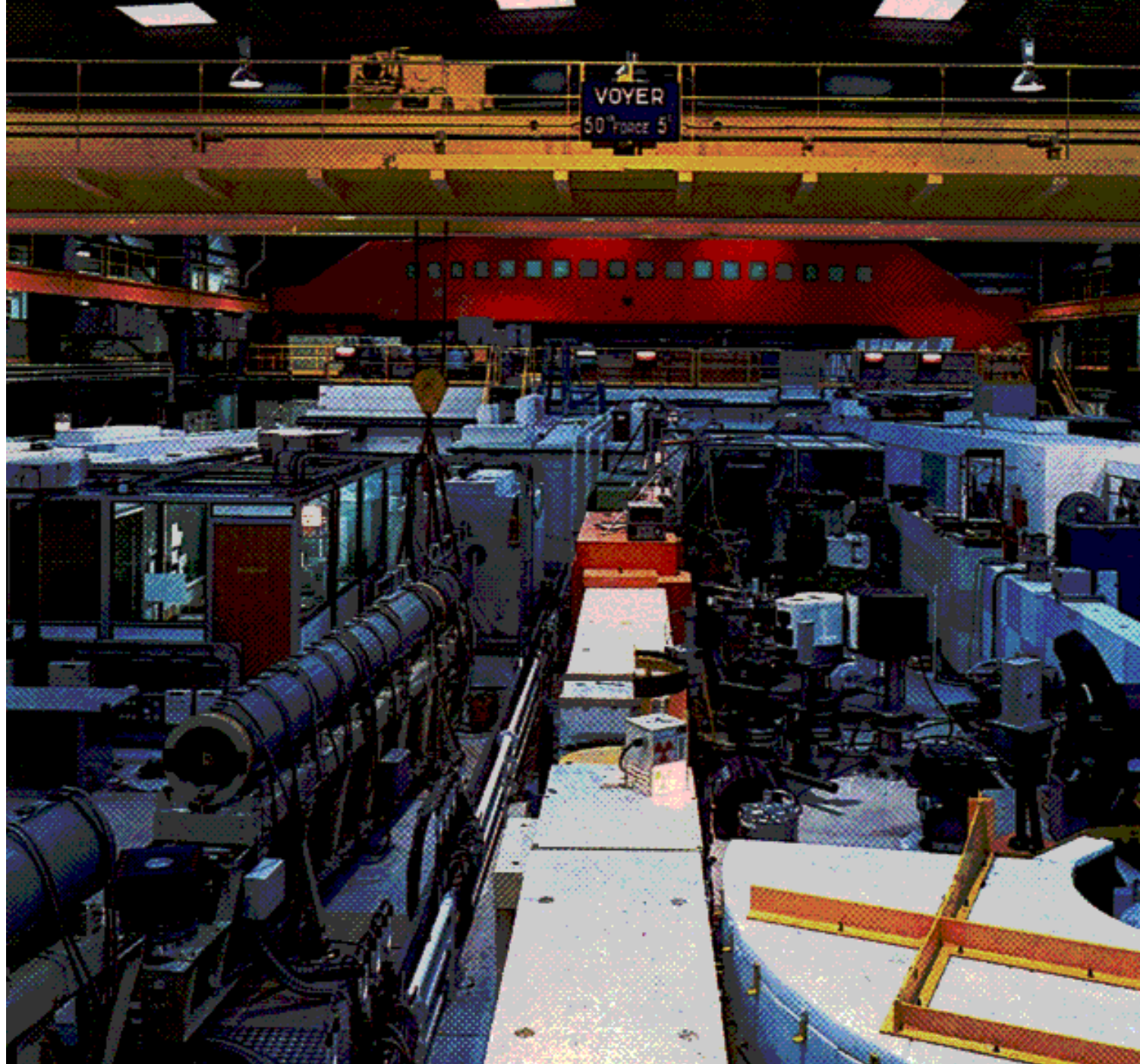
Berechnung für FRM-II, Garching



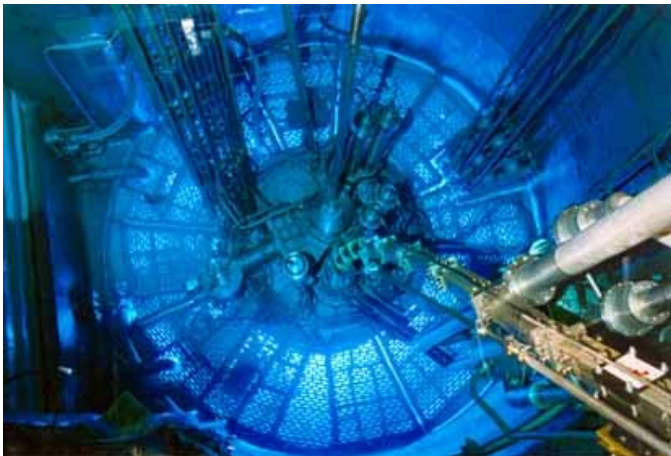








# Institut Laue Langevin (Grenoble)



**Facility:** ILL

**Type:** 58MW High Flux Reactor

**Flux:**  $1.5 \times 10^{15}$  n/cm<sup>2</sup>/sec

**Operational days/year:** 225

**Total number of instruments:**

Approximately 43 including test

**Number of instruments available to users:** 36

**Type of instruments available to external users:**

5 powder/liquid diffractometers

7 single crystal diffractometers

2 SANS\*

3 reflectometers\*

5 polarised neutron instruments\*

2 Nuclear Physics

6 3-axis spectrometers

2 backscattering spectrometers

3 TOF (MET)

2 spin echo

2 Fundamental Physics

\*some double counting

**NB:** 7 of the above instruments are operated and supported by external groups





# BENSC (Berlin)

**Facility:** BER II, BENSC

**Type:** Swimming pool reactor

**Flux:**  $2 \times 10^{14}$  n/cm<sup>2</sup>/sec

**Operational days/year:** 250

**Total number of instruments:** 24

**Number of instruments available to**

**external users:** >17

**Type of instruments available to external users:**

2 powder/liquid diffractometers

3 3-axis spectrometers

4 single crystal diffractometers

1 quasielastic spectrometer

1 membrane diffraction

2 TOF (MET)

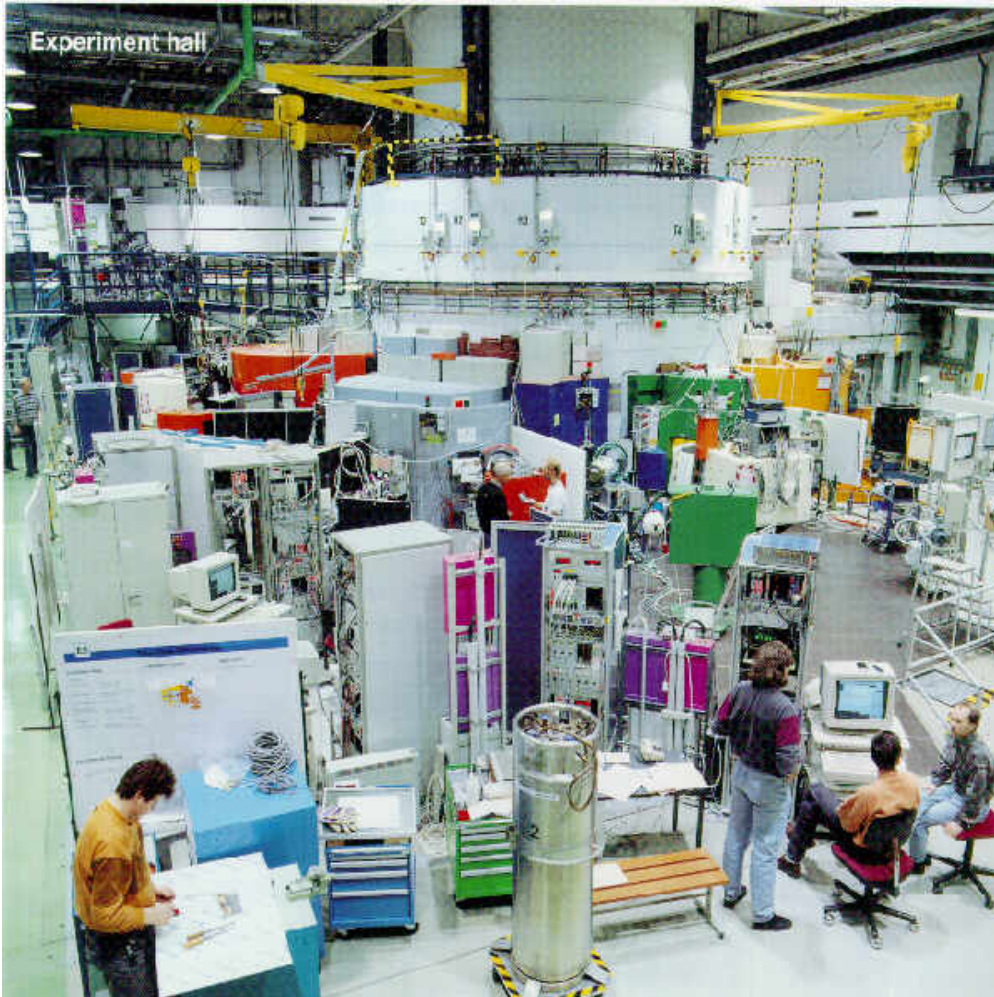
2 SANS

1 spin echo      1 reflectometer

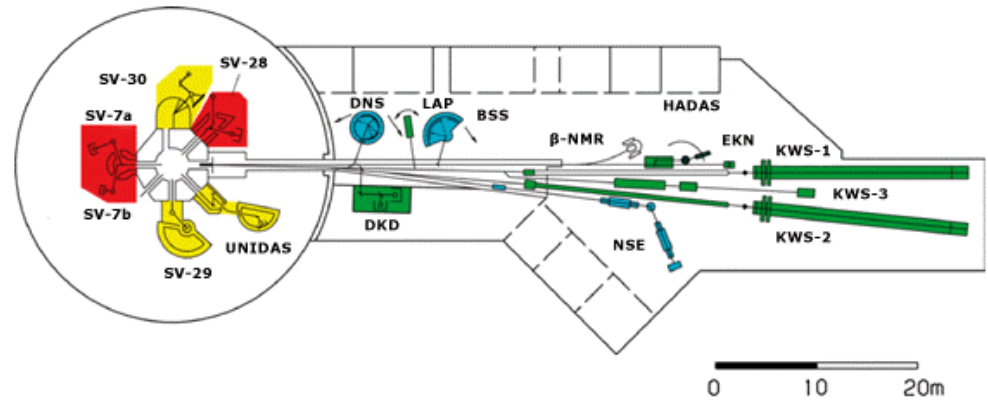
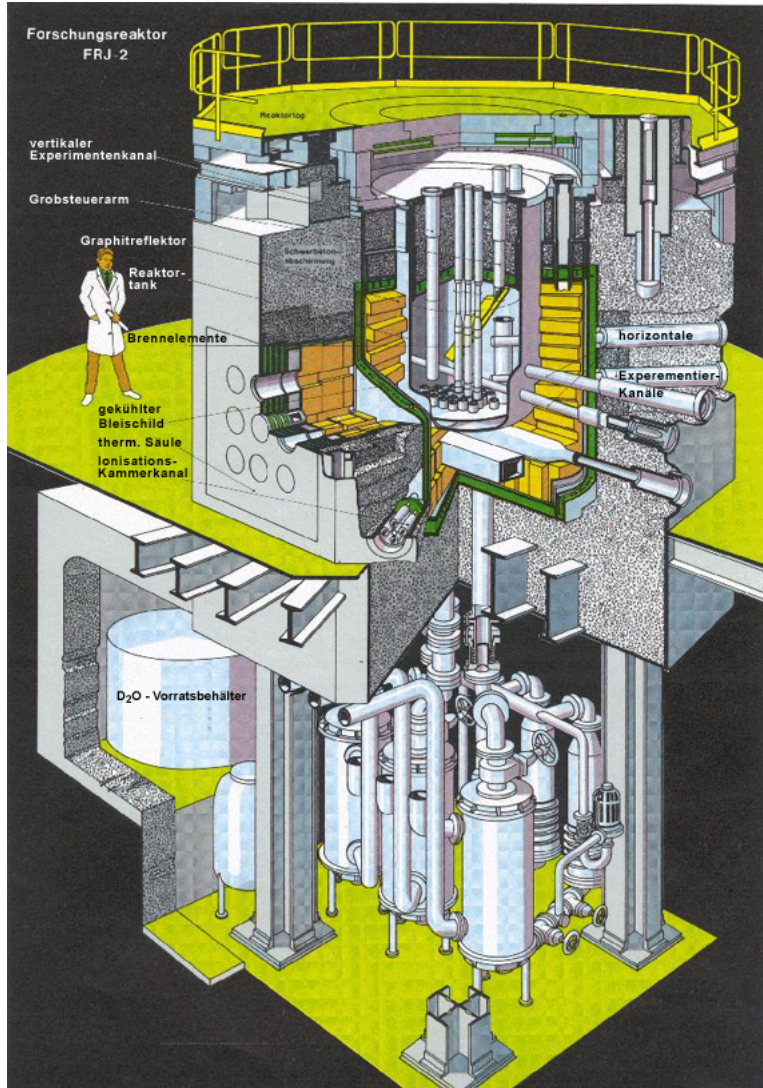
1 neutron interferometer      1  $\beta$ -NMR

1 cold source

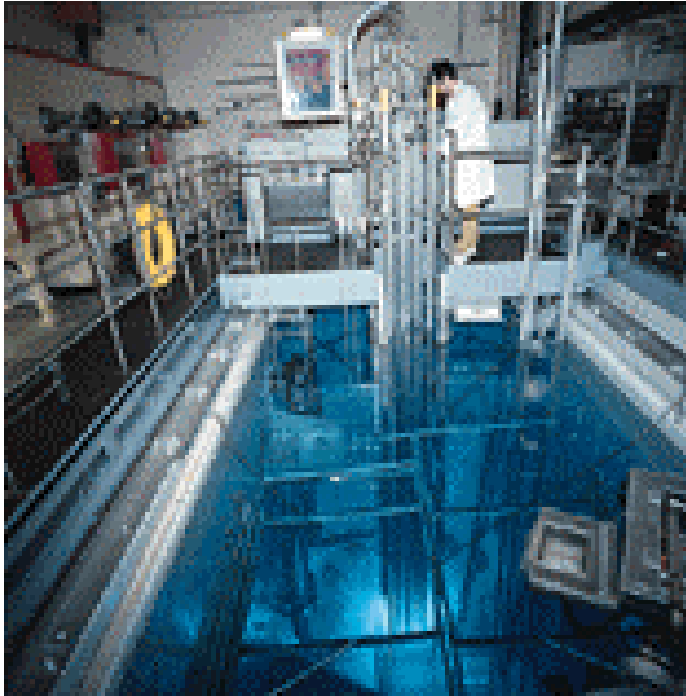
**NB:** For many instruments options include polarisation, high fields, high pressures and low temperatures



# FRJ-2 Jülich



# FRG-2 Geesthacht



**Facility:** FRG-1

**Type:** Swimming Pool Cold Neutron Source

**Flux:**  $8.7 \times 10^{13}$  n/cm<sup>2</sup>/sec

**Operational days/year:** 210

**Total number of instruments:** 10

**Number of instruments available to external users:** 10

**Type of instruments available to external users:**

1 four circle texture diffractometer

2 residual stress diffractometers

2 SANS

2 reflectometers

1 TOF spectrometer for basic research

1 Double crystal diffractometer for high resolution SANS

1 3-dimensional polarisation analysis diffractometer

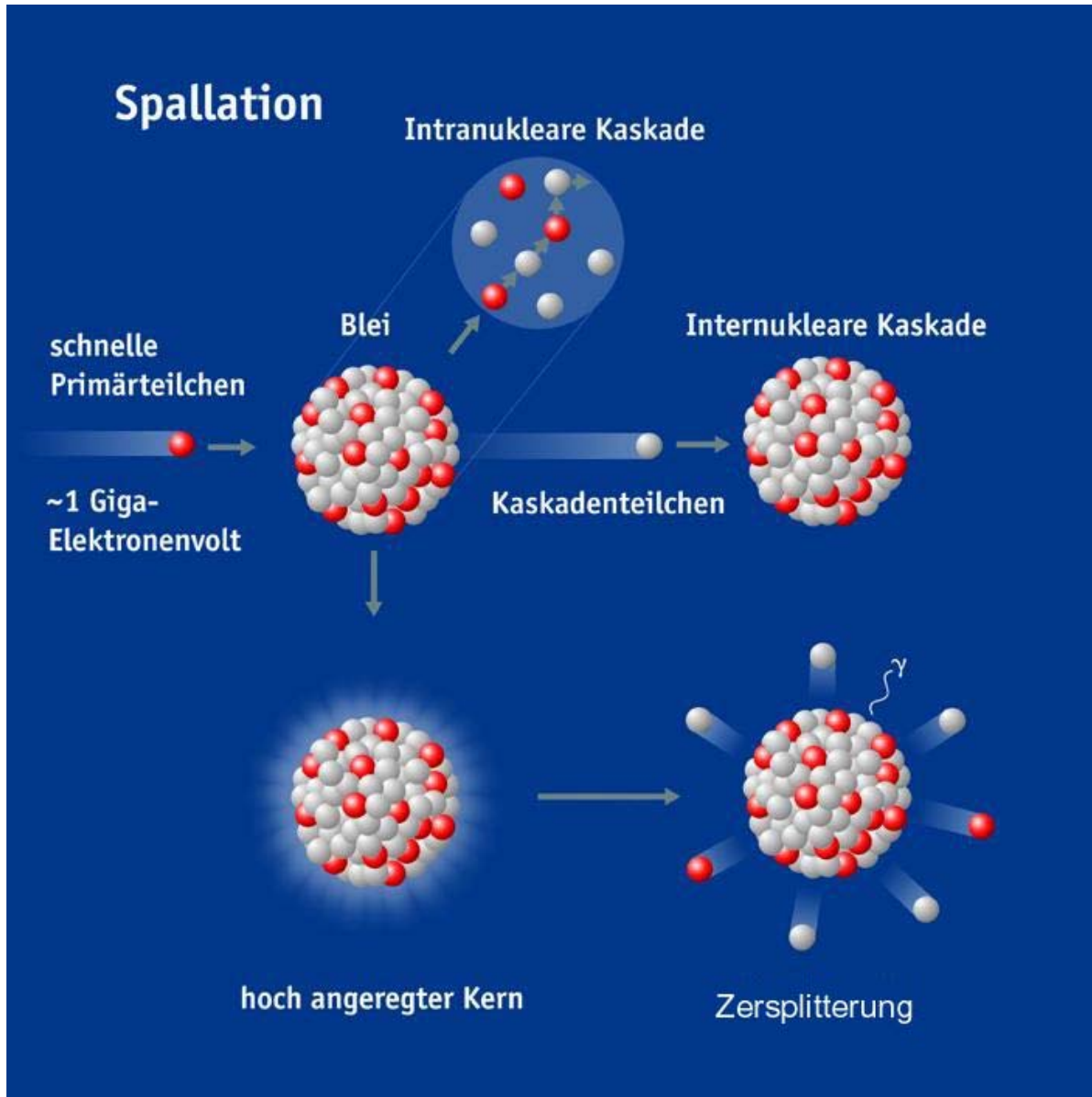
Polarised neutrons available on 5 instruments



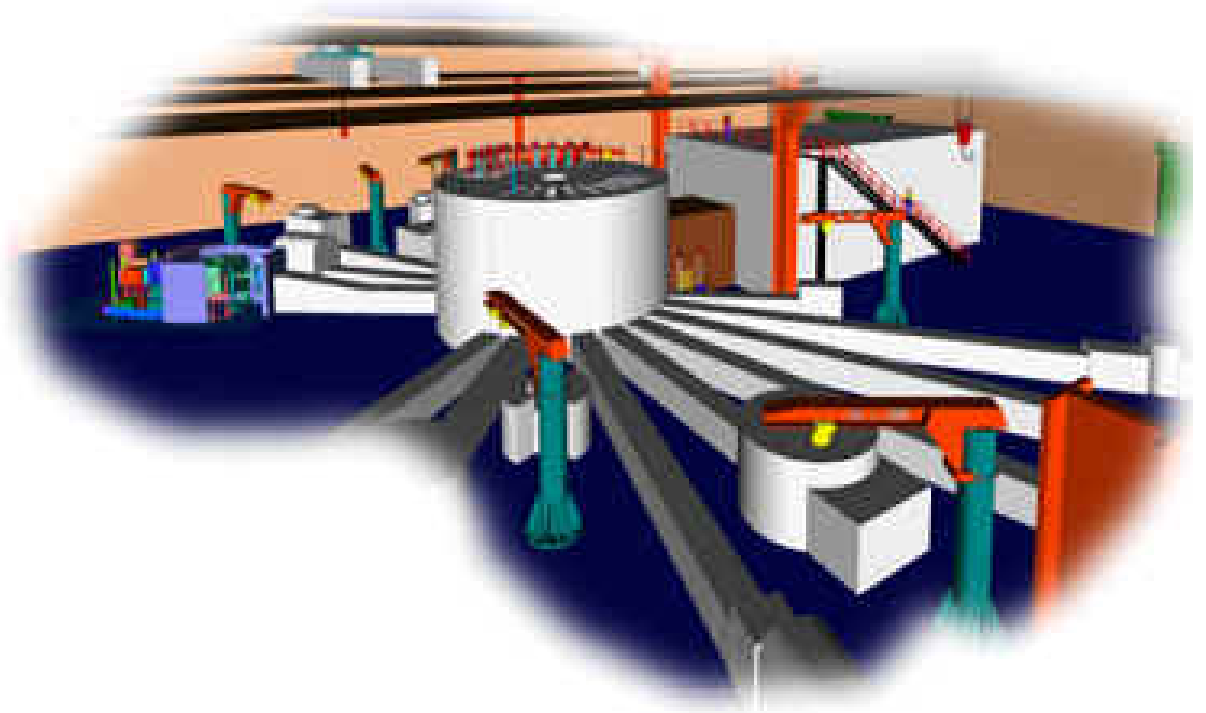
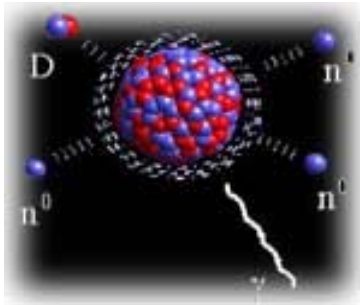
# Spallation

pro freigesetztem Neutron entsteht bei der Spallation weniger Wärme als bei der Kettenreaktion ... daher sind höhere Flüsse möglich: spitzenwerte thermische Neutronen  $2 \times 10^{17} \text{ n/cm}^2 \text{ s}$  (geplant bei der ESS)

Pb: 20n/p +23MeV  
 $^{238}\text{U}$ : 40n/p +50MeV  
(vgl  $^{235}\text{U}$ : 1n/spaltung + 200MeV)



# ISIS, Rutherford (Oxford) Winfried Kockelmann

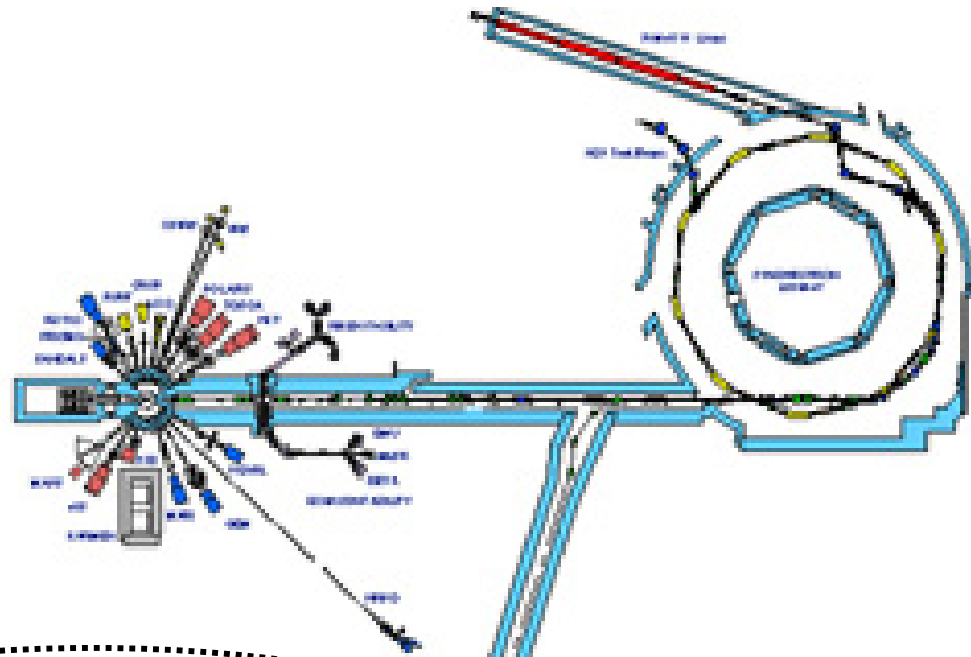




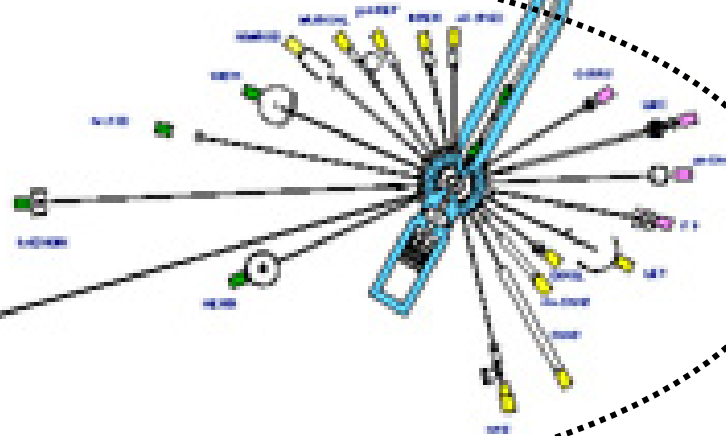
# Spallations-Neutronenquellen

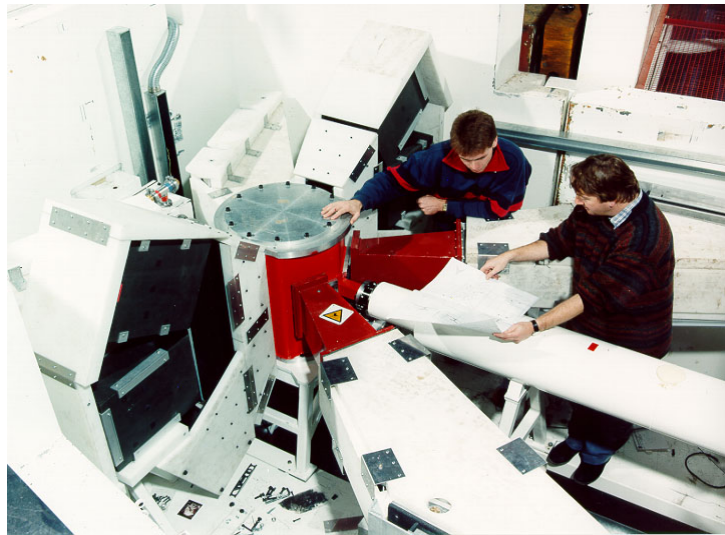
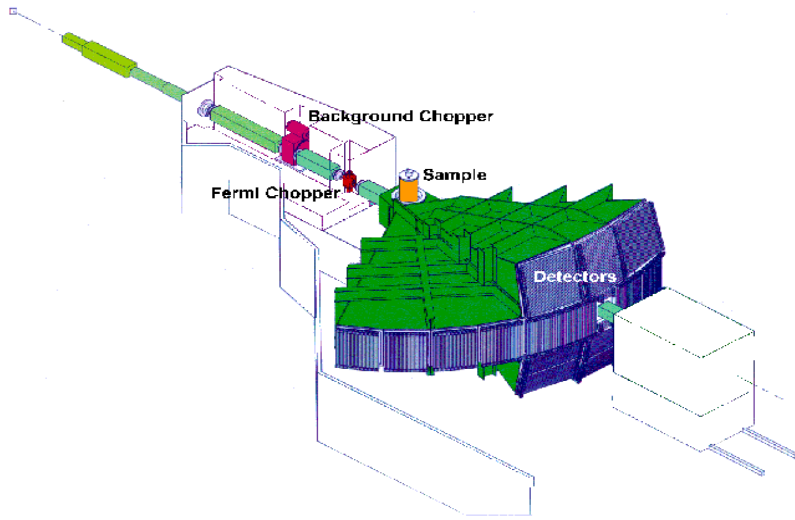
Oxford, UK

ISIS  
Rutherford Appleton Laboratory



In Bau







1.400.000\$

**In Bau**

SNS, Oakridge, USA



1.500.000€

**In Planung ?**

# Neutronen haben weltweit Zukunft ! ! !





# Reactor-Sources

- Budapest Neutron Centre, AEKI, Budapest, Hungary
- Berlin Neutron Scattering Center, Hahn-Meitner-Institut, Berlin
- Center for Fundamental and Applied Neutron Research (CFANR), Rez nr Prague, Czech Republic
- FRJ-2 Reactor, Forschungszentrum Jülich, Germany
- Frank Laboratory of Neutron Physics, Joint Institute of Nuclear Research, Dubna, Russia
- GKSS Institute for Materials Research, Hamburg, Germany
- Institut Laue Langevin, Grenoble, France
- Interfacultair Reactor Instituut, Delft University of Technology, NL
- JEEP-II Reactor, IFE, Kjeller, Norway
- Laboratoire Léon Brillouin, Saclay, France
- Ljubljana TRIGA MARK II Research Reactor, J. Stefan Institute, Slovenia
- St. Petersburg Nuclear Physics Institute, Gatchina, Russia
- Studsvik Neutron Research Laboratory (NFL), Studsvik, Sweden
- Centro Atomico Bariloche, Rio Negro, Argentina
- Chalk River Neutron Program for Material Research, Chalk River, Ontario, Canada
- High Flux Isotope Reactor (HFIR), Oak Ridge National Laboratory, Tennessee, USA
- Los Alamos Neutron Science Center (LANSCE), New Mexico, USA
- McMaster Nuclear Reactor, Hamilton, Ontario, Canada
- MIT Nuclear Reactor Laboratory, Massachusetts, USA
- NIST Center for Neutron Research, Gaithersburg, Maryland, USA
- Peruvian Institute of Nuclear Energy (IPEN), Lima, Peru
- University of Missouri Research Reactor, Columbia, Missouri, USA
- University of Illinois Triga Reactor, Urbana-Champaign, Illinois, USA
- Institute, Australian Nuclear Science and Technology Organisation, Lucas Heights, Australia
- High-flux Advanced Neutron Application Reactor (HANARO), Korea
- Japan Atomic Energy Research Institute (JAERI), Tokai, Japan
- KENS Neutron Scattering Facility, KEK, Tsukuba, Japan
- Kyoto University Research Reactor Institute (KURRI), Kyoto, Japan
- Malaysian Institute for Nuclear Technology Research (MINT), Malaysia
- Australian Replacement Research Reactor, Lucas Heights, Australia
- Canadian Neutron Facility, Chalk River, Ontario, Canada
- FRM-II Research Reactor, Garching, Germany

# Spallation Sources

- ISIS Pulsed Neutron Facility, Rutherford-Appleton Laboratory, Oxfordshire, UK
- Swiss Spallation Neutron Source (SINQ), Villigen SwitzerlandBragg
- Intense Pulsed Neutron Source (IPNS), Argonne National Laboratory, Illinois, USA
- Spallation Neutron Source, Oak Ridge National Laboratory, Tennessee, USA
- Japanese Spallation Source
- Austron Spallation Neutron Source, Vienna, Austria
- European Spallation Source (ESS)
- Japan Proton Accelerator Research Complex (J-PARC), Tokai, Japan

# Nachweis von Neutronen

**n nicht ionisierend – daher immer indirekter Nachweis**  
(keine Energieanalyse bei Nachweis)

**schnelle Neutronen (MeV):** Stöße mit p-haltigen Substanzen

**thermische Neutronen (meV):**

1. Aktivierung durch n – Einfang (z.B. n- $\gamma$  Reaktionen in Au,In)
2. Kernreaktionen und anschließender Nachweis der (geladenen) Reaktionsprodukte – üblich in der Neutronenstreuung: Geiger Müller Zählrohr
  - a)  $\text{BF}_3$ :  $\text{B}^{10}(\text{n}, \alpha) \text{Li}^7 + 2.79\text{MeV}$
  - b)  $\text{He}^3$ :  $\text{He}^3(\text{n}, \text{p}) \text{H}^3 + 0.765\text{MeV}$Nachweiswahrscheinlichkeiten ca 95% (hängt von  $\lambda$  ab,  $\lambda^2 \propto 1/v$ )
3. Szintillationszähler:  $\text{n} \rightarrow \text{B} \rightarrow \alpha \rightarrow \text{ZnS}$  (Photomultiplier)
4. Imageplates : Gd-Schicht  $\rightarrow$  n- $\gamma$  Reaktionen  $\rightarrow$  Detektion wie in x-ray

# Literatur

- S. W. Lovesey      Theory of Neutron Scattering from      Oxford (1981)  
Condensed Matter
- G. E. Bacon      Neutron Physics      Wykeham (1969)
- G. E. Bacon      Neutron Diffraction      Oxford (1979)**
- Shirane, Sahpiro and Tranquada Neutron Scattering with a triple axis spectrometer
- Izyumov, Ozerov      Magnetic Neutron Diffraction      Plenum (1970)
- Marshall and Lovesey      Theory of thermal neutron scattering**
- Squires      Thermal Neutron scattering**



# Neutrons – Photons

## Neutrons:

Particle beam (neutral)

$$E = \frac{h^2}{2m_N \lambda^2} = 81.1 \text{ meV} / \lambda^2$$

Low brilliance (particles/cm<sup>2</sup>/sr/meV)

Interactions with the nuclei and the magnetic moment of unpaired electrons

Scattered by all elements, also the light ones like the hydrogen isotopes

Deep penetration depth (bulk studies of samples)

Less intense beam measuring larger samples

### *Applications:*

Magnetic structures & excitations, critical scattering

## Photons:

Light beam

$$E = hf = hc/\lambda = 12398 \text{ eV} / \lambda$$

High brilliance

Interactions with the electrons surrounding the nuclei

Mainly scattered by heavy elements

Small penetration depth (surface studies of samples)

Very intense beam measuring small or ultra-dilute samples

### *Applications:*

Surface studies, element and shell sensitive resonant magnetic scattering, magnetic dichroism, magnetic Materials with high neutron absorption